A FEASIBILITY STUDY OF
RENOVATION OF SEWAGE LAGOON
EFFLUENT BY RAPID INFILTRATION

VILLAGE OF MARKDALE

September, 1980



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MINISTRY OF THE ENVIRONMENT

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THE ONTARIO MINISTRY OF THE ENVIRONMENT

A FEASIBILITY STUDY OF
RENOVATION OF SEWAGE LAGOON
EFFLUENT BY RAPID INFILTRATION
VILLAGE OF MARKDALE

by
B. Novakovic

Southwestern Region
Technical Support Section
Water Resources Assessment Unit
London
September, 1980

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MINISTRY OF THE ENVIRONMENT

We know accurately only when we know little, with knowledge doubt increases.

Johann Wolfgang von Goethe (1749-1832)

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SUMMARY

The area located immediately south, southwest and west of the Markdale sewage lagoon consists of gravel and sand with some silty and clayey sections. These granular deposits are of glaciofluvial origin. A portion of this land about 60 to 70 m wide situated along the Rocky Saugeen River is wet with small sections of cattail environment, consisting of peat deposits. This wet area along the Rocky Saugeen River is considered a discharge zone of the shallow water table and bedrock aquifer systems. The coefficient of permeability of sand and gravel deposits, 15 m in thickness is about 0.4 cm/s.

In general, the width of the area between the lagoon and the Rocky Saugeen River is about 300 m and it offers an excellent opportunity to achieve tertiary treatment of sewage lagoon effluent by infiltrating the effluent into the porous sand and gravel deposits.

A rapid infiltration study was conducted to determine the feasibility of sewage lagoon effluent treatment by means of groundwater recharge. If feasible, it would eliminate direct discharge of the effluent into the Rocky Saugeen River and thus maintain its high quality.

The study lasted from October 19 to November 26, 1979 during which time approximately $8,000~\text{m}^3$ of sewage lagoon effluent was infiltrated into the granular deposits. In actuality only 27 days were inundation periods during which the average recharge rate was 3.46~L/s. A conservative calculation indicates an average hydraulic loading of $24.91~\text{m}^3/\text{d/m}^2$.

The field installation consisted of a recharge basin with a controlled infiltration area (12 m²). Numerous observation wells and test pits were located at strategic locations in close vicinity to the infiltration basin. They were used to measure water level response to infiltration and to obtain water samples for qualitative analyses.

A slight reduction of infiltration rate appears to be due to gradual clogging of the bottom of the infiltration basin caused by deposition of suspended solids present in the sewage lagoon effluent. This clogging, generated by physical factors, affected only very shallow soils (5 cm) at the bottom of the basin.

The results of qualitative analyses indicated that the dissolved reactive phosphorous was reduced to background levels a short distance from the infiltration basin. The components of the nitrogen cycle were converted into nitrate and a considerable amount of nitrogen was reduced at a relatively short distance from the infiltration basin.

Other chemical constituents present in the sewage lagoon effluent were attenuated by the subsurface environment to various degrees depending partially on the distances from the infiltration basin.

Chemical constituents of the sewage lagoon effluent which may reach the groundwater discharge area will undergo further attenuation by biological and microbiological processes while travelling overland through wet environments before entering the Rocky Saugeen River.

The quality of this small portion of the sand and gravel aquifer which received waste water from the lagoon during the rapid infiltration study can be allowed to deteriorate since it has no prospective use in the future.

In summary, the results of this study are quite encouraging. They provide sufficient background information and preliminary criteria for the design of an operational system with infiltration basins.

INTRODUCTION

This report summarizes the results of a study of sewage lagoon effluent treatment by utilizing mainly hydrogeological, biological and hydrological environments. Two basic aspects are briefly addressed in this study; (a) hydraulics of the receiving aquifer and (b) renovation of the effluent while moving through granular materials.

Other aspects of a recharge system such as design criteria, infiltration system management, winter operating constraints, etc. are also discussed.

Objectives of the Study

The main objective of this study was to examine the possibility of utilizing hydrogeological environments for treatment of sewage lagoon effluent by means of a recharge basin.

Furthermore, the results of the study (if favourable) would be useful in establishing preliminary design criteria for a full scale operational system to treat sewage lagoon effluent by rapid infiltration.

Background

The Village of Markdale is presently serviced by a facultative one-cell lagoon system 5.67 ha in size (Figure 1 and Photo 1). The lagoon effluent is discharged in the spring and fall to the Rocky Saugeen River located about 170

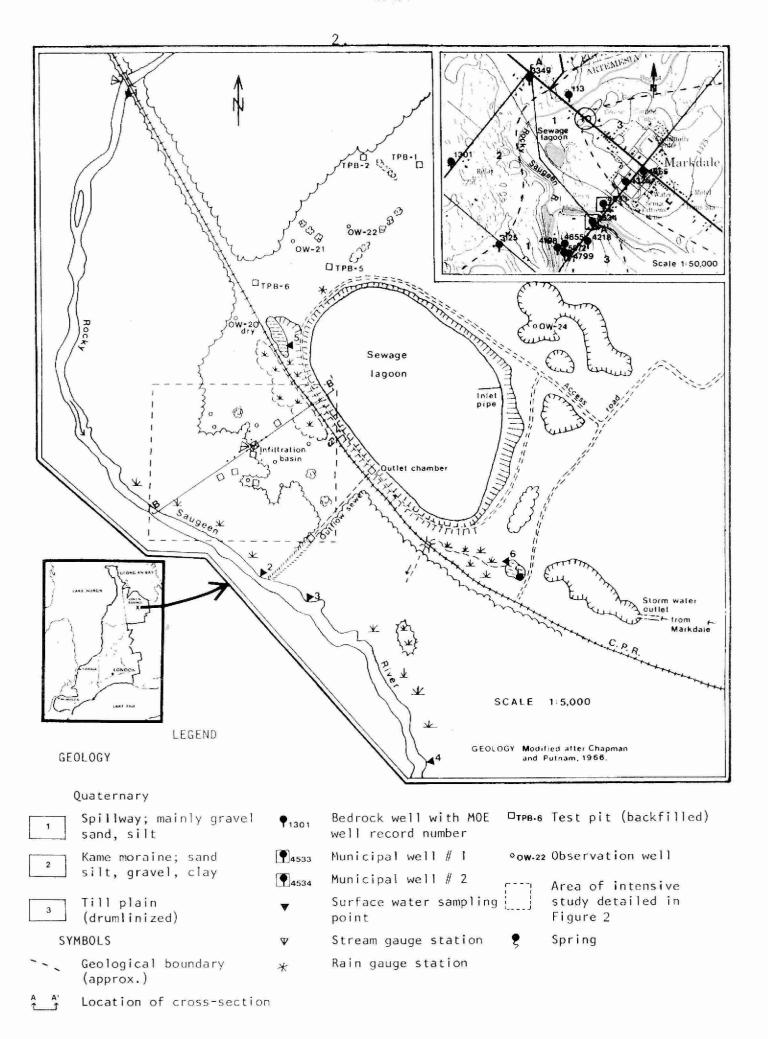


FIGURE 1. Location and surficial geology map.

m from the lagoon (Photo 2). From the control manhole, the effluent is directed by a 131 m long 20.32 cm diameter buried pipe and subsequently by an open ditch to the River (Figure 1).

A water quality survey of the Rocky Saugeen River carried out by staff of the Ministry of the Environment during the fall discharge of the lagoon effluent (MOE, 1979) showed a measurable effect on the river water quality to a considerable distance downstream from the sewage outfall.

The Rocky Saugeen River is a high quality stream which supports salmonid species. Therefore, there are concerns that expansion of the present mode of lagoon operation may have a significant impact on the stream water quality. In view of this, coupled with the fact that the present sewage treatment facilities are at capacity, this Ministry requested that the Village of Markdale look into other options (exfiltration) of improving sewage treatment facilities.

Subsequently, the Village engaged the consulting firms, Henderson, Paddon and Associates Limited and Morrison Beatty Limited to carry out the study.

This initial study by the Village's consultant was carried out in the summer and early fall of 1979. When funds for the feasibility studies were exhausted, the Ministry of the Environment, Southwestern Region, decided to apply staff resources to continue the work in order to take advantage of the experimental system that had been established.



Photo 1. A portion of the Village of Markdale sewage lagoon looking east. Village of Markdale is in right background. Pumping equipment is sitting on the control manhole at the extreme lower right margin.



Photo 2. Looking downstream, Rocky Saugeen River immediately upstream of the lagoon effluent discharge (arrow).

Location

The study area is located about 900 m west of the Village of Markdale, in Lots 94, 95 and 96, Concession I West of Toronto - Sydenham Road, Glenelg Township, Grey County. More precisely, the land of intensive study is situated between the Markdale oxidation pond and the Rocky Saugeen River (Figures 1 and 2).

Figures 1 and 2 also give the locations of monitoring stations and field instrumentations used in this study.

Access to the study area is provided by the access road to the sewage lagoon which is also used to transport sand and gravel extracted from the pit located immediately northeast of the lagoon (Figure 1). The study area is about 900 m from the provincial highway 10.

The study area is traversed by the Canadian Pacific Railway (Toronto - Owen Sound line).

Topography

The area of intensive study is situated on relatively flat ground of the Rocky Saugeen River valley botom, which in this area is about 300 m wide. There is about 6 m difference in elevation between the river and the "highest ground" which is actually a very small ridge running parallel to the river between 150 and 180 m away. The infiltration basin was located near the extreme southern flank of the ridge's crest (Figure 1).

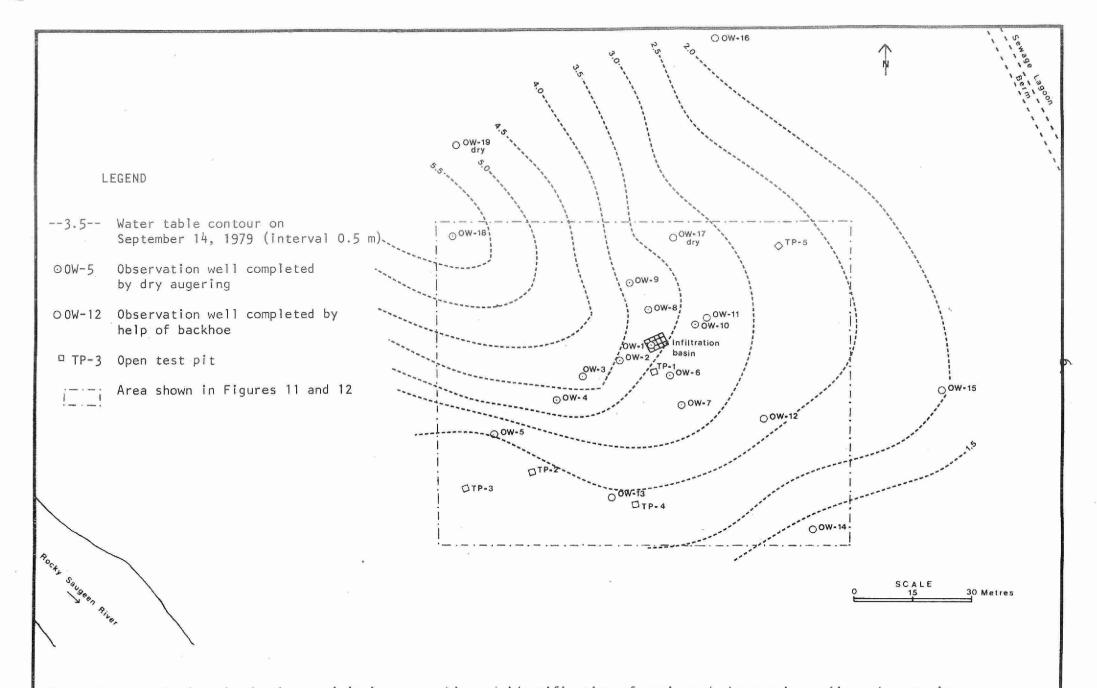


Figure 2. Map showing the depth to original water table and identification of monitored observation wells and test pits.

In general, the area between the river and this minute ridge is overgrown by cedar trees, whereas in other sections occasional cedar trees and willows can be found particularly in the wet portion (Photo 3).

Drainage

The Rocky Saugeen River is deeply incised into the local glacial deposits and any associated drainage system and runoff are directed into the River which is a focus for surface and groundwater discharges.

There is very little surface water runoff occuring after intensive rainfall and snowmelt, because most rainfall infiltrates quickly into the permeable soil. A minor stream rises southeast of the lagoon fed by the natural spring-pond system and periodically by the discharges of storm water released into the former gravel pit some 130 m away (Figure 1). After passing the CPR tracks, this flow disperses into a mucky, wet environment.

A second permanent pond is located immediately northwest of the sewage lagoon (Figure 1). This pond is an integral part of a low lying wet and occasionally water covered strip of land about 25 m wide running parallel to the CPR tracks, but on the opposite side of the tracks from the lagoon.

Another wet and low lying area, overgrown by cedar trees about 60 to 70 m wide, is situated along the Rocky Saugeen River. In this section of land, black peat and small cattail environments are commonly found (Photo 4).

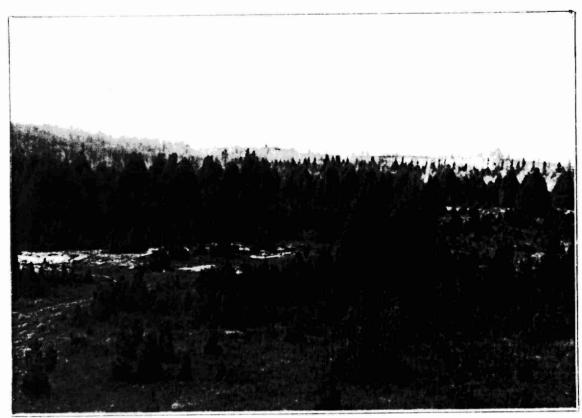


Photo 3. A portion of the study area looking west, with the recharge basin at the extreme left - centre. The boundary between coniferous and decidious trees at left marks the position of the Rocky Saugeen River.

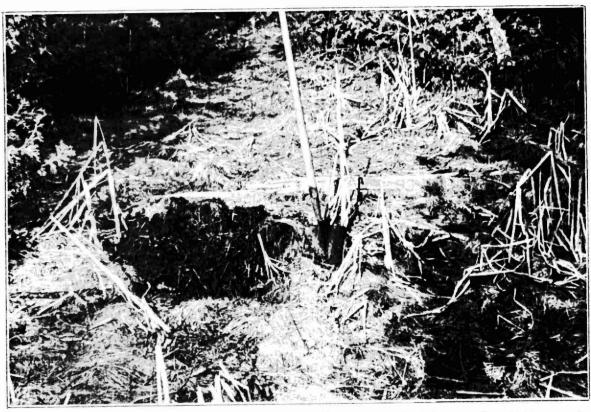


Photo 4. A section about 60 m wide consisting of wet peat deposits with minute cattail environments are present along the Rocky Saugeen River.

Previous Investigations

In connection with the proposed sewage system construction, William Trow Associates Limited (1966) investigated geotechnical conditions in the Village of Markdale and at the property now occupied by the sewage lagoon.

Hydrology Consultants Limited (1973) carried out test pumping and an evaluation of hydraulics of two municipal wells which are now water sources to the Village of Markdale.

Preliminary reconaissance work, including excavation of 16 test pits and construction of all field installations for groundwater monitoring except for TP-1 to TP-5 inclusive, staff gauges and rain gauge was carried out by Morrison Beatty Limited (1979). This work also included a 24-hour recharge test taking water from the Rocky Saugeen River as a recharge source and a 6-hour infiltration test using sewage lagoon effluent pumped directly from the lagoon. The results of their efforts were summarized in a brief report.

Field Work

The field investigation lasted from October 18, 1979 to November 26, 1979. The field work among other things included:

 Installation of two staff gauges, rain gauge and construction of five test pits left open.

- Pumping of the sewage lagoon effluent by a gasoline driven pump into the infiltration (recharge) basin.
- 3. Very frequent sewage flow measurements and maintaining and recording of a constant height of wastewater in the infiltration basin.
- Frequent water level (at least twice a day) measurements in observation wells and test pits.
- 5. Daily checking of rain gauge station and staff gauge in the Rocky Saugeen River.
- Collection of numerous water and wastewater samples for qualitative analyses.
- Collection of soil samples for mechanical analyses.
- 8. Performing a dye test in order to establish underground retention times of the renovated water.

Field Installations

The location of the field installations is given in Figures 1 and 2. Stratigraphy, depth and some details concerning the construction of observation wells and test pits is given in Appendix A.

A recharge basin located about 160 m distance from the pump intake was specifically constructed for the purpose of this study. The side and marginal sections of the basin bottom were lined with polyethylene so that the dimensions of the infiltration area were 3 X 4 m (12 m^2). The slope ratio of the side was 1.5:1 and depth was 0.83 m.

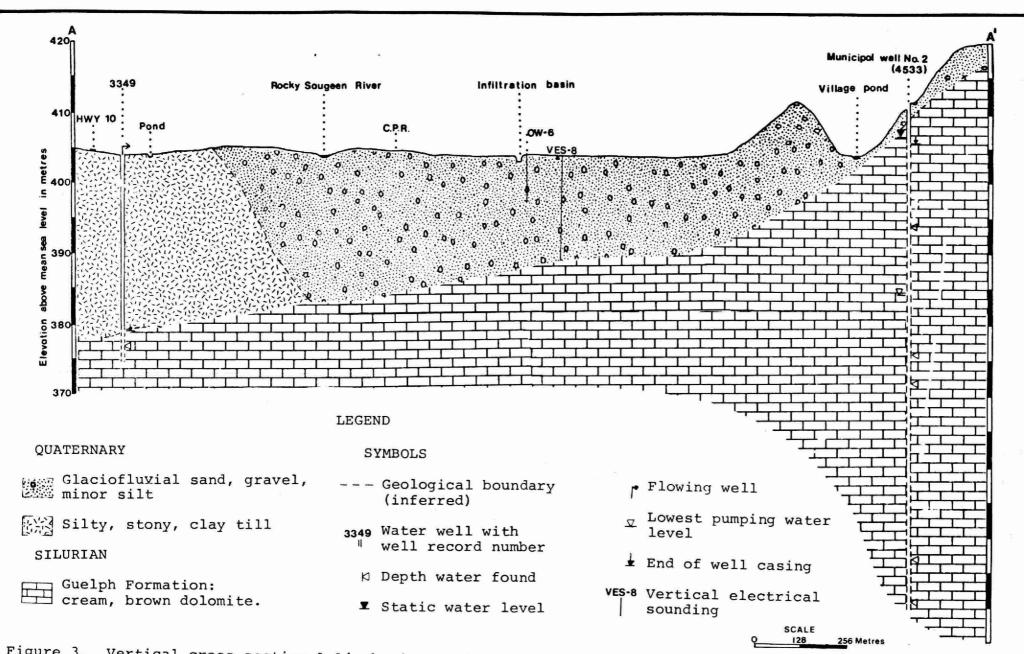


Figure 3. Vertical cross-section A-A' showing geology and hydrogeology in the general area of investigation.

About 20 to 30 cm of the original granular material at the bottom of the recharge basin was replaced with borrowed, well-sorted sand and gravel.

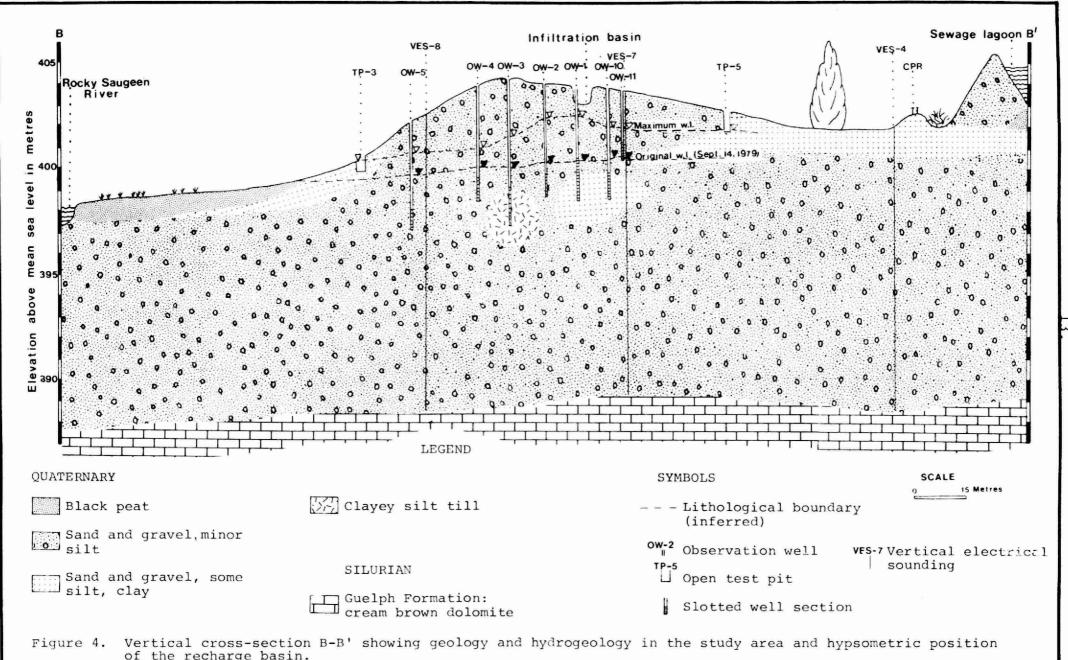
The hypsometric position of the infiltration basin and its relationship with the nearby piezometers is shown in Figures 3 and 4.

There were 11 observation wells (OW-1 to OW-10 inclusive and OW-18) constructed using dry augering techniques, all located in the vicinity (positioned in a cross-like fashion) of the recharge basin. Other observation wells installed with the help of backhoe and five open test pits (TP-1 to TP-5 constructed on October 29, 1979) were also located in close proximity to the basin (Figure 2).

One observation well (OW-1) along with a staff gauge were constructed right in the infiltration basin.

All piezometers were made of PVC pipe 5.71 cm in diameter with 1 to 2 m slotted sections at the bottom. The length of a single slot was about 15 mm, width of slots about 1 mm and slot spacing about 75 mm for OW-1 to OW-10 and OW-18. For the rest of the observation wells the slot width and spacing were irregular. A protective cup was placed at the bottom of the pipe as well as a bentonite clay plug near the surface for OW-1 to OW-10 and OW-18.

The majority of observation wells (OW-1 to OW-18 except OW-17 which was dry) and test pits (TP-1 to TP-5) were frequently used to measure the hydraulic response of the water table aquifer to recharge, and they also provided access to the water table to obtain water samples for qualitative analyses.



of the recharge basin.

A flowmeter was placed a few metres away from the effluent outfall into the recharge basin. A temporary rainfall gauge was installed at the site as well as two staff gauge stations; one was installed in the recharge basin and the second in the Rocky Saugeen River, under the CPR bridge (Figure 1).

GEOLOGY AND HYDROGEOLOGY

Bedrock Geology

Nowhere in the study area does the bedrock outcrop. In the southwestern section of the Village of Markdale at municipal well No. 2 the bedrock is found near the ground surface overlain by about 1.2 m of granular deposits. The bedrock within the study area, and in the wider area as well, is of Middle and Lower Silurian age. It has been ascribed to the Guelph Formation which consists of dolomite. The dolomite is typically massive bedded, white, blue or brown in colour, fine to medium crystalline and highly porous.

The Guelph Formation may be seen in the Rocky Saugeen River bank at the intersection of Grey County Road 12 and associated side road some 1500 m from the study area.

Figures 3 and 4 show the hypsometric position of the bedrock within the study area and in the broader area outside of this investigation. Information from the geophysical survey indicates that the bedrock in the study area is overlain by 13 to 15 m of glacial deposits.

Surficial Deposits

The distribution of unconsolidated deposits is given in Figure 1 (inset map). Numerous excavated test pits and boreholes augered in the study area confirm that the overburden basically consists of sand and gravel, with minor silt and clay (Appendices A and B; Photos 5 and 6). In some

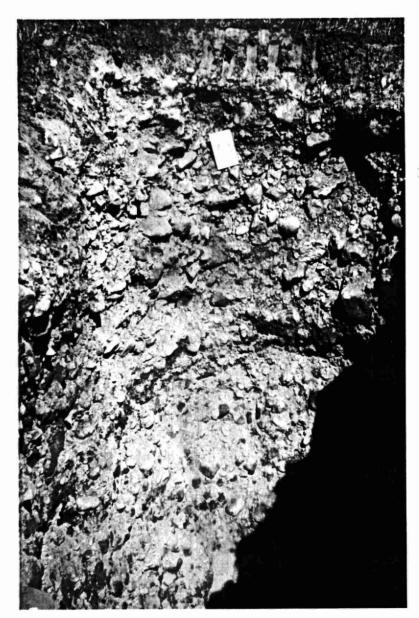


Photo 5. Silty sand and gravel exposed in the test pit excavated at OW-1.



Photo 6. Sand and gravel with very little silt at OW-20.

sections, along with sand and gravel there are cobbles with minor boulders. Granular deposits of reduced permeability consisting of silty sand and gravel with minor clay fractions are present in the low area along the CPR tracks (opposite from the lagoon) and on both sides of the sewer outflow line (Figure 1). This material is most likely present in the lowland area along the Rocky Saugeen River where it is overlain by peat deposits (Figure 4, Photo 4).

Three boreholes (OW-3, OW-6 and OW-7) intercepted poorly permeable material (clayey silt till with some pebbles) at a depth of 4.9 and 5.8 m respectively. This initial information suggested that the sand and gravel deposits may be underlain by poorly permeable material separating it from the bedrock. However, a subsequent geophysical resistivity survey (electrical sounding) carried out in May, 1980 indicated that the surficial deposits mainly consist of sand and gravel (Appendix C). Directly overlying the bedrock formation, these deposits are about 15 m thick in the study area. Because it is not possible to distinguish between sandy silt till and sand and gravel on the basis of resistivity survey, it is quite conceivable that a silty till underlies the sand and gravel, thus separating it from the bedrock. It is also possible that sections of poorly permeable material less than 3 m in thickness which cannot be delineated using the electrical sounding technique may be present within the deeper portion of glacial deposits. Information from three boreholes (OW-3, OW-6, and OW-7, Appendix A) support this hypothesis.

The peat deposit is present in the low lying area along the Rocky Saugeen River in the zone about 60 to 70 m wide (Photo 4). Its thickness is unknown, however, a test hole dug in the present lagoon area reported the presence of 1.8 m of peat.

Hydrogeology and Groundwater Movement

Bedrock Aquifer System

The porous bedrock dolomite constitutes a high capacity aquifer system which is the source for two municipal wells and for many domestic wells in the broad area of investigation.

Permeability of the bedrock formation is due primarily to chemical dissolution of the dolomite along ancient weathered zones, fractures and bedding planes. Because this dissolution can produce a large variety of opening sizes and patterns, permeabilities in the bedrock vary widely. In general, the bedrock appears to be most permeable within 6 m of its upper surface.

In the broader area of investigation (inset map in Figure 1) the potentiometric head in the bedrock aquifer, as obtained from water well records, is higher than the Rocky Saugeen River bed. This strongly suggests that the ground-water movement in the bedrock is towards the river from both sides with the river being the discharge zone of the bedrock aquifer system.

Sand and Gravel Aquifer

All boreholes penetrated into the saturated zone in the surficial deposits. Only three test pits were not deep enough to reach water table (OW-17, OW-19 and OW-20).

Depths to water table in the area of intensive study vary considerably depending on the topographic configuration. In the small ridge area northwest of the recharge basin the depth to water table is over 5.5 m (Figure 2) whereas in the low lying area along the river, it is at the ground surface.

Groundwater flow direction in the shallow water table aquifer is strongly influenced by topography (Figure 5). The flow is to the southwest with the Rocky Saugeen River being the focal point - the discharge zone of both the overburden and bedrock aquifer systems.

Permeability of Surficial Deposits

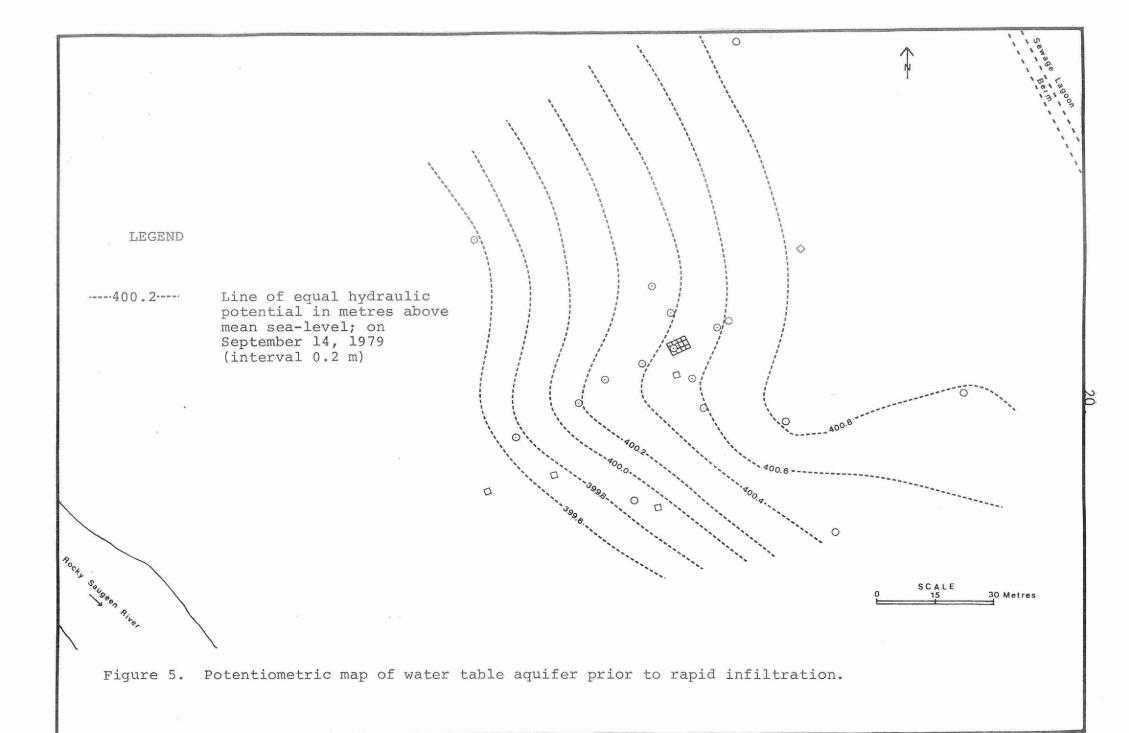
Two basic parameters describe porous media: porosity and permeability. Porosity is "the ratio of pore volume to the total volume of a given sample of material", whereas permeability "is a measure of the ease with which fluids pass through a porous material" (Davis, 1969).

The porosity of unconsolidated deposits depends on their packing arrangements, size distribution and shape. The porosity of sand and gravel is 20 to 35 percent (Todd, 1959).

The intrinsic or specific permeability of a porous medium to a fluid is given by

$$k = \frac{Q\mu}{A G G} \left(\frac{dh}{d1} \right)^{-1}$$
 (1)

Where Q is the fluid volume discharged per unit time through a porous medium of cross-sectional area A, μ is the dynamic



viscosity of the fluid, % is the fluid density, G is the accelaration due to gravity, and dh/dl is the hydraulic gradient in the direction of flow (Davis, 1969). The intrinsic permeability, k, has dimensions of L^2 and is a function of the geometric properties of the medium. Since hydrogeologists are mainly involved with water as the fluid, a more commonly used hydrogeologic variable is the hydraulic conductivity, K. If the terms in Equation (1) are rearranged the hydraulic conductivity may be defined as:

$$K = k g G/u$$
 (2)

which is often loosely referred to by hydrogeologists as the "permeability", i.e., the coefficient of permeability, and which has dimensions L/t .

Most analyses of groundwater flow in some way involve a principle known as <u>Darcy's law</u>, which has already been stated in one form in the definition of intrinsic permeability in Equation (1). It can be rewritten as:

$$q = -K \left(\frac{dh}{d1}\right) \tag{3}$$

where q=Q/A in Equation (1) and is known as the Darcy or filter velocity, or the specific discharge, and K is the hydraulic conductivity in the 1 direction. The term q is not the average pore water velocity, but a seepage flux and has the units L^3/L^2t .

In this study a common method, the Hazen formula, was used to estimate the intergranular hydraulic conductivity of the sand and gravel deposits. This method is

Table 1. Selected values of the coefficient of permeability of surficial deposits determined by the Hazen formula

Location*	Depth	Coefficient of		
source	(m)	permeability		
		K(cm/s)		
TP-1	0.61	0.56		
TP-1	1.22	0.11		
TP-1	1.83	0.17		
TP-2	0.61	0.42		
		2		
TP-3	0.46	1.22×10^{-3}		
TP-4	0.4	0.56		
TP-4	0.73	0.40		
TP-4	1.22	0.43		

^{*}Location is shown in Figure 2

based on the empirical equation developed by Hazen (1893). Although the relation has been refined in recent years, the Hazen formula

$$K (cm/s) = C(d_{10})^2$$
 (4)

essentially relates the hydraulic conductivity, K, to the effective grain-size diameter, d_{10} , by means of the coefficient C (which varies from about 50 cm $^{-1}$ s $^{-1}$ to 150 cm $^{-1}$ s $^{-1}$). The hydraulic conductivity is then proportional to the effective grain size of the deposit and is also a function of the degree of uniformity of the particle size. Since, as a rule, the Hazen formula gives higher values of permeability for clean gravel and sand deposits, the empirical coefficient C = 100 was used in this calculation.

The grain size distribution of the particles comprising surficial deposits in the study area is given in Appendix B. These data were used to calculate the coefficient of permeability applying the Hazen formula. The results are given in Table 1. It can be seen from Table 1 that values for hydraulic conductivity are relatively high.

Water Use in the Area

The water wells for which records are on file with this Ministry are indicated in Figure 1 (inset map). A summary of the water well records for these wells is given in Appendix D. Information from water well records indicates that the confined bedrock aquifer system is a source of water for these wells.

There are two municipal wells which supply water to the Village of Markdale and they are the nearest wells to the study area (about 1070 m distance). Information from water well records for these two wells indicate that water was encountered at several depths in the bedrock in both municipal wells (Figure 3). The pumping water levels in both wells appear to have reached dynamic hydraulic equilibrium condition with the lowest pumping water level at depths of about 27 m in both municipal wells. Water level recovery is rapid after the well production is stopped, indicating high capacities of both wells.

Since the municipal wells are located at relatively short distances from the rapid infiltration area, assurances should be made that water quality in these two wells and in the bedrock in general, remains unaffected. The likelihood that contamination of municipal wells would occur is unlikely because:

- The thickness of surficial deposits (saturated zone) overlaying the bedrock is substantial.
- There is a well defined hydraulic gradient in the shallow aquifer towards the river with prevailing lateral groundwater movement and the river is the ultimate groundwater discharge zone.
- 3. There is an upward gradient in the bedrock aquifer constituting the discharge zone of groundwater flow in the bedrock.

Nevertheless, several boreholes should be drilled in the study area and completed in the bedrock in order to check findings obtained using geophysical techniques. These bedrock wells may be used in the future to monitor water quality in the bedrock aquifer.

INFILTRATION STUDIES

Procedure

Sewage lagoon effluent was directly pumped into the infiltration basin by a gasoline-powered pump. The pumping equipment was placed on the lagoon berm, on the concrete top of the control manhole, whereas the pump intake was set into the sewage outfall chamber (Photo 7). The speed of the pump was throttled to keep constant head in the infiltration basin. Because of its tendency to fail, the pump needed close attendance; therefore, whenever the pump was running, its performance was frequently checked, generally between 6:00 and 23:00 hours.

Using a stop watch, an instant flow rate was obtained by reading the quantity of flow from the flowmeter (usually recording time for flow of 100 gallons). The stop watch and a calibrated container were also frequently used to check the reliability of the flowmeter which also gave a cumulative flow measurement.

The flow rate measurements, hydraulic head of the effluent in the recharge basin (staff gauge), and ground-water level in the observation well 1 (OW-1) were considered the bench-mark parameters of the study (Photo 8). They were, therefore, the subject of watchful attendance and measurements. For example, the only time period when the pump and basin area were not closely attended was between 23:00 and 5:00 hours the following morning.

Water levels were frequently measured in the observation wells and the open test pits. Numerous water samples from groundwater and surface water sources and of sewage lagoon effluent were collected for qualitative analyses.



Photo 7. Continuous pumping (intake in the control manhole)
was provided by a gasoline-driven pump. A full drum
provided about 12 hours of continuous run. Infiltration
basin is located in front of coniferous trees at
extreme left.

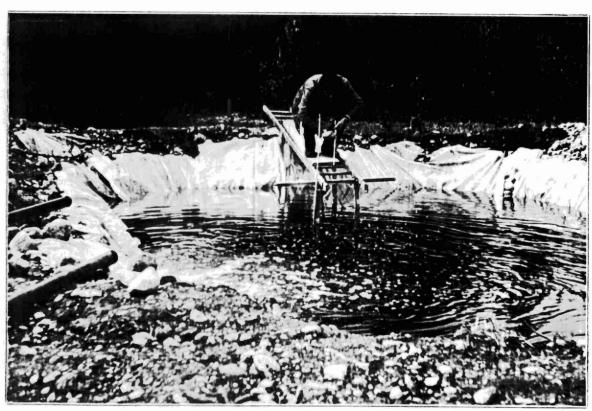


Photo 8. Recharge basin looking southwest. Inflow of 3.8 L/s. Water levelis being measured in OW-1.

Table 2. Data pertaining to the recharge basin conditions and infiltration rates. a

Basin con	dition	Time ;	period	Area of in ba	nfiltration asin	Average rate of	f infiltration
	ydraulic ead (m)			Contact	Area (m²)	L/s/m²	m³/d/m²
Lined on	0.61		19, 1979 to 22, 1979	Bottom	12	0.28	24.19
No lining	0.4		22, 1979 to	Bottom	12	0.24	20.74
		Nov.	26, 1979	Sides	4.4	0.26	22.46

This table should be read in conjunction with data shown in Figure 6.

Infiltration Rates

A continuous controlled infiltration study commenced on October 19, 1979 and continued with some dry up (resting) periods until November 26, 1979. Some of these stoppages were purposely planned, while a few of them were caused by failure of the pumping equipment.

In actuality, there were only 27 days of infiltration during which period about 8,020 m³ of wastewater were infiltrated. This value was obtained from the spot measurements using a stop watch and a calibrated container for the initial six days of infiltration (before the flow-meter was installed) and from the accumulative values as recorded by the flowmeter for the remainder of the study period. Although not very reliable, an evaluation of the flowmeter performance showed that its resulting flow data should be increased by at least five percent. This correction of flow has been included in the above calculation.

Instantaneous flow measurements using both stop watch and flowmeter on one hand, and a calibrated container and stop watch on the other are plotted in Figure 6. It is apparent that flow values reported by the later technique are considerably higher than the conservative values recorded by the flowmeter and used in the above and the following calculations. Thus, when the infiltration reported by the flowmeter is adjusted to the flow values obtained by stop watch and calibrated container, close to $9,000~\rm m^3$ (2 MIG) were infiltrated to the granular deposits.

The results of the infiltration rates under different conditions in the infiltration basin are summarized in Table 2. It is apparent that they are relatively high infiltration rates.

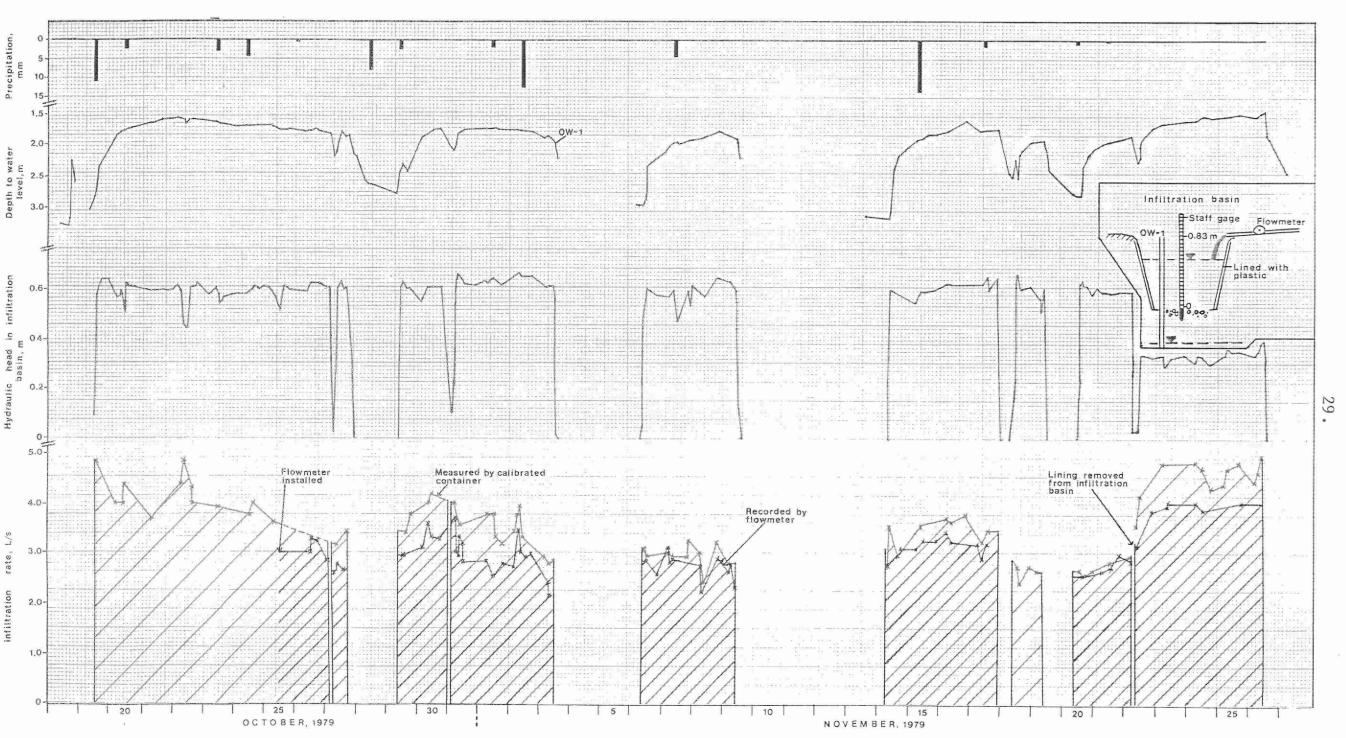


Figure 6. Infiltration rates, hydraulic head in basin, depth to water level in OW-1 and rainfall at site.

As indicated earlier, the sides of the infiltration basin were lined for the most part of this study. However, the lining was removed on November 22, 1979, four days before the termination of the recharge study. It should be noted therefore, that in calculating the infiltration rate for the lower portion in Table 2, it was assumed that the infiltration rates through the bottom of the basin remained constant at 2.88 L/s, as it was immediately before the lining was removed.

Heights of effluent in the recharge basin were fairly constant at 0.6 m, except for the last four days of the study. During this later period, the weakening of the pumping equipment, in concert with rapidly high infiltration rates, generated a relatively low hydraulic head in the basin (0.4 m).

The cumulative infiltration rates were plotted in Figure 7.

Hydraulic Response of the Shallow Water Table Aquifer to the Recharge

There were over 20 measuring points of the water table situated close to the infiltration basin. Although, there was no true piezometer nest with intakes placed at various depths, the obtained information from the available installations provides reliable data for an assessment of the hydraulic characteristics of the sand and gravel aquifer system.

The original static water level was at a depth of 3.5 m below ground level at the recharge basin, before the start of the infiltration program. The response of the

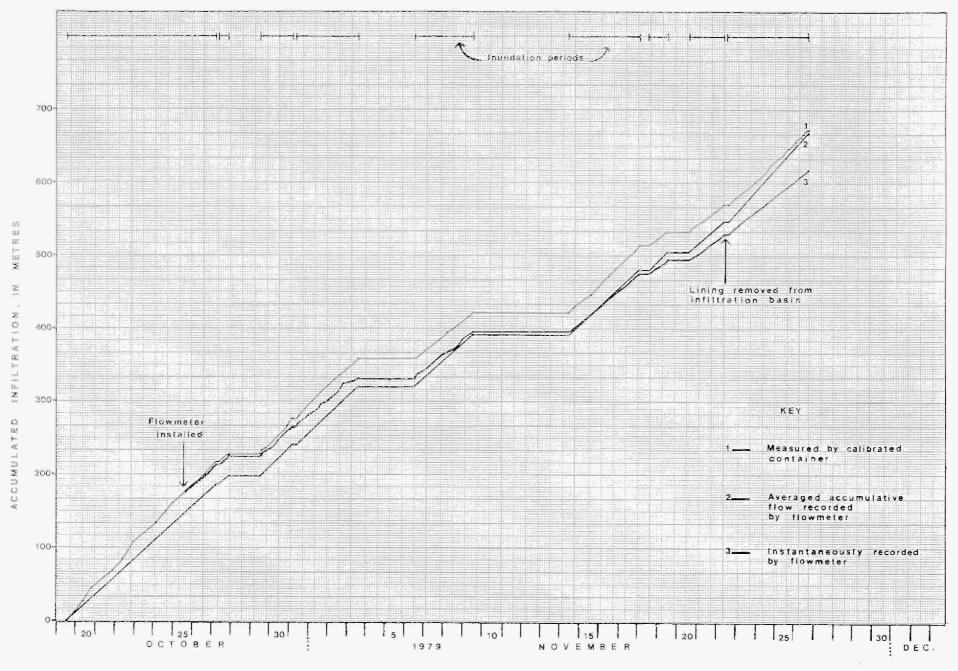


FIGURE 7- Cumulative infiltration of the lagoon effluent.

water table to infiltration was rapid, reaching an apparent hydraulic equilibrium underneath the recharge basin two days after infiltration commenced (Figure 6). In general, depths to water level below the basin remained relatively consistent at 1.7 m during infiltration periods. During the dry up periods the water level dropped rapidly to about 3 m and then the rate of recession slowed down considerably.

A similar situation prevailed in other observation wells and test pits located at various distances from the infiltration basin (Figure 8). It can be seen from this figure that the water table response was relatively rapid depending upon the distance of the observation point from the recharge basin. In general, a steady state condition in the affected wells was reached two to three days after infiltration began. It generally remained constant with periods of quick recessions during dry-up periods. One exception to this general pattern of water level behaviour occurred during the last four days of infiltration when the rate of recharge was the highest therefore causing the water level to reach its peak (Figure 8).

Sudden occasional drops in water levels in several observation wells, as shown in Figure 8, are caused by taking water samples from observation wells for qualitative analyses. Slow water level recovery was most likely caused by silting of the slotted section of the pipe which prevented water from moving quickly into the pipe.

The lateral distribution of the influence of infiltration on the water table is shown in Figures 9 and 10. The comparison of Figure 5 with Figure 9 shows how the well-defined groundwater flow (Figure 5) is modified by recharge (Figure 9), resulting in the formation of the groundwater hydraulic mound. The irregular shape of the hydraulic mound closely reflects the topographic gradient. There was no apparent indication of hydraulic channelling, or short circuiting.

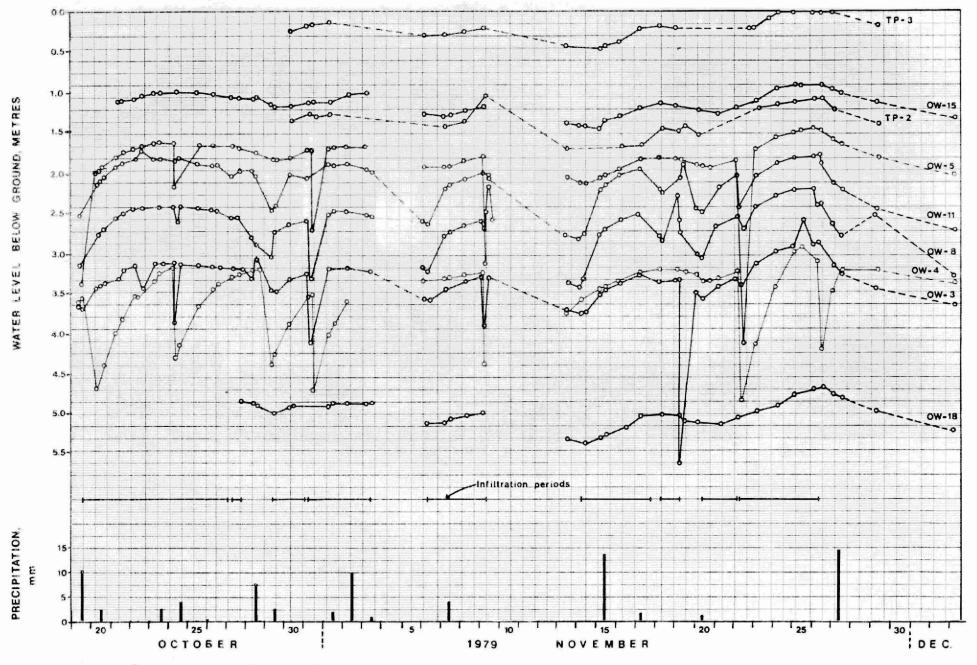


Figure 8. Response of water level to infiltration in observation wells and in test pits.

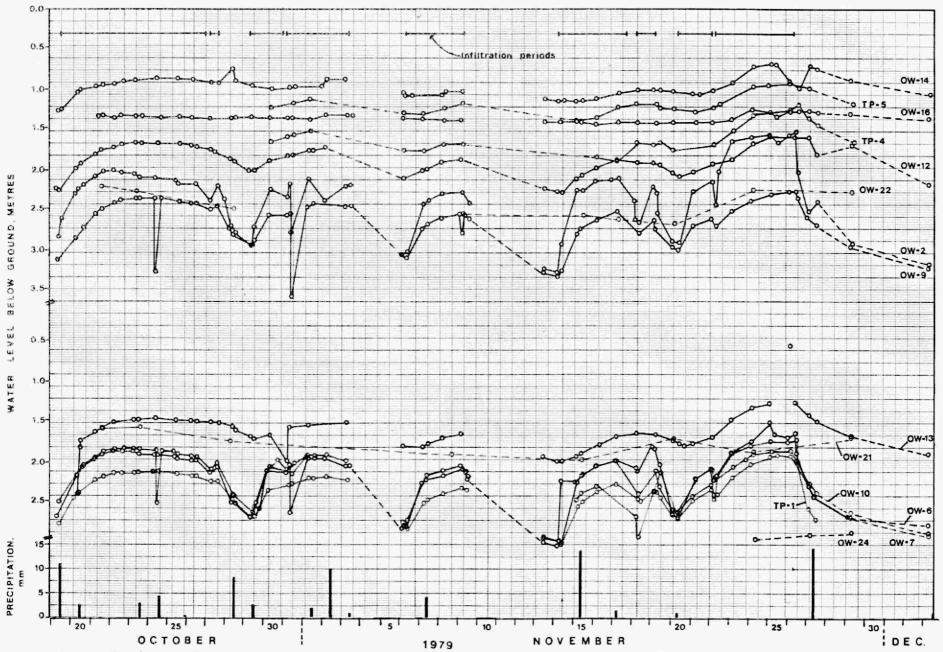


Figure 8 (Cont'd). Response of water level to infiltration in observation wells and in test pits.

An isopach of the groundwater mound is shown in Figure 10. Although the water level below the infiltration basin rose quite close to the bottom of the basin (0.6 m) it never actually reached it.

An apparent effect of groundwater mounding was a lateral shift of the highest margin of the groundwater discharge zone for about 30 m. This situation is illustrated in Figure 4.

Underground Detention Times

One of many important aspects of a rapid infiltration study such as this is to establish how long it will take for the effluent to reach the groundwater discharge area. This is because renovation of wastewater, which is influenced by several major factors, is directly related to the underground travelling time of infiltrated effluent. Detention times could be calculated in several ways such as by observing arrival of the pollution front, monitoring water level response, measuring the hydraulic gradient along the flow lines, dye test, etc.

The underground detention time is inversely proportional to the infiltration rate and to the permeability of the coarse granular material.

In this study, an attempt was made to use a direct method to calculate the permeability of granular materials. On November 20, 1979 after letting the basin dry up, 170 gm of fluorescein dye along with 213 litres of artificially made brine were placed at the bottom of the basin. The brine solution was made by dissolving 28 kg of uniodized salt using lagoon effluent as dissolvent. The concentration of the brine was as follows:

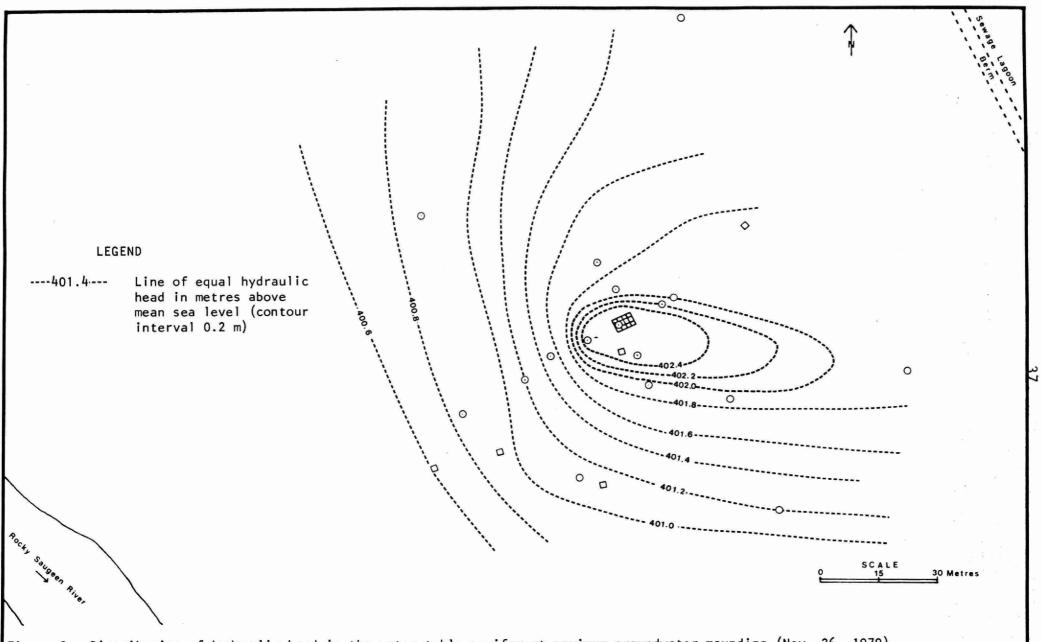


Figure 9. Distribution of hydraulic head in the water table aquifer at maximum groundwater mounding (Nov. 26, 1979).

Chemical parameter

Concentration

Conductance (umho	/cm³ at 25°C)	153,615
Sodium as Na (mg/	L)	48,179
Chloride as Cl (m	g/L)	73,307

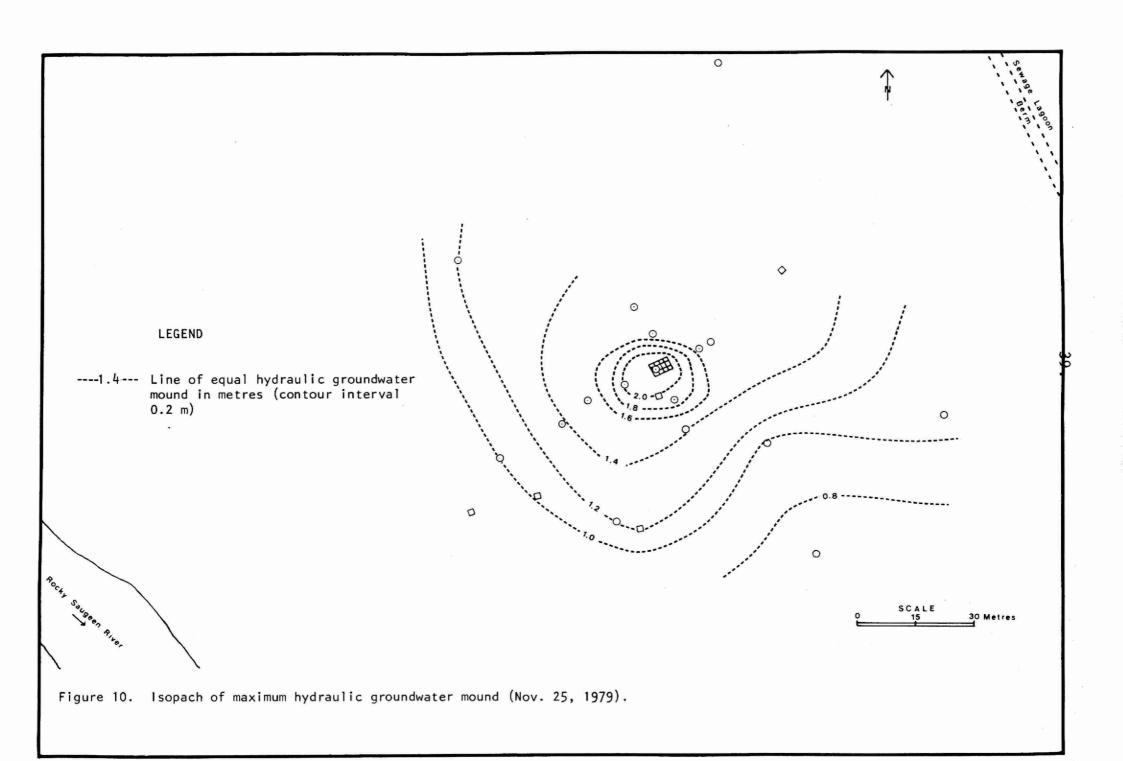
At the designated time the infiltration began, followed by frequent measurements of water levels, conductance and taking water samples from observation wells and test pits for fluorescein dye analyses.

Measurements of the specific conductance in observation wells was not successful in indicating the arrival of the contamination front (sodium chloride) nor was the fluorescein dye present in detectable concentrations in several monitored observation wells, and in TP-2, TP-3 or TP-4. It is reasoned that sodium chloride was quickly diluted in the rapidly infiltrating effluent, while the fluorescein dye was absorbed by silt and clayey particles. Nevertheless, closely monitored response of the water table in several observation wells and test pits gave valuable information on the hydraulic conductivity of the granular deposits.

The calculated values of the coefficient of permeability using the information from the mechanical analyses and the empirical formula presented by Hazen (1893) are given in Table 1.

The following equation was used to calculate the average linear pore-water velocity, $\mathbf{V_v}$,

$$V_{X} = \frac{K}{n g_{N}} \text{ grad h}$$
 (5)



where K is hydraulic conductivity, n is the porosity, G_N is an empirical constant and grad h is the gradient of hydraulic head. Experimental data on relatively uniform sands indicate that G_N is close to 1 for these materials. It was assumed that the sand and gravel deposits have an intergranular porosity of 0.30 and the empirical coefficient was unity. Grad h for the original water table was 0.017, whereas during the groundwater mounding effect it was 0.032. As data on Table 1 indicate, the hydraulic conductivity varies somewhat; however, an average of seven values for the hydraulic conductivity was used (0.4 cm/s). Using these values the obtained average groundwater velocities were 20 m/d for the conditions prior to the commencement of infiltration and about 38 m/d during the groundwater mounding effect due to artificial recharge.

Based on the obtained data, it is estimated that the underground detention time was in the range of one and a half to three days. Since the original upper limit of groundwater discharge is located about 60 to 70 m from the river, there will be additional time for the effluent to travel overland through wet land before entering the river system. While travelling overland through favourable attenuating environments, it will be subject to a significant additional attenuation of certain parameters involving numerous processes present along the travelling paths.

Basin Clogging

Maintaining a high infiltration rate for long periods of time is essential for successful sewage lagoon effluent treatment by rapid infiltration. Understanding soil clogging processes that causes infiltration rate reduction is necessary for the design and long term operation

of the recharge system. There are three basic factors that can cause soil clogging; chemical, biological and physical.

Chemical clogging is largely caused by chemical interaction between chemical constituents dissolved in the effluent and present in the soil resulting in decreased pore diameter and consequently, lower permeability. Research shows that chemical clogging seldom occurs unless the sodium content of the effluent is high.

Biological clogging occurs when bacterial growth or its by-products reduce the pore diameter. Biological clogging is frequently associated with anaerobic conditions and usually occurs at the soil surface, but it can also occur at different depths (McGauhey et al., 1967).

Physical clogging is the result of suspended solids in the effluent blocking the soil pores. Some particles may penetrate to a greater depth if the pore size is larger than the diameter of the suspended solids. A restricting layer is therefore built up by these suspended solids. The hydraulic properties of the clogged layer are best expressed in terms of the hydraulic resistance or impedance, which is defined as the thickness of the layer divided by its hydraulic conductivity. According to Bouwer (1972) the impedance can be calculated as the head loss through the layer divided by the infiltration rate.

The gradual reduction in infiltration rates, as indicated in Figure 6 was caused primarily by physical clogging of the bottom of the infiltration basin (Photos 9 and 10). The examination of Figure 6 indicates that during the first inundation period, an initial infiltration rate of about 4.55 L/s was reduced to 3.4 L/s just before the pump was shut off eight and a half days later. A dry up period which followed, along with 8 mm of rainfall helped to somewhat restore the recharge rate to 3.78 L/s.

The infiltration rate at the end of the inundation period on November 9, 1979 was considerably low at 2.8 L/s. During the subsequent resting period, the longest in this study which lasted about five days, the subzero temperatures caused several inches of the basin to freeze. Consequently, the basin was thoroughly raked and harrowed with a pick before the next inundation period started. This action resulted in restoring the recharge rate to 3.5 L/s.

During the short dry up period on November 19 and 20, 1979 the basin was thoroughly raked again. This action had no effect on restoring the infiltration rate, probably because the weather during the resting period was wet and too short to have an effective improvement on the accumulated sludge at the bottom of the basin. Also, fluorescein dye and artificial brine, which were introduced in the basin just before the start of the flooding period, may have some negative effect on improving the recharge rate.

An obvious indication that physical clogging occurred affecting about 5 cm in depth of the bottom of the infiltration basin (Photo 10) is demonstrated by the data shown in Table 2. After the plastic liner was removed from the sides of the infiltration basin, the infiltration rate through the "fresh" sides of the basin was higher than through the basin's bottom, although the former area was 2.72 times smaller than the latter infiltration area.

During the initiation and termination of the infiltration stages the arrival of the initial flux of the effluent was carefully measured in OW-1. Likewise, "drawdown" of the effluent in the basin when the pump was shut off was also recorded at designated time intervals.

Using this information the increase of impedance was calculated. Since the restricting layer was not com-

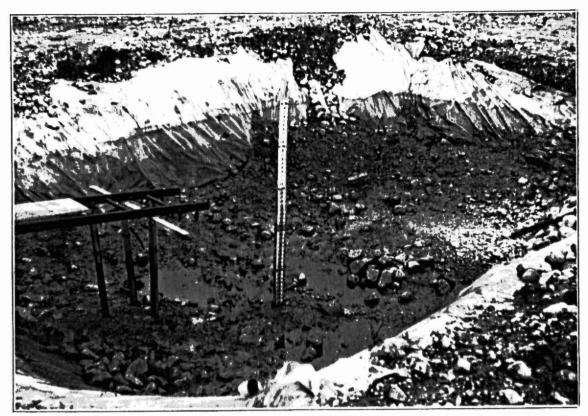


Photo 9. Deposition of suspended solids after 8 days of continuous infiltration indicates gradual basin clogging.



Photo 10. A thin crust at basin bottom began to develop at the end of the first inundation period. This basin clogging appears to have significantly affected very shallow depth.

pacted and biological clogging is unlikely a factor, the resistance should increase in proportion to the suspended solids added. The suspended solids in the lagoon effluent are mostly organic and easily compacted. They also tend to develop more rapidly in the lower sections of the infiltration basin (Photo 9).

WATER QUALITY STUDIES

General

In this section, a brief discussion of the water quality of the aquifer system which received the lagoon effluent as well as quality of surface water is given. The results of the qualitative analyses of water and the sewage lagoon effluent are given in Appendices E through J inclusive. These analyses can be used by individuals interested in further pursuing the qualitative aspect of this study.

Procedure

Numerous grab samples were obtained from the observation wells, test pits and from the selected surface water sampling stations. Several samples of the lagoon effluent were obtained directly from the infiltration basin and analysed for generally the same qualitative parameters. The frequency of sampling is summarized in Table 3.

A bailer 2.54 cm in diameter was used to obtain water samples from observation wells, whereas, a grab sample was taken from the test pits. The bailer was washed out with distilled water following each sampling. While the bottom of test pits 1, 2 and 4 are normally in the unsaturated zone, the rise of the water table due to infiltration provided favourable conditions to obtain water samples from these test pits as well.

Table 3. Frequency of sampling during the rapid infiltration study.

Date	Sept					С	ctob	er,	1979			1	Nove	embe	r, 1	979					Jan. 1980
Sampling Point	12	18	19	22	23	24	27	28	29	31	3	6	9	12	16	20	22	23	25	26	31
		bm						b									CC				
OW-1	C	CCC	CC		C	C		cm	L.	C	ca		С	С	С	b	ca		C	bm	bm
		bm						b												la un	ha
OW-2	C	CC	C			C		bm	l,	C	a		C			b	ca		C	bm	bc
OW-3	C	cmb				C		CIT	l.	C	a		C			b	ca		C	m	bc
OW-4	C	b	C			C		cn	1	C	a		C	C		b	ca		C	m	bc
OW-5	С	bm	C			С		cm	ı	С	a		С	С		b	ca aa		С	m	ba
OW-6	С	c cm	C			C				C	a		C		С	b	CCC		C	m	bc
OW-7	C	cm				C				С	a		C		C	b	ca		C	m	bc
OW-8	C	CCM	C			C				C	a		C			b				m	bc
OW-9	C	m	C			C				C	a		C			b	ca			m	bc
OW-10	C	ccm				C				C	a		C			b	ca			m	be
OW-11	•	0 0111									a					b					
OW-12	C										a					b					be
OW-13															C	b			C		be
OW-14	С										a					b			C		bc
OW-15	C										а	Ĺ,				b			C		b
OW-16											a					b			C		bc
OW-18										cm	а	L	C			b	ca			m	bc
OW-21					(2															
OW-22					(2							C								
										CC	ļ										
TP-1										mm	C	:	CC				ca		C	bm	
TP-2										CM	ı a	L.	C			b	ca		C	bm	-
TP-3										cm	ı a	Į.	C		C	b	ca		C		bc
TP-4										CI	ı a	ı	C			b	ca		C	bm	
TP-5										cm	ı a	L.	C		C	b	ca		C	bm	
RSR-1				bc			(cm		C	ä	ì	C			b	ca			m	bc
RSR-2												cm	C			b	ca				
RSR-3												CM	C			b	ca			m	
DOD 4				he			2	cm.		a .	a	,	C			b	ca			CCM	bc
RSR-4				bc				cm cm	,	С	ā		0			b	C	a		A (2)A	bc
P-5								cm			c					~	-				
P-6			-					cm			c	,			С		b	Ca	ı c	bm	bc
Lag. eff		CM	C								(-			0		~	-	-		

a Location of sampling points is shown in Figures 1 or 2.
OW - Observation well; TP - Test pit; RSR - Rocky Saugeen River; P - Pond water

Symbols: c - Chemical analysis

m - Heavy metals analysis

a - Anionic detergent analysis

b - Bacteriological analysis

Because there was a considerable amount of silt in water samples bailed from the well, for the purpose of uniform sample treatment, all groundwater samples were filtered except for the initial few sets, and for all samples taken for phenolic compounds and for carbons. The samples were filtered through a glass fiber filter paper with two micron openings.

In order to assess the effects of well bailing prior to taking water samples and filtering several sets of water samples from OW-1 and OW-6 were obtained and analyzed subsequent to various conditions. The results are shown in Table 4. It is apparent from this table that the differences between the chemical analyses of water samples taken from observation wells prior and after bailing were insignificant.

Quality of the Lagoon Effluent

During the inundation periods of the infiltration basin, several samples of the lagoon effluent were collected directly from the infiltration basin for qualitative analyses. A few of these samples were filtered so that they would be comparable with groundwater samples. However, the majority of effluent samples were not filtered prior to analyses.

The results of the qualitative analyses of the sewage lagoon effluent is given in Appendices G and J respectively.

The mean values of the results for four unfiltered chemical analyses of the sewage lagoon effluent were calculated and shown in Table 5. The loading by various chemical parameters from the conservative calculation of the total infiltrated lagoon effluent was calculated and also shown in Table 5.

Table 4. Results of chemical analyses of water samples obtained and analysed under various conditions.^a

Sampling Point	Observatio	n Well 1	(OW-1)	Observation Well 6 (OW-6)				
	Before bai	ling Aft	er bailing	Before bailing After bailing				
Constituent	Non- fi	ltered	filtered	Non-	filtered	filtered		
	filtered			filtered				
Hardness (calc.)	270	255	258	329	278	275		
Alkalinity as CaCO3	269	262	264	273	273	275		
Iron as Fe	8.4	0.20	1.32	38	0.30	0.60		
рн	7.50	7.88	7.69	7.45	7.99	7.70		
Conductance -			Ē					
umho/cm³ at 25°C	755	755	760	778	780	799		
Chloride as Cl	Cl 62.0		63.5	66.5	67.0	68.5		
Sulphate as SO ₄		36.5	36.5		34.5	37.5		
Calcium as Ca	63.0	58.0	57.0	104	63.0	61.0		
Magnesium as Mg	27.2	26.8	28.0	32.0	29.2	29.8		
Sodium as Na	56.0	57.0	56.0	53.0	55.5	41.8		
Potassium as K	7.25	7.65	7.75	6.00	4.75	1.95		
Free ammonia as N	2.7	1.0	2.9	2.0	2.2	2.4		
Total Kjeldahl as N	8.25	4.00	4.90	2.50	2.95	3.70		
Nitrite as N	0.08	0.145	0.03	0.24	0.22	0.03		
Nitrate as N	2.4	4.3	2.1	1.8	2.0	2.2		
Total P. as P	3.10	2.05	2.00	0.70	0.053	0.08		
Diss. react. P. as	P 2.15	1.85	1.75	0.20	0.001	L0.05		
Inorganic carbon		63.0	65.5		67.5	69.5		
Organic carbon		19.0	27.0		16.5	18.0		
Total carbon		82.0	92.5		84.0	87.5		
Phenols in ppb			Ll.O			L1.0		

a sampled on November 22, 1979

b All results except pH reported in mg/L unless otherwise indicated.

L - Refers to less than

The concentrations of suspended solids were relatively low during the time of year in which the study was carried out.

Groundwater Quality

Basic Considerations

The existing hydrogeological conditions in the study area and the hypsometric position of the sewage lagoon may suggest that exfiltration of the lagoon effluent has occurred particularly during the initial stage of the lagoon operation. This aspect could be further advanced and utilized to the existing conditions with an attempt to assess the renovation (which occurred in the past) of the effluent by natural hydrogeological environment.

This study showed that there was no convincing evidence that the quality of the shallower zone of the water table aquifer from which most of the samples were obtained was affected by the leaks from the lagoon. Therefore, such approach will be speculative at this time. For example, groundwater quality from sampled wells and test pits located between the lagoon and the recharge basin showed no indication that water quality is affected by the nearby lagoon. However, few sampled observation wells which obtain water from a somewhat lower elevation and which are located between the recharge basin and the river showed occasionally variation in water quality that may have been influenced by an "outside" source. This aspect is discussed in the later sections of this report.

In any case, in this study groundwater quality obtained prior to the commencement of the lagoon effluent infiltration is considered as the existing "natural" back-

Table 5. Quality of the infiltrated sewage lagoon effluent and calculated loading by recharged sewage lagoon effluent.

Constituent	Average value of four analyses (unfiltered) (mg/L)	Calculated loading (kg)
Hardness (calc.)	258	2069
Alkalinity as CaCO3	292	2342
Iron as Fe	0.07	0.56
рH	×	7.65
Chloride as Cl	57.25	459
Sulphate as SO4	34.12	274
Calcium as Ca	57.12	458
Magnesium as Mg	27.95	224
Sodium as Na	55.37	444
Potassium as K	8.0	64
Free ammonia as N	5.66	45
Total Kjeldahl as N	10.6	85
Nitrite as N	0.43	3
Nitrate as N	0.19	1.5
Total P as P	3.06	25
Dissolved reactive P.	2.45	20
Inorganic carbon	66.62	534
Organic carbon	33.5	269
Total carbon	100.12	803
BOD5	20.05	161
COD	284	2277
Suspended solids	19.33	155
Phenols (ppb)	F1	
A distribution of the state of		

F refers to less than

groundwater quality. Any subsequent changes of groundwater quality during this study were related to the recharge and compared with the initial water quality. It is considered that this approach did not alter the validity of any groundwater quality interpretation.

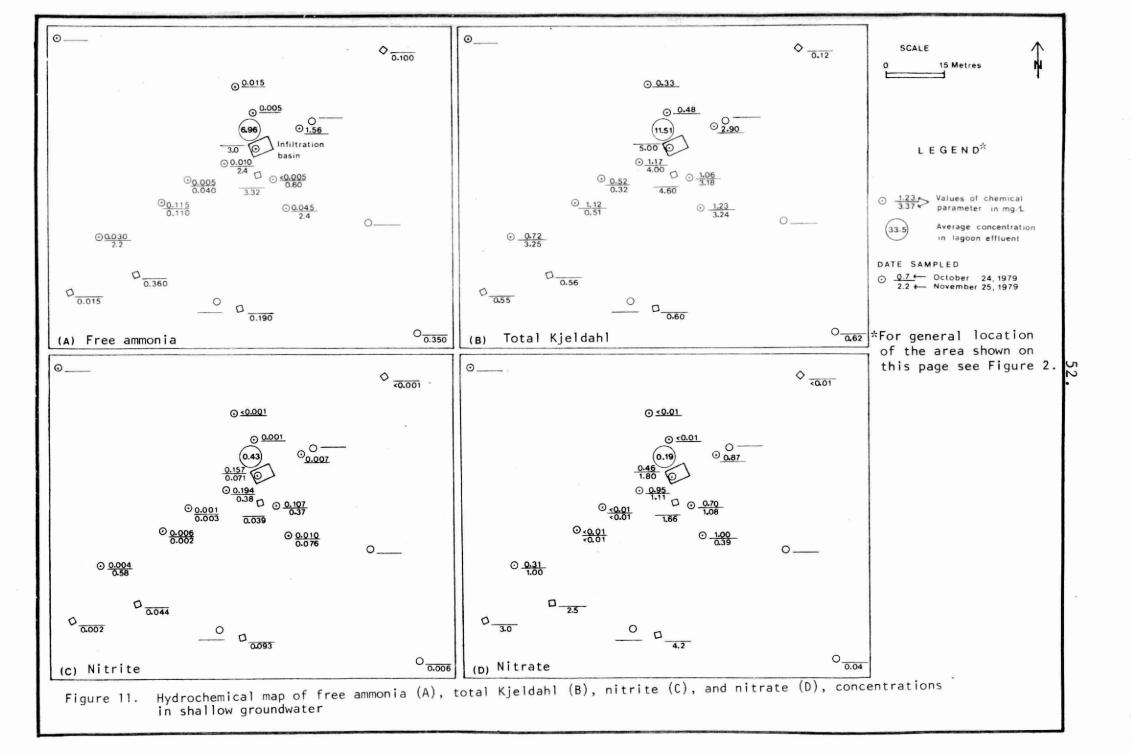
It is generally accepted and confirmed in this study that the concentrations of the majority of chemical constituents in groundwater samples obtained during this study mainly depend on several factors such as (a) distance of the sampling point from the recharge basin, (b) duration of inundation of the recharge basin before sample was taken, (c) thickness of the unsaturated zone below the recharge basin, (d) lithological composition along the travelling path before the sampling point, etc.

The results of the chemical, heavy metals and bacteriological analyses of groundwater are given in Appendices E and H respectively.

Nitrogen Cycle

The total nitrogen content of the sewage lagoon effluent generally ranged between 10.5 and 13 mg/L. Almost all of the nitrogen in the effluent was in the form of free ammonia and total Kjeldahl nitrogen.

The areal distribution of the major components of the nitrogen cycle at the beginning and at the end of the study is shown in Figure 11. This map shows that almost all nitrogen has been either converted into nitrate or attenuated at a distance of less than 40 m from the recharge basin along the shallow groundwater path toward the Rocky Saugeen River. However, the concentrations of the components of the nitrogen cycle show irregularity along the main



groundwater flow component at different distances from the basin. There are several factors that could contribute to this difference such as: (a) elevation of intake of the sampling station, (b) depth of intake below water table, (c) type of sampling point (observation well vs open test pit in which case nitrification may readily occur), (d) distance of the sampling station from the recharge basin, (e) the duration of basin flooding and quantity of effluent recharged before the sample was taken, (f) thickness of unsaturated zone (presence of oxygen), (g) travel time (time necessary for a population of nitrifying bacteria to develop), etc.

Nevertheless, the data indicate that almost all nitrogen which will be discharged to the ground surface some distance from the Rocky Saugeen River will be in the nitrate form. This nitrogen will be subject to further reduction by microbial processes thriving in the wet cattail environment along the river.

There are various processes which can remove nitrogen from waste water while it moves through soil, including adsorption of ammonium to the clay, silt and organic fraction of the soil, fixation of ammonium by the organic fraction, fixation of nitrogen by micro-organisms, and nitrogen uptake by vegetation. Volatalization of ammonia and denitrification are the only processes which cause a net removal of nitrogen. The other processes merely store nitrogen in the soil.

Research shows that the mechanism of nitrogen denitrification occurs when a new basin flooding is started. When inundation commences there is entrapped air in the soil for nitrification of ammonium during the initial part of the flooding period. However, as the oxygen is consumed, anaerobic conditions will begin to develop where nitrate and organic carbon can both be present, creating conditions

favourable for denitrification. With continued flooding all nitrogen will stay in the ammonium form, which can be adsorbed by the soil particles. If flooding is not stopped before the cation exchange complex in the soil is saturated with ammonium, increased ammonium levels in the receiving groundwater can be expected. When the flooding is stopped, air will enter the soil and the resulting aerobic conditions in the shallow unsaturated zone will enable nitrification of Some of the nitrate formed in this the adsorbed ammonium. process may diffuse to anaerobic microenvironments in the same soil and denitrification can occur if organic carbon is also present. Some of these nitrate ions may also mix later with the newly infiltrating water when a new flooding period is started and move down to anaerobic environments where denitrification may occur.

In conclusion, setting up proper inundation dry up periods is of great importance with respect to the denitrification processes and therefore nitrogen removal.

Phosphorous

The concentration of phosphorous in the lagoon effluent and the detrimental impact on water quality in the Rocky Saugeen River is one of the prime concerns with respect to the present mode of the sewage treatment system operation at Markdale.

Phosphorous removal seems to occur rapidly as the effluent water moves through the sand and gravel deposits (Figure 12). Although a slight increase of the dissolved reactive phosphorous is indicated (as the infiltration proceeded) at a distance of 10 m there is no doubt that favourable conditions do exist in the sand and gravel de-

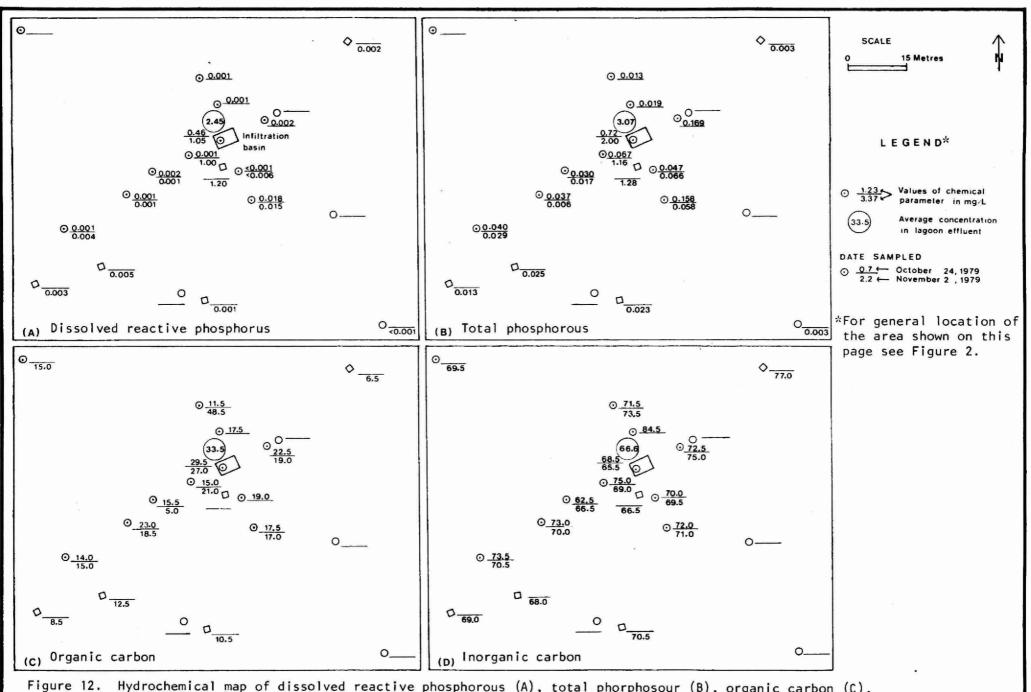


Figure 12. Hydrochemical map of dissolved reactive phosphorous (A), total phorphosour (B), organic carbon (C), and inorganic carbon (D), concentrations in shallow groundwater.

posits for the removal of phosphorous from the effluent. The presence of a calcium-rich environment and iron and aluminum oxides in the granular materials are the main contributing factors for the removal of phosphorous. It is considered, however, that the removal of phosphorous is probably due to precipitation of calcium-phosphate complexes, which are formed in the slightly alkaline and calcium-rich environment of the effluent water as it moves through slightly silty sand and gravel deposits.

Total removal of phosphorous is very encouraging since it is considered the main constituent to have an adverse effect on river water quality.

Phosphates are the only constituents of the effluent that precipitate in quantity in the soil. This precipitator may affect the porosity and hence the hydraulic conductivity of the sand and gravel in the long-term operation. However, considering the pore space of the sand and gravel deposits to be about 30 percent and relatively small amount of precipitate (oxy-apatite), it will probably be a very long time before phosphate accumulation in the sands and gravels will have a significant effect on the hydraulic performance of the system.

Carbons

The areal distribution of the organic and inorganic carbons is shown in Figure 12. It is indicative that at a distance of less than 40 m from the basin, the organic carbon is removed to the background concentrations.

Phenolic Compounds

Of six analyses of sewage lagoon effluent five showed concentrations of phenolic compounds to be less than 1 part per billion (ppb) and only one sample indicated that the phenols content was 25 ppb. However, this sample was taken in January, 1980, directly from the lagoon under ice conditions long after field work of this study was completed.

Contrary to the very low concentration of phenolic compounds in the lagoon effluent, several samples of groundwater showed considerably high concentrations of phenols. These include samples obtained from OW-4, OW-5, OW-8 and OW-9. The highest values of phenol concentrations in groundwater were in the samples collected on the day when the flooding of the basin started (October 19, 1979). This may suggest that these phenolic compounds are naturally occurring in the shallow groundwater. Another more likely explanation is that since two of these sampling points obtain water from a deeper elevation than other sampling stations their water quality is influenced by the deeper groundwater flow system whose quality may have been affected by exfiltration of the lagoon effluent. Increased concentration of phenolic compounds in the deeper sampling wells is in tune with the sporadic increased concentrations of few other chemical parameters in these wells. A third possibility is that the groundwater was slightly contaminated with phenols during the drilling of the observation wells.

Biochemical and Chemical Oxygen Demand

Since the majority of groundwater samples contained considerable amounts of silt and consequently were filtered, the results of biochemical (${\rm BOD}_5$) and chemical

oxygen demand (COD) were influenced by these factors. For this reason only few sets of initial samples were analysed for biochemical and chemical oxygen demand. Thus, wastewater renovation with respect to these constituents cannot be assessed.

 $$\operatorname{\mathtt{The}}$\; \operatorname{\mathtt{BOD}}_5$$ of the sewage lagoon effluent was relatively high and in the 20 to 30 mg/L range.

The COD of the lagoon effluent usually was in the 280 mg/L range.

Total Dissolved Solids and Major Ions

Total dissolved solids were calculated by measuring the specific conductance of water samples and multiplying it by a factor of 0.65.

The concentrations of total dissolved solids of the lagoon effluent were in the 400 to 515 mg/L range (Table 5) and concentrations in native groundwater were in the 400 to 470 mg/L range.

A brief discussion regarding the behaviour of the major ions in groundwater during the rapid infiltration study is given in the following paragraphs.

In general the concentrations of calcium fluctuated between 55 and 104 mg/L with maximum value found in OW-6 and the lowest value in TP-1. The concentrations of calcium in unaffected groundwater were between 70 and 80 mg/L.

Magnesium content in the native groundwater was around 30 mg/L but it may go as high as 70 mg/L. In renovated water the maximum concentration of magnesium was 62 mg/L (OW-4) while the minimum value was 23.6 mg/L (OW-1).

Concentrations of sodium in renovated water fluctuated between 67 mg/L (TP-1) and a minimum value of 3.8 mg/L (OW-4). The background concentration of sodium in groundwater was about 11 mg/L.

Background content of potassium in groundwater was about 0.7 mg/L. The maximum concentration of potassium in affected groundwater was reported at 9.8 mg/L (OW-1), whereas the minimum content was 0.7 mg/L (OW-10).

Chloride concentrations in renovated water fluctuated between a maximum value of 91 mg/L (OW-5) and a minimum concentration of 13.5 mg/L (OW-3). The background value of chloride in groundwater was about 20 mg/L.

The background concentration of sulphate in ground-water was around 10 mg/L. In renovated water the content of sulphate fluctuated between a maximum value of 61 mg/L (OW-4) and a minimum concentration of 4.0 mg/L in observation well 10.

рН

The pH of the lagoon effluent was usually between 7.69 and 8.01 and pH of the renovated water was between 7.35 and 8.0 mg/L. The pH of the native groundwater ranged from 7.5 to 7.7 mg/L.

Anionic Detergents

Anionic detergents are not found in natural conditions. Therefore, the presence of these substances in natural waters indicates contamination by domestic or industrial wastes.

Two sets of samples were obtained from most of the monitoring stations and analysed for anionic detergents (November 3 and November 21, 1979). The results of the November 3 set of samples indicated the concentration at or below the detection limit of 0.1 mg/L. The only exception to this, was OW-1 which had a concentration of 0.4 mg/L.

The results of the second set of samples showed a slight increase of anionic detergent concentrations in several observation wells and test pits. However, none of these observation points had values exceeding 0.4 mg/L.

The concentration of anionic detergent in the sewage lagoon effluent of only one sample was 0.1 mg/L.

Heavy Metals

It is indicative that non-filtered samples of groundwater have relatively high concentrations of heavy metals, in particular aluminum with concentrations of several hundred milligrams per litre, then manganese and zinc not exceeding the concentrations of 10~mg/L. In contrast, the concentrations of heavy metals in the sewage lagoon effluent were exceptionally low with the exception of zinc (14 mg/L).

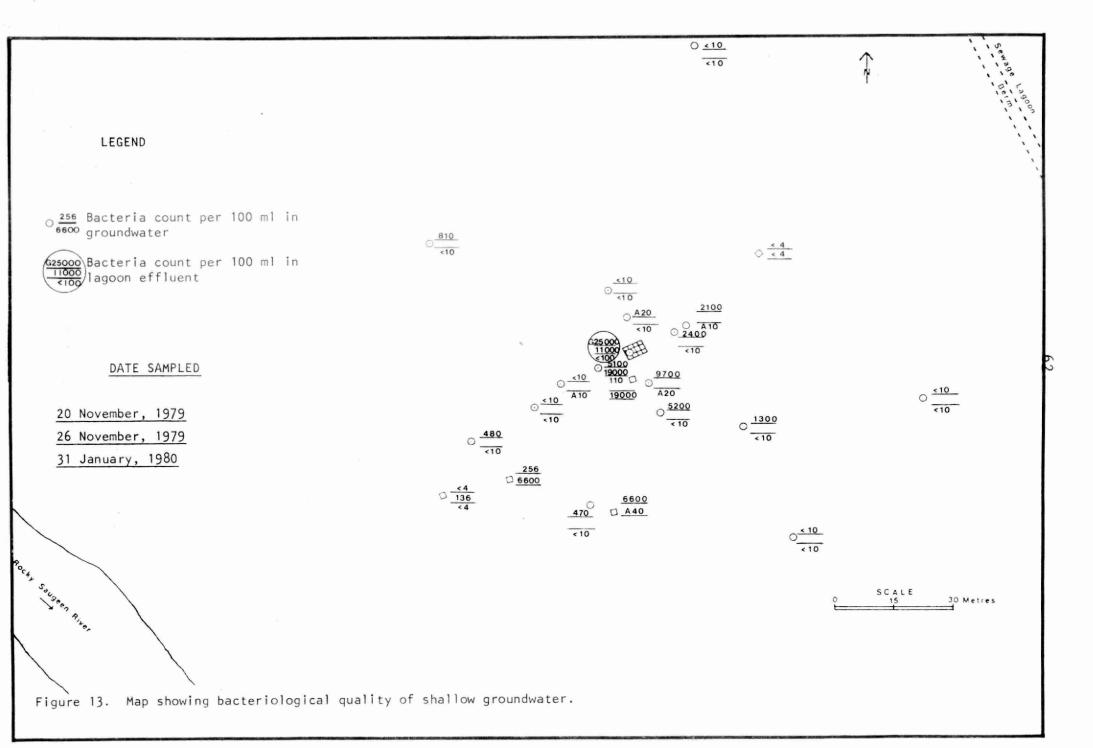
The results of heavy metal analyses on filtered samples of groundwater show a different picture with very low concentrations. This suggests that the elevated concentrations of heavy metals in non-filtered samples are associated with fine soil particles present in these water samples.

Bacteriological Quality

The fecal coliform bacteria density in groundwater is shown in Figure 13 which indicates that some fecal bacteria were reaching distances of 60 m from the recharge basin. It is expected that if sequences of long flooding periods were held, the fecal coliform count in groundwater would tend to increase when newly infiltrated water arrives at the observation wells and then would decrease as infiltration continued.

The coliform removal was probably due to "filtering" at the surface and mortality in hostile and competitive subsurface environments. Filtering and competition, and hence the fecal coliform removal, can be expected to increase with continued infiltration because of increased surface clogging and increased bacterial populations in the soil.

It is reasoned that any bacterial population which may reach the groundwater discharge area will be further reduced by microbial activities present in wet peat deposits along the river.



Surface Water Quality

In order to assess whether there was any measurable effect on surface water quality, four sampling stations were established at strategic locations in the Rocky Saugeen River. In addition a few samples were obtained from two existing surface water ponds located in the immediate vicinity of the sewage lagoon. These two monitoring stations provided information with respect to possible infiltration of the effluent into the shallow water table aguifer directly from the lagoon.

The locations of the surface water sampling stations are given in Figure 1, while the results of qualitative analyses are provided in Appendices F and I.

Chemical Quality

The results of the chemical analyses obtained from the four sampling stations in the Rocky Saugeen River show that there was no measurable increase in any of the chemical constituents in the river due to the infiltration of about 8,020 m³ of sewage effluent to the shallow groundwater system. This is understandable taking into account effluent dilution by groundwater, surface water and other attenuating processes in the groundwater, hydrologic and surface systems existing between the upper limit of the groundwater discharge area and the Rocky Saugeen River.

In actuality, the water quality of the river has much lower concentrations of almost all chemical constituents than background concentrations in the shallow groundwater except for pH and nitrate which are somewhat higher in the River. The river water is slightly on the alkaline side compared to the groundwater.

Concentrations of the anionic detergents were below the detectable limit (0.1 mg/L) in the samples obtained on November 3, 1970 at two sampling stations in the Rocky Saugeen River (upstream and downstream from the lagoon). No increase in the content of anionic detergent was revealed by the second set of samples taken from four sampling stations in the Rocky Saugeen River on November 21, 1979.

Although the elevation of the water table in the pond located immediately west of the lagoon is lower than the elevation of the bottom of the lagoon, it appears that water quality in the pond has not been affected by the leakage from the lagoon. It should be noted though, that during the course of pumpage of the effluent into the infiltration basin one hose developed a leak and as a consequence a small amount of the effluent found its way (along the ditch parallel to the berm) into this pond. This resulted in slight increases in the concentrations of several chemical parameters including a few components of the nitrogen cycle and phosphorous (Appendix F). A similar trend was exibited by anionic detergents which increased to 0.1 mg/L.

Heavy Metals

The concentrations of heavy metals in the river were exceptionally low at all four sampling points indicating that there was no measurable influence on their concentration in the river caused by the effluent infiltration, or due to the existance of the sewage system at this locality (Appendix F).

Bacteriological Quality

Bacterial counts in river water samples taken upstream and downstream from the study area were quite low (Appendix I). It is therefore concluded that rapid infiltration of the lagoon effluent into the coarse granular deposits has had no adverse effect on the bacteriological quality of the river water.

The presence of fecal coliform bacteria in the pond water located immediately west of the lagoon was most likely caused by leakage of the effluent from the pumping equipment which found its way to the pond.

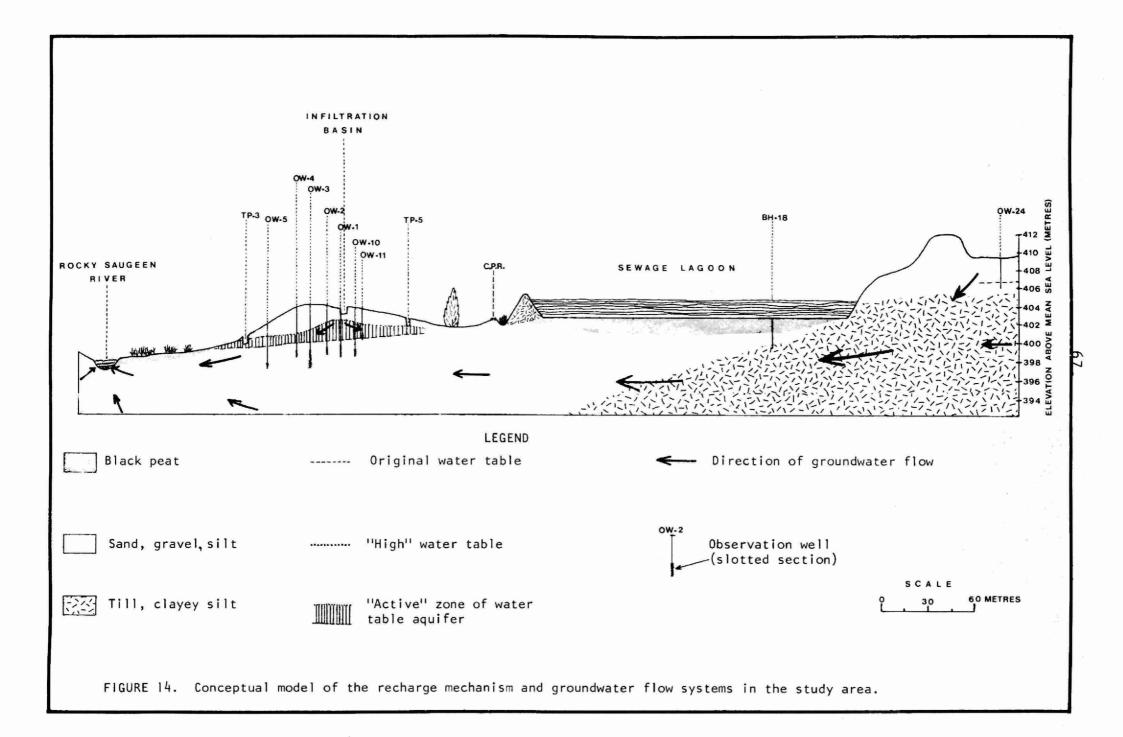
DISCUSSION

Significance of the Recharge Mechanism and Groundwater Flow Systems

The resulting effect of the infiltration of the sewage lagoon effluent into granular deposits was the development of a groundwater hydraulic mound. Initially, the percolating water will move downward and then after reaching the water table, it radiates in all directions until the mounding effect has reached a steady state condition. From here on, the lateral movement of water will be only in the direction of the original groundwater gradient, that is towards the Rocky Saugeen River.

It is expected that a considerable mixing with native groundwater will occur to a certain depth largely depending on the direction of groundwater flow, the quantity of recharged wastewater and the permeability of the granular deposits. In this study, because large quantities of wastewater were infiltrated in a relatively small area, it appears that a bulk of recharged wastewater occupied or "floated" above the native groundwater and hence the lateral movement prevailed. Therefore, only a limited portion (up to 2 m) of the sand and gravel aquifer, an "active" (dynamic) zone, largely participated in the wastewater renovation (Figure 14).

In view of this situation, it is important to recognize the significance of the qualitative results of groundwater in relation to the depth from which the samples were obtained. For example, it is obvious that samples taken from the open test pits (TP-1 to TP-5) will be representative of the "active" zone of the shallow aquifer,



whereas some of the deeper observation wells would depict a portion of the aquifer which is less affected by infiltrated wastewater.

Furthermore, water quality in these deep wells seems to be affected by an "outside" source most likely agricultural fertilizers. Exfiltration from the lagoon may also be a contributing factor to some of the discrepancies in water quality results in the deeper observation wells. The mechanism of groundwater flow and the potential pollution sources likely influencing groundwater quality in the study area are schematically illustrated in Figure 14. This figure shows that any exfiltration from the lagoon which is probably insignificant at this time, because of self-sealing processes at the bottom of the lagoon would have an affect on the deeper portion of the sand and gravel aquifer system. The results of qualitative analyses obtained from several sampling stations support this conceptual model.

The bulk of the renovated water will be discharged in the newly formed groundwater discharge zone created by the rise of the water table, as well as in the original groundwater discharge zone, both constituting the section along the river about 80 m wide. The discharged groundwater will undergo further renovation while travelling overland through a wet, small cattail environment.

Finally, a small quantity of groundwater from the shallow aquifer will probably reach the river directly through the banks and river bed.

Potential Hydraulic Loading

The infiltration rates generated during the initial period of this study (first five days) could be used as the potential or bench-mark infiltration rates since they were basically affected only by the permeability of the underlain sand and gravel deposits and were not influenced by clogging factors.

Using this data, the potential application rate could be as high as $32~\text{m}^3/\text{d/m}^2$. This value should be increased by the amount of infiltration which will occur through the contact area of the effluent with the soil on three sides of the infiltration basin. This additional recharge rate will largely depend on the effluent depth maintained in the recharge basin.

More conservative loading rates are given in Table 2. Recharge rates given in this table represent an average rate generated during certain periods of time, under specific hydraulic and infiltration conditions of the recharge basin. For this reason, they should be considered more reliable than the former values.

The operational sequence of an infiltration system (inundation - dry up periods ratio) should be considered in close conjunction with optimal loading rates. In other words, inundation - dry up periods are highly influential factors to the optimal recharge rates.

General Design Aspects

A large-scale infiltration system for the renovation of wastewater from the sewage lagoon should be considered for design. Although not in abundance, the most favourable land appears to be the area located immediately northwest of the infiltration basin used in this study (Photo 3). The system may consist of a chain of elongated

recharge basins parallel to the Rocky Saugeen River. The effluent could be distributed into the basins by a channel which could run along the recharge strip. Since a low suspended solids content in the effluent would be advantageous in order to prolong flooding periods and therefore require minimum maintenance of the basins, a sedimentation reservoir could be constructed at the head of the channel system carrying the effluent to the recharge basin.

The following criteria may be used for the design of a system for renovating sewage lagoon effluent by ground-water recharge with infiltration basins at this locality.

- During infiltration the depth to water table below the recharge basin should be at least 1.5 m. Maintaining a sufficiently thick unsaturated zone is necessary for aerobic decomposition of organic matter before it reaches the water table.
- Since the above criteria may be difficult to achieve, an alternative would be oxygenation of sewage lagoon effluent by "splashing" into the basin.
- 3. The bottom of a recharge basin should be relatively flat with up to 1.0 m of relatively uniform sand maintained at its bottom. This imported filter material should be of slightly lower permeability than the parent material. The relatively thick sand filter will allow a longer time between additions of more sand, as clogged layers are removed.

Some Aspects of Recharge Basin Management

It would be difficult to predict the optimum wet/dry ratios without extensive operational experience.

This ratio will largely depend on the quality of the effluent at different times of the year and it may change accordingly. Therefore, the ratio should be determined by experimentation with the effluent in question. The inundation period is dictated mainly by aerobic conditions in the sand filter and in the ground, whereas the dry up period serves to dry the sediment and algae layer and to allow oxygen to penetrate the sand filter and the unsaturated material below it (Hore, 1980).

Initially, a large operation system with infiltration basins, longer inundation periods alternated with dry
up periods (suggested initial ratio is 10:6) should be used
to achieve maximum efficiency. Because of relatively high
suspended solids in the sewage lagoon effluent, basically
present in the form of organic matter, it is a matter of
time until these suspended solids accumulate at the bottom
of the basin (sludge layer), thus reducing infiltration rate
to a certain degree. It is expected that upon drying up,
the sludge layer would shrink and brake up into curled up
flakes.

Initially, dry up periods may help to restore original infiltration rates without the help of any device. However, if necessary, at the end of dry up periods the basin could be raked or harrowed using manual labour only (hand operation). This procedure may be repeated at the end of each dry up period until such time that infiltration rates are reduced by build up of a thicker sludge layer thus necessitating its physical removal. Again, hand operation should be employed to accurately remove a small amount of ("shaving" of the basin bottom) the sludge and sand filter followed by harrowing, if necessary. Use and movement of any heavy vehicle will have a negative effect on the basin's performance because it will compact the sand filter thus causing permanent reduction of its permeability and consequently

decreasing infiltration capabilities in a recharge basin. In any event, the removal of the top of the sand filter should be done as infrequently as possible to maintain the biologically active (top) layer in the sand filter (Hore, 1980).

After a certain period of basin operation, it may be necessary to replace the thin (top) layer of granular material at the bottom of the basin. The usual practice of filter removal and replacement is to place a uniform layer of sand up to 1 m thick in the bottom of a basin, and remove several centimeters of this sand as it becomes clogged. Another layer of sand is placed in the basin only after the removal of the top layer reduces the original filter to about 0.3 m (Sibul, 1980). The medium sand filter is used because (a) this type of material would act as an immediate filtering media and would retain most of the suspended solids near the bottom of the basin thus minimizing the possibility of formation clogging at a greater depth and (b) it was found that granular material with larger pore space recovered its infiltration capacity more slowly than those with small pore spaces.

The effect of rainfall during dry up periods on basin performance could be significant in either reducing or improving the infiltration rate. It was observed that rain drops tend to brake the accumulated sludge layer (while it was still wet) and wash it into the soil.

An example of the effect of rainfall and snow on the basin partially clogged with a sludge layer is illustrated in Photos 11 and 12. Photo 11 shows the situation of the "unshaved" bottom of the basin at the end of the infiltration study (November 26, 1979). Rain and snow actions have "purified" it as illustrated by Photo 12 taken on April 19, 1980. Actually, all suspended solids from the surface were moved "inside", thus affecting several centimetres of granular material immediately below the bottom of the basin.

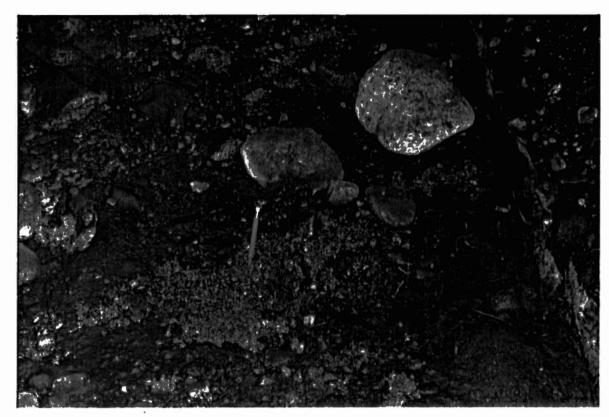


Photo 11. A thin sludge layer accumulated at basin bottom at the end of the infiltration study November 26, 1979 had slightly reduced infiltration rate.



Photo 12. Rainfall and winter weather conditions had "shaved off" the basin left partially clogged with a thin sludge layer at the end of the infiltration study shown in Photo 11.

Photo taken on April 19, 1980.

Other aspects of an efficient management of an infiltration system would be learned during its actual operation.

Winter Operating Constraints

In a climatic region such as the Dundalk upland in which the Village of Markdale is located, with sub-zero temperatures which may last over four months, operation of an infiltration system for the lagoon effluent during the winter months could pose serious difficulties. Notable concerns are discussed as follows:

- During the sub-freezing temperature, as long as the inundation period is continued, there is little danger (with some exceptions) that the bottom of the basin may freeze resulting in building up of ice and snow. Build up of ice and snow would disrupt smooth continuation of the operation of the system.
- 2. Pump failure, or basin clogging would eventually necessitate stoppage of the basin inundation resulting in accumulation of ice and snow.
- 3. A severe blizzard accompanied with strong winds may generate large snow drifts into the infiltration basin, thus obstructing continuous operation. Although infrequent, this could provide a seriously disruptive condition.
- 4. Research shows that application of waste water from the stabilization pond did not prove to be effective for melting snow and preventing ice accumulation. Therefore, continuous flooding at the time of the earliest

hard freeze might provide an insulating cover of snow thus enabling continuation of recharge into early winter. This would be possible provided that snow would not accumulate in such large quantities from the blizzard or snow drifts.

- 5. It is also anticipated that a greater depth of waste water in the infiltration basin would be advantageous during the early winter to extend the infiltration period under the ice. Therefore, if the flooding period is continued at least one month beyond the freezing date, the necessary winter storage of waste water would be less than four months.
- 6. Because of freezing soil conditions and other adverse weather effects, basin cleaning and drying in the winter will not be practical. If the infiltration rate is reduced to near zero while water remains in the infiltration basins, their renovation in the spring will be difficult. The remaining water would have to be pumped out in order to clean the basin filters (Hore, 1980).

Considering these adverse factors, it is doubtful that the efficient (if any) operation of the system during the winter months can be accomplished.

CONCLUSIONS

- 1. The results of a rapid infiltration study of sewage lagoon effluent renovation with an infiltration basin, carried out from October 19 till November 26, 1979, showed that wastewater can be effectively renovated by this method. In an operational system, the basin would be located immediately northwest of the basin used in this study, but generally parallel with the Rocky Saugeen River.
- 2. An infiltration of over 700 m/year of the lagoon effluent can be obtained with tentative flooding periods of 6 to 10 days alternated with dry up periods of 5 to 6 days. These are tentative suggestions, but experience gained from an operational system would provide data for optimal and more firm operational schedules.
- 3. The response of the water table aquifer beneath the infiltration basin and in close proximity to it, enabled an evaluation of the vertical and horizontal hydraulic conductivity of the aquifer. The values of hydraulic properties of the shallow aquifer can be used in the design of an operational infiltration system to predict underground detention times and water table configuration for various shapes of the recharge basin.
- 4. The attenuation processes in sand and gravel deposits effected complete removal of dissolved reactive phosphorous.
- 5. Free ammonia was entirely converted into total Kjeldahl nitrogen at a distance of less than 40 m, whereas over 40 percent of Kjeldahl nitrogen was removed or converted into nitrate at the same distance from the

recharge basin. Nitrate concentrations were still relatively high. Short, frequent inundations of the basin alternated with equally long dry up periods, would probably yield the most complete conversion of nitrogen in the effluent water to nitrate in the renovated water. This short inundation period may be performed when the need for it arises.

- 6. The renovation processes in the granular materials yielded almost complete removal of fecal coliform bacteria at a distance of about 60 m from the recharge basin.
- 7. There was no indication of gradual clogging of the formation at depth, although a thin layer of sludge matter gradually accumulated at the bottom of the infiltration basin thus reducing, to some degree the recharge rate. With periodic "shaving off" and the replacement of the granular material at the bottom of the basin a rapid infiltration system for renovating sewage lagoon effluent should have a long, useful life.

RECOMMENDATIONS

Based on the results of this study it is apparent that the approach using an infiltration basin system can be utilized on an operational scale for tertiary treatment of sewage lagoon effluent. However, a careful evaluation of the present sewage flow, 15-to 20-year projected sewage flow and subsequently required storage capacity in conjunction with the operational mode of the rapid infiltration system should be carefully assessed before the design stage is undertaken.

In the light of some uncertainty with respect to lithological conditions in the study area (as indicated by geophysical survey) it is required that

 At least three test holes should be completed into the bedrock in order to obtain more accurate information as to the thickness and nature of the overburden.

The data obtained may clarify any possible impact on groundwater use in the area (Markdale municipal wells). This should be completed before the design stage is undertaken.

It appears that the most favourable area (considering just the area between the river and CPR tracks) for the establishment of an operational system is located immediately northwest from the infiltration basin used in this study. Additional field investigation may be necessary to confirm that this prospective area consists of permeable sand and gravel deposits. This could be accomplished with the help of a backhoe.

However, the available land between the Rocky Saugeen River and CPR tracks is insufficient to accommodate the entire sewage lagoon effluent in an operational infiltration system. For this reason, the field investigation should also include an assessment of the availability of the additional land which will be suitable for sewage effluent treatment by rapid infiltration. The property immediately west of the sewage lagoon appears to be quite promising.

If the permanent infiltration system is installed, the important water quality parameters should be monitored as well as the long term water level response to recharge. The formation clogging aspects should also be studied. Underground detention time should be monitored along with renovated groundwater "polishing" by biological processes in the groundwater discharge area in the wet section near the river.

It is suggested that the municipality engage a consultant for the expansion of the sewage works based on a groundwater recharge approach philosophy. Data collected during this study and presented in this report should serve as a basis for the design of an operational system with infiltration basins.

The consultant should work closely with the staff of this Ministry in view of the innovative nature of this treatment method and to ensure that all concerns are adequately resolved before proceeding to final design.

ACKNOWLEDGEMENTS

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Messrs. Denis Veal and Daniel Brown read the manuscript and made many valuable suggestions.

Messrs. Ron Hore and Ulo Sibul, Water Resources
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Branch carried out a geophysical survey in the study area.

The field work was carried out by the author and Mr. Tom Ervasti. The levelling survey was performed by Mr. Brian Jaffray and grain size analyses by Mr. Steve Check. Drafting and assemblage of the appendices were carried out by Miss Nancy Rennie and Mr. Francois Sylvestre.

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APPENDIX A

bservation ell or test it		Ground elevation a.m.s.l.	Total well (m) At time of	(pit) depth On April 19	Depth to water level at time of completion	"Screen" Depth to	details Litho	Description
	2	(m)	completion	1980	(m)	(m)	(m) (m)	
0W-1	10/09/79	403.97	5.49	4.79	3.53	3.49	2.0 0 - 0.83 0.83- 3.8 3.8 - 5.5	Sand, gravel (recharge pit) Sand, gravel, minor silt
							3.0 - 5.5	Sand, gravel, silt, clay layers
0W-2	05/09/79	404.02	5.30	5.14	3.61	4.30	1.0 0 - 4.0	Sand, gravel, slight
							4.0 - 5.5	Sand, gravel, some clayey silt
0W-3	05/09/79	404.25	7.0	6.85	4.15	5.0	2.0 0 - 4.0	Sand, gravel with cobbles
							4.0 - 4.9	Sand, silty with gravel
							4.9 - 7.6	<u>Till</u> , clayey silt, fine gravel
OW-4	05/09/79	403.97	5.6	5.78	3.77	4.60	1.0 0 - 3.1	Sand, gravel, some
							3.1 - 6.1	Sand, gravel, some
0W-5	05/09/79	402.30	5.20	5.06	2.49	4.20	1.0 0 - 3.7	Sand, gravel, minor
							3.7 - 5.5	<u>Gravel</u> , <u>cobbles</u> , silt

Observation well or test pit		Ground elevation a.m.s.l.	Total well (m) At time of completion	pit) depth On April 19 1980	Depth to water level at time of completion (m)	"Screen' Depth to top (m)	details Length (m)	Litho Depth (m)	logic log Description
0W-6	10/09/79	403.95	5.70	4.59	3.36	4.70	1.0	0 - 4.0 4.0 - 5.8 5.8 - 6.1	Sand, gravel, slight silt Sand, gravel, silt layering Till, clayey silt, massive
0W-7	10/09/79	403.90	5.34	4.82	3.31	4.34		0 - 4.1 4.1 - 5.9 5.9 - 6.1	Sand, gravel, slight silt Sand, gravel, silt layering Till, clayey silt, massive
0W-8	05/09/79	404.16	5.30	5.00	2.77	4.30	1.0	0 - 5.5	Sand, gravel, some cobbles, slight silt
0W-9	05/09/79	404.04	5.30	5.4	3.76	4.30	1.0	0 - 5.5	Sand, gravel, some cobbles, slight silt
OW-10	10/09/79	403.90	5.34	4.38	3.23	4.34	1.0	0 - 4.0	Sand, gravel, slight silt Sand, gravel, silt, clay layers
OW-11	15/08/79	403.68	3.40	3.5	3.26	2.40	1.0	0 - 2.4	Gravel, cobbles, sand minor silt Sand, medium gravel, light silt

Observation well or test pit		Ground elevation a.m.s.l.	Total well (m) At time of	pit) depth On April 19	Depth to water level at time of completion	"Screen' Depth to top	details Length	Lithol Depth	ogic log Description
		(m)	completion	1980	(m)	(m)	(m)	(m)	
0W-20	20/07/79		3.50		dry	2.5	1.0	0 - 2.7	Cobbles, boulders, coarse sand and gravel, silt
								2.7 - 3.5	Gravel, coarse with silty sand
0W-21	20/07/79	402.78	2.70	2.14	2.23	1.7	1.0	0 - 1.5	Cobbles, boulders with
								1.5 - 2.7	sand, silty Sand, gravel, little silt
0W-22	20/07/79	407.54	2.71	2.86	2.69	1.7	1.0		Gravel, cobbles sand Silt, sandy, dense Till, clayey silt
0W-24	14/09/79	409.83	3.50	2.85	3.12	2.5	1.0	0 - 3.5	Sand, medium with gravel, slight silt
TPB-1	20/07/79		3.5	backfilled	dry			0 - 2.8	Gravel, cobbles, very
								2.8 - 3.5	silty with sand Silt, sand, clay
TPB-2	20/07/79		3.9	backfilled	dry			0 - 2.1	Cobbles, boulders, coarse
								2.1 - 3.9	sand, gravel, silty Silt, sandy, dense
TPB-5	20/07/79		3.9	backfilled	dry			0 - 2.0	Gravel, cobbles, silty
								2.0 - 3.9	Gravel, coarse, silty

Observation well or test pit		Ground elevation a.m.s.l. (m)	Total well (m) At time of completion	(pit) depth On April 19 1980	Depth to water level at time of completion (m)	Depth to top (m)	details Length (m)	Lithol Depth (m)	ogic log Description
TPB-6	20/07/79		3.9	backfilled	dry			0 - 3.9	Gravel, cobbles, with coarse sand, silty
TP-1	29/10/79	403.9	2.5		dry				Top soil, silty Sand, gravel, occasionally silty
TP-2	29/10/79	402.06	1.5						Top soil Gravel, sand, coarse minor silt
TP-3	29/10/79	400.60	1.0						Top soil, silty Gravel, sand, silty clayey
TP-4	29/10/79	402.32	1.5						Top soil, silty, sandy Sand, gravel, silt
TP-5	29/10/79	402.78	1.4						Top soil <pre>Gravel, sand, silty, clayey</pre>

APPENDIX B

GRAIN SIZE ANALYSES OF SURFICIAL DEPOSITS

GRAIN SIZE DISTRIBUTION

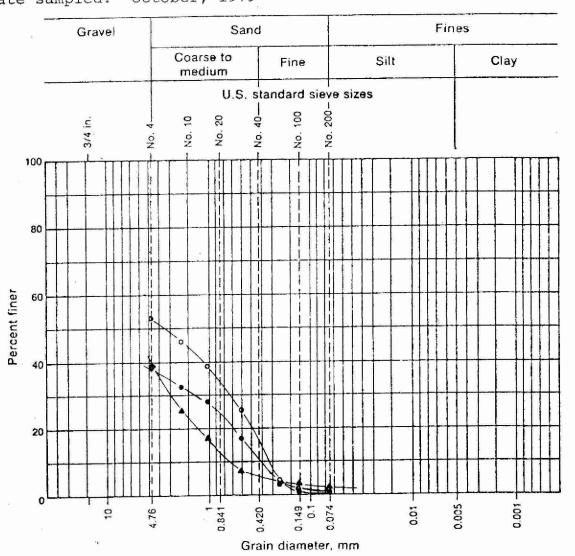
Project Study of sewage lagoon effluent renovation by rapid infiltration

Location of Project Markdale Boring No. Sample No.

Description of Soil Depth of Sample

Tested By. Steve Check Date of Testing January, 1980

Date sampled: October, 1979



LEGEND

Test pit no.	Depth (m)	Symbol	Soil Description
1	0.61		Gravel with coars
1	1.22	-0	Gravelly sand wit
1	1.83	-•	Gravel with coars

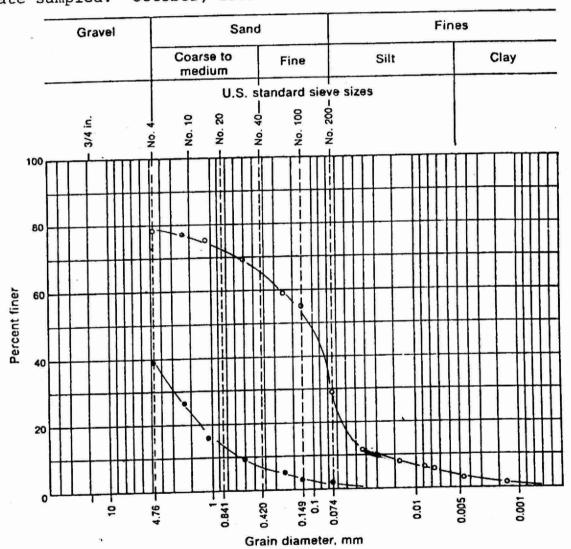
Gravel with coarse sand, some fines Gravelly sand with some fines Gravel with coarse sand, some fines

GRAIN SIZE DISTRIBUTION

Project Study of sewage lagoon effluent renovation by rapid infiltration Location of Project _____ Markdale ____ Boring No. _____ Sample No. _____

Description of Soil ______ Depth of Sampl. _____ Date of Testing _____ January, 1980 Steve Check

Date sampled: October, 1979



LEGEND

Test pit no.	Depth (m)	Symbol	Soil Description
2	0.61	-•-	Gravel with coarse sand, some fine
3	0.46	-0-	Gravelly sand with some silt and clay

fines

GRAIN SIZE DISTRIBUTION

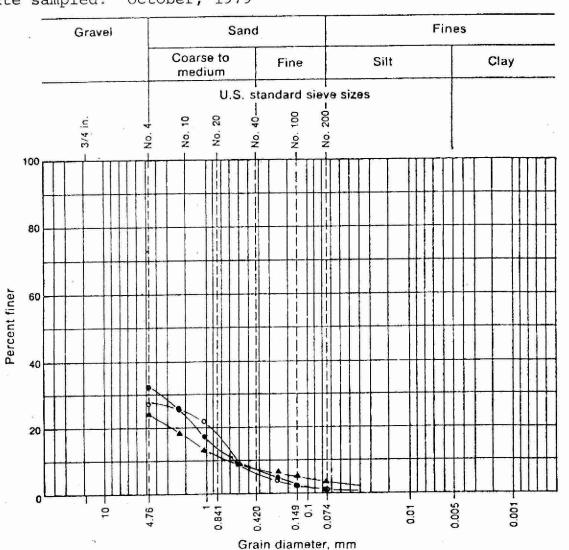
Project Study of sewage lagoon effluent renovation by rapid infiltration

Location of Project Markdale Boring No. Sample No.

Description of Soil Depth of Sample

Tested By. Steve Check Date of Testing January, 1980

Date sampled: October, 1979

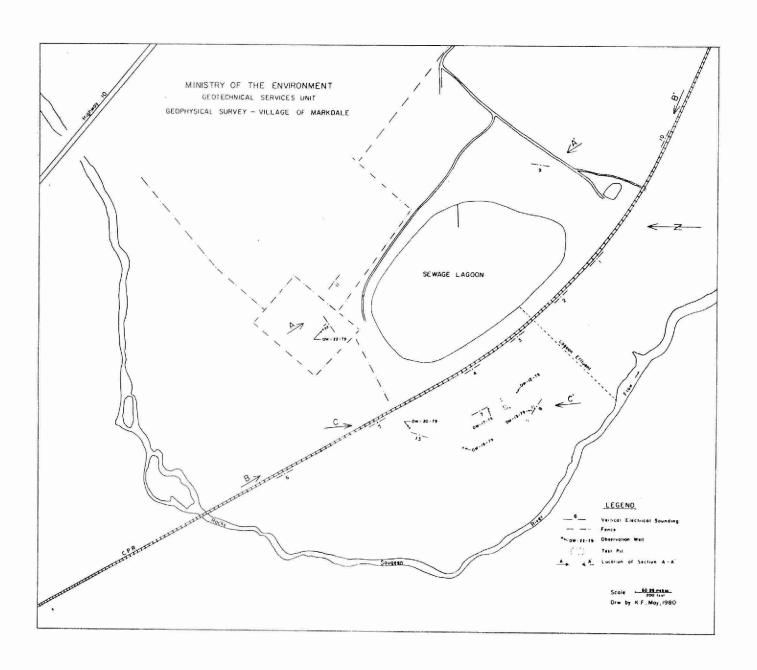


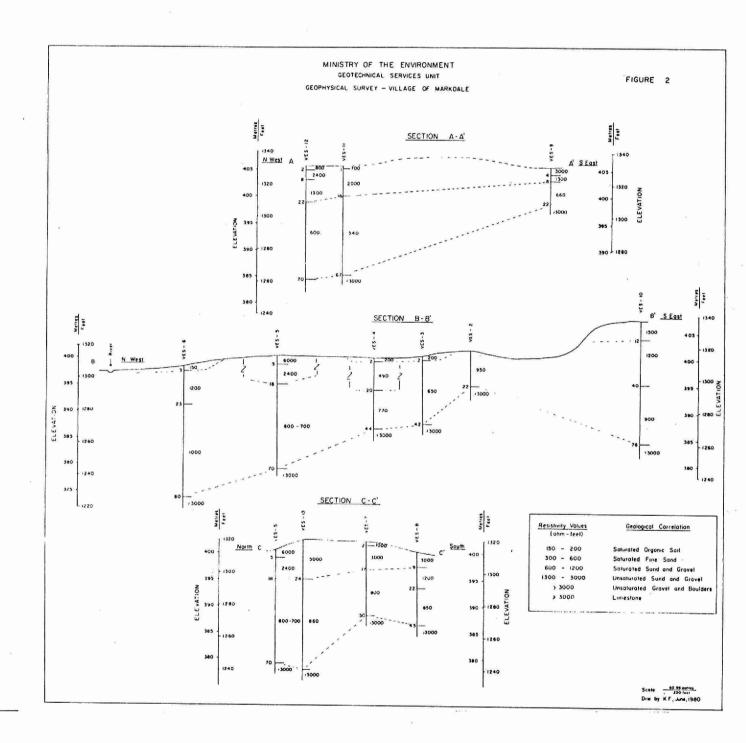
LEGEND

Test pit no.	Depth (m)	Symbol	Soil description
4	0.4		Gravel with coarse sand some fines
4	0.73	-0-	Gravel with coarse sand some fines
4	1.22	-	Gravel with coarse sand some fines

APPENDIX C

RESULTS OF GEOPHYSICAL SURVEY





APPENDIX D

SUMMARY OF WATER WELL RECORDS



SUMMARY OF WATER WELL RECORDS

Southwestern Region

Technical Support Section

985 Adelaide St. South, London N6E 1V3

Compiler: N. RENNIE

County: GREY

Township (s): GLENELG

Date compiled: 19/04/30

County	. 6	KEY				IOW	nship (s): G	LENELG	>				Date c	ompii	ea. /	04/80	Compiler: N. RENNIE
Wei	Lo	cati	on	Elev			Dat drii	ĭ ¥e∃	Length casing,	Well	Depth	Original level, in	Pumpi	ng	test	Kind	Water	
number 1	Township	Lot	Concession	Elevation, in feet	Owner	Driller	Date 5	Well diameter, in inches	Length of casing, in feet	Well depth, in feet	Depth water found, in feet	Original static level, in feet	Drawdown , in feet	Pumping rate, in Igpm	Duration of pumping,hrs	of water ²	er use 3	Well log and remarks
1301	G	90	II Wisr	1390	ALVIN FOSTER	M. BELLERBY	23/ 06/ 51	4	111	138	139	20	10	6.6	1	FR	DO 51	0-21 DUG WELL 21-111 SAND and CLAY 111-138 LIMESTONE ROCK
4325	G	100	II WTS.R	1370	AUGUST STRAUTNIERS	MES WELL DRILLING	21/ 08/ 73	4	30	105	87 102	46	29	4	-	FR	Do	0-4 BROWN FILL 52-105 BROWN 4-21 HARDPAN LIMESTONE 21-29 GRAVEL, STONES 29-52 GREY BROKEN LIMESTONE
4655	6	100	IL WTSR.	1360	ALLAN OLSEN	RAY SPENCER WELL DRILLING	18/06/74	4	63	146	118	25	65	14	2.5	FR	DO	0-2 TOPSOIL 62-146 BROWN 2-17 GRAVEL LIMESTONE 17-26 CLAY, GRAVEL 26-62 GRAVEL, SAND
4658	G	100	II Wisr	1355	GORD HEDGES	Ú,	17/06/74	4	63	90	83	24	16	24	2.75	FR	∞	0-2 TOPSOIL 62-90 BROWN 2-17 GRAVEL LIMESTONE OO 17-26 CLAY, GRAVEL 26-62 BROWN GRAVEL
3125	G		III WT.S.R	1350	R.I MEI KLEHAM	ALLAN	3/11/69	5	67	106	100	60	15	10	6	FR	Do	0-67 CLAY, GRAVEL 67-107 YELLOW LIMESTONE
3349	G	91	I WTSR	1325	D.L. THOMSON	ц	14/10/70	5	82	98	90	-1	-	20	6	FR	Do	0-82 CLAY, GRAVEL 82-48 YELLOW LIMESTONE
1665				1390	E TRAFFORD	m.S. BELLERBY	29/ 04/ ₆₃	5	43	94.5	-	-	-	_	-	-	Со	0-43 STONEY GRAVEL, CLAY 43-945 GREY LIMESTONE
4198		100	I WT.S.R.	1350	J. BROPHY	ALLAN LOUCKS	18/ 06/ 73	5	45	100	95	27	13	10	4	FR	DO	0-45 BROWNCLAY, GRAVEL 45-80 GREY LIMESTONE 80-100 BROWN LIMESTONE
4218		101	II WT.S.R	1370	MARKDALE LEGION .	JIM CLARKE WELL DRILLING	26/ 01/ 73	5	30	111	62	35	55	5	١	R	co	0-5 BROWN CLAY 5-30 CLAY, GRAVEL 30-90 BROWN LIMESTONE 90-111 GREY LIMESTONE

Location is shown in Figure . FR = fresh; SA = salty; SU = sulphur; MN = mineral . 3DO = domestic; ST = stock; IR = irrigation; IN = industry; CO = commercial; MU = municipal;

PS-public supply; CA- cooling or air conditioning . MOE 0488/11/78



SUMMARY OF WATER WELL RECORDS

Southwestern Region

Technical Support Section

985 Adelaide St. South, London N6E 1V3

Compiler: N. RENNIE

County: GREY Township(s): GLENELG

ship(s): GLENELG Date compiled: 19/04/80

County	1:61	KET				TOW	nship (S); G	LENC					Date c	ompine	<u> </u>	~ / 8	Compiler: N. RENNIE
Well number	L Township	catio	n Concession	Elevation, in feet	Owner	Driller	Date Strilled	eter,	Length of casing, in feet	Well depth, in feet	Depth water found, in feet	Original static level, in feet	Pumpi Drawdown,	Pumping rate, in Igpm	buration of pumping,hrs	Kind of water ²	Water use 3	Well log and remarks
4533				10-1	MIN. OF THE ENVIRONMENT	F.E. SOHNSTON DRILLING CO. LTD.	15/8/73	જ	10	240	110	27	45	4∞	24	FR	mυ	0-10 BROWN SAND 10-56 SAND, ROCK, GRAVEL 56-71 GRAVEL, ROCK 71-240 GREY, BROWN LIMESTONE
4534				1365	i,	li	31/1/13	8	19	232	56 115 128 203	17	35	200	24	FR		0-4 BROWN SAND 4-131 WHITE BROWN COLOMITE 131-225 GREY WHITE POLOMITE 225-232 RUST SHALE
4799				1360	MAGEE CARSON	RAY SPENCER WELL DRILLING	24/ 07/ 74	4	54	OP	75 88	14	l)	20	t	FR		0-3 BROWN FILL 64-90 WHITE 3-5 BLACK TOPSOIL LIMESTONE 5-53 GRAYEL 53-64 BROWN LIMESTONE
5872	G	ıω	II Witsr	1350	PETER STRAIN	1)	4/09/16	5	57	146	135	20	55	8	2.25	FR	DO	0-22 GRAVEL 55-146 LIMESTONE 22-29 PED CLAY 29-55 GRAVEL
																		a.
																	ul.	-
																		*

Location is shown in Figure . FR - fresh; SA - salty; SU - sulphur; MN - mineral . 3DO - domestic; ST - stock; IR - irrigation; IN - industry; CO - commercial; MU - municipal;

PS - public supply; CA - cooling or air conditioning . MOF 0488 11/78

APPENDIX E

SUMMARY OF CHEMICAL ANALYSES OF GROUNDWATER



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R.

Date compiled: 15/04/80 Compiler: N. RENNIE

Nun	Owner			tion.	Date	Hydrogen Sulphide	Hard	Alka	Iron	P	Apparent in Haze	Turb	Cond	Flu	Chic	Sulp	Calcium	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number ¹	Owner or Source	Township	Lot	Concession	e Sampled ×	ogen as hide H ₂ S	Hardness caco ₃	Alkalinity as	as Fe	at lab	arent Colour, Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	ا بوبر Phenois,in
* ow1	OBSERVATION WELL NO. 1	2			12/09/79		368	340	0.15	7.9			730		42	н	80	41	20	4.5	0.3	1.4	.02	0.1	.04	,760	
"	п				18/10/19															1.4							1.0
h	N:				п П:04		325	299	0.98	7.52			750		36.0	0.0	7 7.0	32.2	29.5	1.50	0.010	0.62	0.011	4.39	0.007	0.046	1.5
и	и				18:45		344	272	* 40.01	7:37			720		* 39.5	* 30.5	83.5	32.8	* 32.0	* 1.55	4 0.030	* 0.69	* 0.137	* 3.82	* 0.009	* 0.618	102.
11	и				10/19		304	301	0.08	7.52			770		56.5	₩ 45.5	73.0	29.4			≠ 0.355	* 1.51	* 0.83	* 1.97	* 0.013	* 0.0 59	41.0
ır	l ₉						279	269	× 0.03	7.49			760		55.0	* 41.0	64.0	29.0		3 .5	# 0.125	1.43	# 0.030	* 0.54	* 0.027	0.086	41.0
п	a				23/ 10/ 79		279	278	≯ 0.24	7. 54			782		55.0	4 9.0	0.7ما	27.0	51.5	* 7.70	* 4.4	* 4.4	* 0.235	* 0.36	* 0.215	* 0.570	1.5
#	36				24/ /10/ /79		262	276	0.22	7.73			778		55.5	41.0	60.5	26.8	51.5	7.65			0.157	0.46	0.460	0.726	1.5
*	n				28/ /10/ 70		272	184	3.05	7-91			780		56.5	41.0	62.5	28.0	51.5	8.30	5.0	6.30	0.355	0.14	0.340	0.380	1.5

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp. gal; 1 µg/l = 1 ppb.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/I unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CON I, W.T.S.R.

Date compiled: 15/04/80

Compiler : N. RENNIE

	0																					/ 80	- т	7			_			\neg
Ident	Owner or Source	т	1	ion	Date	Bioch	Dema	Chem	С	arbo	n	Petroleum Hydrocarb	Tannins Lignins	Reactive Silicate	Total Solids	Seler	Arsenic	Barium as	Cadmium	Chromium	Copper	Cyan	Lead as Pb	Mang	Nickel	Zinc	Alum			
Identification Number ¹	er or	Township	Lot	Concession	Sampled 3	Biochemical Oxygen Demand (BOD)	Demand (COD)	Chemical Oxygen	Inorganic	Organic	Total	Petroleum Hydrocarbons	ns and	ive te as Si	Total Dissolved Solids	Selenium as Se	nic as As	m as Ba	ium as Cd	nium as Cr	er as Cu	Cyanìde as CN	as Pb	Manganese as Mn	as Ni	as Zn	Aluminum			
					P k																									
οω1	OBSERVATION WELL NO. 1				18/10/19 11:34				73.0	15.0	88.0										1.0		۷.25	2.0		0.50	36			
	н				17:04				74.5	JI.O	85.5																		â	
μ	N				N:45	7.6	5	53																					-	103
н	11				P/ /10/19 /4:32				70.5	23.0	93.5																			
н	И				11				70.5	22.5	93.0																			
11	W				23/ /10/ /79				70.5	29.6	100.0																		*	
*	u.				24/ /10/ /79				6 8.5	29.5	ବ୍ୟ .୦																	-		
*	11				29/10/79				6.Pg	0.ول2	95.0					۷.00	0.002	0.06	4.005	40.02	0.02		<0.03	0.23	<0.02	<.01				

¹ Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1µg/l=1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: COEU

Township(s): GLENELG, LOT 95, CONI WT.S.R Date compiled: 15/04/2. Compiler: N. RENNIE

County	GREY							Iowns	inip(s)	GLEN	JELG,	LOT 4	o, cor	or m	1. S. K			Date	compi	led:	04/80	Con	npiler:	N.RE	NNIE		
lden Num	Owner			tion	Date	Hydrogen Sulphide	Hard	Alka	Iron	РН	Apparent in Haze	Turbidity, Formazin	Cond	Fluoride	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitro		as	N	Phosph as	orus P	Phe
Identification Number ¹	er or ce	Township	Lot	Concession	Sampled M	as H ₂ S	Hardness CaCO3	Alkalinity caco3	as Fe	at lab	arent Colour, Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	oride as F	ride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved	Total	Phenols, in µg/l
	OBSERVATION WELL NO. 1	2			31/ /10/ /79		2106		0.90								67.0	23.6	51-0	8.25		6.00				0.970	3.0
* *	41				3/11/19		272	272	1.12	7.87			760		56.0	39.5	63.5	27.6	52.0	8.10	3.85	6.10	0.178	1.49	1.43	1.67	حا. ٥
*	Ш				9/11/19		266	265	0.13	7.80			770		57.0	42.5	٥. اوا	27.6			3.22	5,30	0.063	248	1.94	2.20	41.0
* "	и				12/11/79		276	274-	0.80	8.07			780	0.23	56.5	52.0	64.0	28.2			3.12	4.60	0.146	0.90	0.710	0.79	۵.۱۷
* "	Ĵτ				16/11/79		264	268	0.14	7.75			760		56.0	39.0	61.5	26.8	54.0	7.70	3.6	0ءا.ما	0.030	2.5	1.98	2.50	41.0
d	п				22/11/19		270	269	8.4	7.50			755		62.0		63.0	27.2	56.0	7. 25	2.7	8.25	0.08	2.4	2.15	3.10	
*	N.				12/ 11/ 19/ 19/ 19/ 29:20		255	262	0.20	7.88			755		6١٠5	36.5	58.0	26.8	57.0	7.65	1.0	4.00	0.145	4.3	1.85	2.05	
*	h				22/ /11/ /79 116:41		258	264	1.32	7.69			760		63.5	36.5	57.0	28.0	56.0	7.75	2.9	4.90	0.03	2.1	1.75	2.00	41.0
*	п				15/ 11/19 19:03		250	258	0.02	7.51			745		56.0	40.5	57.0	26.2	51.0	9.8	3.0	5.00	0,071	1.80	1.05	2.00	

¹ Location is shown in Figure 2; N.D. – Not detected; < – Refers to less than; 1 mg/l = 1 ppm = 1|b|100,000 lmp. gal; 1 µg/l = 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



SUMMARY OF CHEMICAL ANALYSES OF

GROUNDWATER

All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CON I W.T.S.R

Date compiled: 15/04/80

Compiler : N. RENNIE

Num	Soun		\neg	ion	Date	Bioch	Cher	С	arbo	n	Petro Hydro	Tannins Lignins	Reactive Silicate	Total Solids	Sele	Arsenic	Barium	Cadr	Chro	Copper	Cyar	Lead	Mang	Nickel	Zinc			Anionic
Identification Number ¹	Owner or Source	Township	Lot	Concession	Sampled 3	Biochemical Oxygen Demand (BOD ₅)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	ins and	ate as Si	Dissolved	Selenium as Se	nic as As	um as Ba	Cadmium as Cd	Chromium as Cr	per as Cu	Cyanìde as CN	Lead as Pb	Manganese as Mn	e as Ni	as Zn		ā	nic Detergent
	OBSERVATION WELL NO. 1	4			31/ 10/ 79			69.5	20.0	99.5																		0.4
* "	le				3/1/79	4.6	44	0.ماما	25.5	91.5																		
* "	И				%1/ ₇₉			65.0	24.0	93.0																		
																												105
*	u				14/49			68.0	24.5	92.5																×		
*	u				22/ 1/19 09:20			63.0	19.0	82.0																		
*	le				16:41			65.5	27.0	92.5																		0.4
*	И				26/ 11/ 79										∠.∞	.001	.03	<.005	<.02			04	.04	۷.02	4.01			

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1µg/l=1 ppb.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 15/04/6 Compiler: N. RENNIE

County	GKEY							TOWIIS	ilib(a)	· GLEI	TELO,	12019	3, 001	<u> </u>), I. J. K			Date	compi	iea .	/80	Cor	npiler	N. 72			
Identific Number	Sour	-	atio	⊣∌	Sulph	Hydrogen	Hardnes	Alka	Iron	РН	Appa in H	Turbidity, Formazin	Cond	Fluc	Chio	Sulphate	Calcium	Мад	Sodium	Pota	Nitro	ogen	as	N	Phosph as	oru s P	Phe
Identification Number ¹	Owner or Source	Township	Concession	Sampled ×	H ₂ S		ness _{CaCO3}	Alkalinity CaCO3	as Fe	at lab	Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	hate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved	Total	Phenols, in µg/l
* 0w1	OBSERVATION WELL NO: 1			31/	0		313	317	0.24	7.63			640	0.13	20.0	11.5	72.5	32.0	13.2	2.65	0.590	1.03	0.019	0.75	0.13	0.180	Z1.0
			\downarrow	-	\downarrow																						
* ο ω2	OBSERVATION WELL NO. 2	3		12/09/	, , ,		369			7.80			690		420						0. 20	1.40	0.01	40.1	0.02	0.26	
11	1			19/	, q 4		310	298	0.96	7.60			645		32.5	10.0	65.0	35.8	15.6	2.85	0.010	1.26	0.008	0.02	0.001	0.052	2.5
11.	11			17:2	14		322	316	<u></u>	7.39			680		* 32.5	* 21.5	71.0	35.0	* 20.0	* 2.20	* 0.370	* 0.97	* 0.009	* 140	* 0.034	* 0.0 43	1. 5
п	и			10	79		304	301	* 0.08	7.52			770		* 56.5	* 46.5	73.0	29.4			o. 355	¥ 1.51	* 0.83	* 1.97	* 0.013	¥ 0.059	21.0
п	43			24/ 10/1	_		353	294	0.22	7.55			740		34.5	54.5	38.0	32.4	21.10	1.85	0.010	1-17	0.194	0.95	0.001	0.067	2.0
*	ár -			10/	9		354	344	11.2	7.71			740		46.0	39.5	89.0	32.0	33.0	6.90	0.420	1.50	0.470	0.08	0.022	0.330	2.5

Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp.gal; 1 µg/l = 1 ppb. MOE 3495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R Date compiled: 15/4/80 Compiler: N. RENNIE

Carbo Carbo Carbo Chemical Oxygen Si.O 16.9

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1µg/l=1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R.

Date compiled: 15/04/80 Compiler: N. RENNIE

		T-						_											оотпр		′ ./8	6 OO.	- piici				
Identifica Number ¹	Owner Source	-	ati	\neg	Date	Hydrogen Sulphide	Hardness	Alkalinity	Iron	P	Apparent in Haze	Turbidity, Formazin	Condu	Fluoride	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitre	ogen	as	N	Phosph as	orus P	Phe
ation	or	Township	2	Concession	Sampled ×	gen as de H ₂ S	ness caco ₃	linity as	as Fe	at lab	arent Colour, Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	ride as F	ride as CI	hate as SO4	um as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved Reative	Total	Phenois, in µg/l
χ 0ω2	OBSERVATION WELL NO. 2				31/ 19/ 79		313		0.07								71.0	32.6	37.8	3.90		1.40				0.039	2.5
*	10				9/11/79		291	277	0.44	7.8			770		54.5	49.0	67.0	30.0			2.12	3.10	0.290	0.31	0.004	0.081	1.0
*	11				22/ 11/ 179			264	0.80	7.69			810		75.0	40.0				5.10	2-1	3.30	0.15	1.0	0.30	0.36	۷۱.0
*	jı				25/ 11/79		255	264	0.80	7.75			740		57.0	35.0	57.0	27.4	52.0	7.55	2:4	4.00	0.38	1.11	1.00	1.16	,
* "	п				31/ /01/ 80		304	313	0.64	7.71			665	1.28	26.0	17.0	<i>1</i> 4.0	31.8	20.4	4.7	1.25	1.67	0.008	0.19	0.073	0.148	
			-																								
*	OBCCO (DCCA)				12.4																						
Λ 0ω3	OBSERVATION WELL NO. 3			-	12/ /8/ /79		265	238	0.26	7.9			500		20.0	0.وا	54.0	32.0	6.0	4.6	0.3	1.2	.01	40.1	0-02	0.36	
t)	tir				18/ 10/ 79 18:49		284	265	* <0.01	7.62			530		* 17.0	* 20.0	55.0	35.6	* 5.7	* 4.60	* 0.085	* 1.04	<i>⋆</i> o.∞3	* 0.04	* 0.00b	* 0.013	2.5

Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1mg/l= 1ppm = 1lb/100,000 lmp.gal; 1µg/l= 1ppb.

MCE 3495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 15/4/80

Compiler: N. RENNIE

-		_											_	_				Y				7	—	- 1	· · · · ·		т-т	-		7
Ident	Owner or Source	Loc	\neg		Date	Bioch	Chem	С	arbo	n	Petroleum Hydrocarb	Tannins Lignins	Silicate	Reactive	Total Solids	Selenium	Arsenic	Baric	Cadn	Chro	Copper	Cyanide	Lead as Pb	Mang	Nicke	Zinc	Alum			Anionic
Identification	er or ce	Township	2	Concession	Sampled 3	Biochemical Oxygen Demand (BOD _S)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	Tannins and Lignins	ste as Si	ive	Dissolved	nium as Se	nic as As	Barium as Ba	Cadmium as Cd	Chromium as Cr	er as Cu	i)de as CN	as Pb	Manganese as Mn	Nickel as Ni	as Zn	Aluminum		- 1	ic Detergent
)wz	OBSERVATION WELL NO. 2	4			31/ 19/ 79			74.5	13.0	97.5																				
<i>"</i>	t ₁				9/1/19			69.0	19.5	98.5																			0.	A
*	IN.				22/ /1/ /79			ьq.0	21.0	90.0																			0	3
* "	lı				16/ 11/ ₇₉			•								∠.∞1	∠.03	-04	∠.∞5	۷.02			.06	.12	<.02	۷.0۱				1.601
* "	11				31/ /01/ /80			78.5	368	465																				
		\coprod																			<u> </u>									
													-																	_
ow3	OBSERVATION WELL- NO : 3	7			18/49 18:49	23. b	777	63.0	12.0	75.0											1.0		۷.25	9.0		1.5	110			

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1|b|100,000 lmp.gal; 1 µg/l=1 ppb. A - sampled on November 3, 1979. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI WITS R. Date compiled: 15/80 Compiler: N. RENNIE

Num	Owner Source	-	ation	⊣ ≱	Sulphide	Haro	Alka	Iron	РН	Appa	Turbidity, Formazin	Cond	Fluc	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number ¹	er or ce	Township	Concession	Sampled	H ₂ S	(n)	Alkalinity caco3	as Fe	at lab	Apparent Colour, in Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	ride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenols, in µg/
0W3	OBSERVATION WELL NO. 3			24/ /10/ 71	3	298	250	0.16	7.67			580		19.5	36.0	62.5	34.4	6.0	3.80	0.005	0.52	0.001	۷٥.0١	0.002	0.03	3.0
*	11			28/ 10/ 79		300	266	0.04	7.88			580		17.0	25.5	61.5	35.b	4.4	3.40	0.010	0.25	0.004	40.01	6.003	0.003	3.0
*	n			31/19/70		303		0.16								59.5	37.4	5.1	4.20		0.18				0.005	1.0
*	h			9/11/79		30%	279	0.04	7.83			590		5. ما ا	22.0	60.5	38.0			0.020	0.13	0.002	40.01	0.009	0.012	<1.0
*	IJ			22/	1	292	300	0.26	7.79			555		13.5	22.0	58.0	35.6	4.0	2.10	0.10	0.15	20.01	40.1	40.05	<0.05	41.0
*	11			25/ /11/ /7º		298	271	0.08	7.76			560		14.5	20.0	60.0	35.8	4.4	2.55	0.040	0.32	0.∞3	40.01	0.001	0.017	
*	l1			31/ 101/ 180		261	247	0.12	8.00			496	40.10	9.5	17.0	53.5	31.0	3.6	1.85	0.045	○ 43	0.001	0.23	0.003	0.006	دا.٥
							-																			

Location is shown in Figure $2 \pm N.D.$ — Not detected; < — Refers to less than; 1 mg/l = 1 ppm = 1 lb/100,000 lmp. gal; 1 µg/l = 1 ppb.MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG , LOT 95 , CON. I W.T.S.R. Date compiled: 15/04/80

Compiler: N. RENNIE

Location D D B D C Carbon T D C B S D S D C	tion D D B D C Carbon H D C D S B S D S O O	D D B D C Carbon H D C T S B S O O	D B D C Carbon H D L T S B S O O	Carbon T P C T S P S O O	Carbon T P C T S P S O O	arbon H P L T S P S O O	n Hy Sa So Se	Hy Sa	SH Se	S S S S S S S S S S S S S S S S S S S	S 10 S	Se	Γ	Ar	B	Ca	Ç	Co		T	Z a	Z	Zii			A	7
95 S	S			ochei	man		_		trole	gnin	activ	lids	en	sen	riur	d m	rom	ppe	ani	ad a	nga	ckel				onic	
	wnship	ncession	<u> </u>	mical Oxygen d (BOD _S)	cai Oxygen d (COD)	Inorganic	Organic	Total	erbons	s and	e as Si	Disagrado	ium as Se	ic as As	n as Ba	um as Cd	ium as Cr	r as Cu	de as CN	IS PB	nese as Mn	as Ni	as Zn			Detergent	
RVATION - 3			/79			62.5	15.5	78.0																			
tx			28/ 10/ 79			67.0	8.5	75.5					۷.00	0.001	.03	<.005	Z.02	.04		<.03	ط4.	<.02	.04				
1 <u>i</u>			31/ 10/ 79			L&.5	٥.0	74.5																			
h			9/11/79			70.0	11.0	81.0																		40.	, a F
μ			22/ 11/ 79			ld0.5	5.0	71.5																		0.2	2
h			26/ 11/ ₇₉																								
IV.			31/ 101/ 180			60.0	400	460					٥.٥٥	14.001	4.02	<.005	∠.02			:06	.39	۷.02	<.01				
	Owner or Systion 3	Arions 3	hip ssion	Second S	hip ssion oxygen OD (C)	## Pled Oxygen Ox	## Pled op Nygen Converse Nygen Nygen Converse Nygen Nygen Converse Nygen	No No No No No No No No				Pole Pole		Owner or Chemical Oxygen Demand (COD) Demand (BOD) Deman	Ni Ni Ni Ni Ni Ni Ni Ni	Docation Concession Command (Cod) Cod Cod	Demand (COD) Biochemical Oxygen Demand (BOD) Demand (BO	Owner or Carbon Car	Location Total Dissolved Selenium as 8a Carbon Total Dissolved Solids Solid	Copper as Cu Copp	Location Cyanide as CV C		Nickel as Nick	Nickel as Ni	Location Carbon Carbon Carbon Cyanide as Cu Chromium as Cu C	Location Carbon Location Location Carbon Location Location	

1 Location is shown in Figure 1; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 µg/l=1 ppb. A - sampled on November 3, 1979.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONC. I W.T.S.R. Date compiled: 15/04/80 Compiler: N. RENNIE

	· GRET									0			15, ~	3100.1	0.	112.15					7 700						gran lane and the lane
Nun	Owner Source		-	ion.	Date	Hydrogen Sulphide	Harc	Alka	Iron	Н	Apparent in Haze	Turbi	Cond	Fluc	Chic	Sulp	Calcium	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number ¹	er or	Township	Lot	Concession	Sampled ×	-	Hardness caco ₃	Alkalinity CaCO3	as Fe	at lab	arent Colour, Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenols,in µg//
¥ 0₩4	OBSERVATION WELL NO. 4	3			12/ /09/ /79		374			8.1			710		72.0						0.2	2.2	0.01	20.1	20.02	0.58	
h	ìı				19/10/19		478	287	* 0.02	7.58			755		* 59.0	* 25.5	89.0	62.0	* 11.2	* 1.5	* 0.0 1 0	* 1.70	* 0.007	* 0.04	* 0.006	* 0.035	27.5
п	TI				24/ /b/ /79		373	294	0.20	7.68			735		43.0	41.0	81.5	41.2	10.1	1.65	0,115	1.12	ه∞.٥	اه.ه>	0.001	0.037	12.5
*	10				28/ 10/ 79		383	295	0.07	7.86			755		42.5	51.5	85.5	41.2	9.5	1.90	0.175	1.24	0.006	0.01	0.002	0.012	18.5
*	tr				31/10/79		369		0.24								78.5	42.0	8.9	0 عا - ا		0.64				0.016	/2.0
* ,	to				9/11/ ₇₉		372	297	0.13	7.83			720		34.0	0.10	83.0	40.0			0.205	0.95	0.002	<0.0I	0.004	0.013	1.0
*	И.				12/11/19		375	296	0.06	7.90			720	0.10	32.0	54.0	84.0	40.0			0.185	1.37	0.004	0.04	0.004	0.013	7.0
*	ri .				22/11/19		335	279	0.59	7.83			640		28.5	17.0	69.0	39.4	3.8	2.1	0.1	0.60	20.01	20.1	40.05	40.05	2.5
X	ć,				25/11/79		330	296	0.01	7.81			645		25.5	21.0	69.5	38.0	9.8	1.50	0.110	0.51	0.002	۷٥.0۱	0.001	ماحون	

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1lb/100,000 lmp. gal; 1μg/l = 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



GROUNDWATER SUMMARY OF CHEMICAL ANALYSES OF

All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONCI W.T.S.R. Date compiled: 15/04/80

Compiler: N. RENNIE

Iden Num	Owner	Loc	1	-	Date	Bioch	Chen	С	arbo	n	Petro Hydro	Tannins Lignins	Reactive Silicate	Total Solids	Sele	Arsenic	Barium	Cadr	Chro	Copper	Cyanide	Lead	Mans	Nickel	Zinc		Anion
Identification Number ¹	Owner or Source	Township		Concession	Sampled *	Biochemical Oxygen Demand (BOD)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	Tannins and Lignins	tive ate as Si	Dissolved	Selenium as Se	nic as As	um as Ba	Cadmium as Cd	Chromium as Cr	per as Cu	nide as CN	Lead as Pb	Manganese as Mn	el as Ni	as Zn		Anionic Detergent
	OBSERVATION WELL NO: 4				19/ /10/ /79	-51	150	74.5	31.5	106																	
н	II				24/ /10/ /79			73.0	23.0	96.0																	
*	u				10/ 10/ 79			74.0	23.5	97.5					١٥٥١, ٢	0.007	.04	<.005	۷.02	.04		<.03	.20	۷.02	.06		.113
*	и				31/ 10/ 79			78.0	12.5	90.5																	
⊁ π	n				9/11/79			75.0	14.0	99.0																	(O.1
*	w				12/19			76.0	15.0	91.0																	03
h	1,			$\overline{}$	27/19 17/19			75.0	18.5	93.5																	
* II	ų				26/1/79										ا‱.	.00%	.03	∠.∞5	2.02			.06	0.18	∠.02	0.18		

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 µg/l=1 ppb. A - sampled on November 3, 1919. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG LOT 95, CON. I W.T.S.R. Date compiled: 15/04/ Compiler: N. RENNIE

Name and Address of the Owner, where the Owner, which is the Owner, where the Owner, which is) GIVE T	-		alle produces assessment	_	y-	-	y mp(o)	VLL	NAME OF TAXABLE PARTY.	, 101	45, C	7N. T	W.T.	S.K.	W THE COLUMN TWO IS NOT THE COLUMN TWO IS NO	Date	comp	neu .	104/8	COL	npner	: N. KI	ENNIE		
Identifica Number 1	Owner Source	-	ation	Date	Hydrogen Sulphide	Hardness	Alka	Iron	H	Apparent (in Hazen	Turbidity, Formazin	Condu	Fluoride	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number 1	or or	Township	Concession	Sampled M	as H ₂ S	ness caco ₃	Alkalinity caco3	as Fe	at lab	rent Colour, lazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	ride as F	ride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved	Total	Phenois, in µg/1
* 0w4	OBSERVATION WELL NO. 4			31/01/80		334	293	0.15	8.09			642	40.10	20.0	39.0	75.0	35.6	8.2	1.30	0.095	0.76	0.002	0.28	0.000	0.012	3.0
X 0w5	OBSERVATION WELL- NO. 5			19/19		700	259	0.54	7.7			1410		128	340	160.0	73.0	47.0	5. 4	0.4	1.8	0.32	0.98	40.02	0.22	
lı	Iţ			19/19/79		437	291	* 0.02	7.66			900		* 91.0	* 52.0	76.0	60.0	* 24.6	* 2.30	* 0.230	* 0.79	* 0.011	头 0.11	* 0.001	* 0.004	5.5
н	tr			24/ /10/ /79		349	286	0.24	7.56			730		48.0	35.0	82.5	34.6	24.6	1.45	0.030	0.72	0.004	0.31	0.001	0.046	41.0
*	l)			28/ 10/ 79		33%	290	0.05	7.71			760		54.0	36.0	74.0	37. 2	33.0	1.80	0.025	0.84	0.141	0.05	0.001	0.016	3.5
* 1	I _I .			31/10/79		334		0.18								77.0	34.4	41.5	2.20		0.80				0.035	1.0
*	1,			9/11/49		280	283	0.07	7.74			758		55.5	37.5	3.5ما	29.4			1.42	2.70	0.023	0.02	0.003	0.044	1.0

¹ Location is shown in Figure λ ; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp.gal; 1 pg/l = 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG , LOT 95, CON. I W.T.S.R. Date compiled: 15/04/80 Compiler: N.RENNIE

Ident Numl	Owner or Source	Loca	T	Date	Bioch	Chem	С	arbo	n	Petroleum Hydrocarb	Tannins Lignins	Reactive	Solids	Sele	Arsenic	Barium	Cadn	Chro	Copper	Cyanide	Lead	Mang	Nick	Zinc	Alun		Anionic
Identification Number ¹	er or	Township	Concession	Sampled *	Biochemical Oxygen Demand (BOD ₅)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	Tannins and Lignins	ate as Si	Solids	Selenium as Se	nic as As	ım as Ba	Cadmium as Cd	Chromium as Cr	er as Cu	i)de as CN	Lead as Pb	Manganese as Mn	Nickel as Ni	as Zn	Aluminum		ic Detergent
	OBSERVATION WELL NO. 4			31,			72.5	228	3∞																		
													-														
ഡട	OBSERVATION WELL NO. 5																										717
tr	VI			19/ /10/ /79		56	72.5	13.0	85.5										1.0		<.25	3.0		1.0	95		
ħ	n			24/ 19/ 79			73.5	14.0	97.5																		
* "	н			28/ 10/ 70			730	17.0	90.0					د.00	0.001	-04	<.005	۷.02	ماه		۷.03	.16	۷.02	-08			
*	h			31/ 19/70			73.5	/4.0	87.5																		
*	ŋ			9/1/49			74.0	12.5	86.5																		O.1 ^A

Location is shown in Figure 2; N.D.—Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1|b|100,000 lmp.gal; 1 µg/l=1 ppb. A - sampled on November 3, 1979.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CON I W.T.S.R. Date compiled . 16/04/ Compiler N RENNIE

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Identifica Number 1	Owner Source			tion.	Date	Hydrogen Sulphide	Hardness	Alkalinity	Iron	PH	Appar in H	Turbidity, Formazin	Condu	Fluoride	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitr	ogen	as	N	Phosph as	orus P	Phe
Identification Number ¹	e or	Township	Lot	Concession	Sampled ×	as H ₂ S	ness CaCO3	linity as	as Fe	at lab	Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	wride as F	ride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenois,in µg/
* 0w5	OBSERVATION WELL NO. 5	2			12/1/19		291	285	0.06	7.81			770	40.10	56.5	40.0	66.5	30.4			1.58	3.15	0.020	0.02	0.002	0.037	1.5
* 1:	b				22/ 11/19		285	272	0.62	7.82			770		59.0	37.0	62.0	31.6	50.0	2.65	2.4	3.20	0.12	1.7	<0.05	0.05	<1.0
* 11	lı				25/ /11/ /79		287	278	0.16	7.77			765		59.5	36.0	60.0	33.2	49.5	4.50	2.2	3.25	0.58	1.00	0.004	0.029	
*	П				31/		282	290	0.18	7.87			000	∠o.10	33.0	18.0	60.0	32.0	28.8	2.90	0.825	1.40	0.078	1.75	0.003	0.010	<1.0
* 0W6	OBSERVATION WELL NO.6				12/9/79		406			7.8			810		84.0						0.2	0.4	0.01	۷٥.١	40.02	0.08	
п	11				18/ 19/ 79 11:34		296	296	0.80	7.49			585		17.5	10.0	70.0	29.4	(1.8	1.35	0.015	0.98	0.064	0.06	(0.001	0.042	2.0
n 1	ţı.				18:12		373	320	* <0.01	7-35			715		* 25.5	* 32.0	91.5	35.0	* 17.0	* 1.45	* 0.030	.≭ ⊙.30	* 0.018	* 0 · 31	* 0.004	* •.010	1.5

¹Location is shown in Figure λ ; N.D. — Not detected; < — Refers to less than; 1 mg/l = 1 ppm = 1/b/100,000 lmp. gal; 1 µg/l = 1 ppb. MOE 3495 :1/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG, LOT 95, CONT W.T.S.R. Date compiled: 16/04/80

Compiler: N. RENNIE

N d	S O S	Lo	cat	ion	Date	D 80	0	Ç	С	arbo	n	Hyo	۳	릴	SI	Solids	Se	Ars	Ва	Ca	Chi	Col	Су	Lea	X a	Nic	Zinc	Alu		Ani
Identifica Number ¹	Owner or Source	70	Lot	ို		man	3	Chemical	_ 1			droc	Lignins	Tannins	Reactive			Arsenic	Barium as	mi	Chromium	Copper	Cyanide	Lead as Pb	ngai	Nickel	C as	Aluminum	-	Anionic
Identification Number ¹	or	Township	t	Concession	Sampled 3	Biochemical Oxygen Demand (BOD ₅)	Demand (COD)	al Oxygen	Inorganic	Organic	Total	Petroleum Hydrocarbons		and	e as Si	Dissolved	um as Se	C as As	n as Ba	Cadmium as Cd	ium as Cr	r as Cu	le as CN	s Pb	Manganese as Mn	as Ni	s Zn	m		Detergent
* თან	OBSERVATIO WELL NO. 5	7			12/1/19				71.5	16.0	87.5																			
* "	VA				22/ 11/ 79				70.5	15.0	9 5.5																		· ·	0.3
*	н				26/ /11/ 79																									
* II	11				31/01/80				68.5	222	290						20	2.00	1 <.07	4.005	۷.02			وا0.	.20	4.02	4.01			11/7
Om-e	DENEZVATION				12/9/19				-																					
ow6	OBSERVATION WELL NO. 6	90			19/ /10/ /79 11:34	-			85.0	9.0	94.0											1.5		۷.25	3.5		1.0	110		
'n	n.				18:12				88.0	6.0	94.0																			

Location is shown in Figure 2; N.D. — Not Detected; P — Present; < — Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 μg/l=1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95. CONI W.T.S.R. Date compiled: 1%44 Compiler: N.RENNIE

County	GKEY		_					TOWNS	snip(s)	, OL	ENELO	,	, د۳	CONIT	ω.τ.	3. K.		Date	compi	iea .	104/8	。 Con	npiler	, N.A.	ENNIE		
Num	Owner Source	-		ion.	Date	Hydrogen Sulphide	Haro	Alka	Iron	рн	Apparent in Haze	Turbi	Cond	Fluc	СИС	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number ¹	er or ce	Township	DO T	Concession	Sampled S	ogen as nide H ₂ S	Hardness CaCO ₃	Alkalinity caco3	as Fe	at lab	arent Colour, Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	hate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenols, in µg/l
owb	OBSERVATION WELL NO.6				19/ 10/ 79		381	350	* 0.03	7.40			690		* 26.0	* 19.0	83.0	42.0			* <0.005	* 0.260	* 0.018	* 0.51	* 0.010	* 0.012	۷۱.0
н	и				24/ 10/ 79		323	275	0.36	7.44			740		50.6	43.0	81.5	29.0	39.5	1.20	CO.005	1.06	0.107	0.70	<0.00 l	0.047	1.0
*	s.f.				31/10/19		284		0.24								66.0	28.8	47.5	2.50		1,21			 	0.033	<1.0
*	IJ				9/1/19		268	271	0.03	7.72			740		56.0	35.0	0.0	28.0			1.35	2.15	0.500	1.70	0.002	0.043	۷1.0
*	η,				16/11/79		284	275	0.24	7.77			762		54.5	36.0	67.0	28. 2	50.0	3.40	1-70	2.90	0.29	2.3	0.004	0.053	41.0
oel Halil	ţ,				22/ 11/79 9:20		329	273	38.0	7. 45			778		66.5		104	32.0	53.0	6.00	2.0	2.50	0.24	1.8	0.20	0.70	
*	11				27/19 9:20		278	273	0.30	7.99			790		67.0	34.5	63.0	29.2	55.5	4.75	2.2	2.95	0.22	2.0	0.00/	0.053	
*	N				n 14:19		275	275	0.60	7.70			799		68.5	37.5	61.0	29.8	41.8	1.95	2.4	3.70	0.03	2.2	40.05	0.08	<1.0
* n	Ų				25/11/79		257	267	0.52	7.75			900		73.5	35.5	58.5	27.0	64.5	4.25	0.60	3.18	0.37	1.08	0.006	0.066	

Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1mg/l = 1ppm = 1lb/100,000 lmp. gal; 1μg/l = 1ppb.

MOE 3495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG, LOT 95, CON. I W.T.S.R. Date compiled: 16/04/80

Compiler: N. RENNIE

													1	T		1				_						N. 1		
dent	≥ ₹	-	-	ion	Date	Bioch	Chem	C	arbo	n	Petroleum Hydrocarb	Tannins Lignins	Silicate	Reactive	Total Solids	Seler	Arsenic	Barium	Cadm	Chromium	Copper	Cyanide	Lead as Pb	Mang	Nickel	Zinc		Anionic
Identification Number ¹	er or	Township	ot	Concession	Sampled 3	Biochemical Oxygen Demand (BOD ₅)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	ns and		₩ (P)	Dissolved	Selenium as Se	nic as As	ım as Ba	Cadmium as Cd	nium as Cr	er as Cu	ide as CN	as Pb	Manganese as Mn	as Ni	as Zn		c Detergent
0W6	OBSERVATION WELL NO: 6				19/10/19			85.5	6.5	92.0																		
h	lą				24/ 19/ 19			ס.סל	19.0	89.0																		
*	н				31/ 10/19			70.5	14.5	% 5.0																		
*	11				9/11/79			64.0	14.5	83.5																		0.1
*	11				16/ 11/79			71.0	17.0	88.0																		
X , "	lt				22/ 11/19 9:10			67.5	16.5	94.0	(
* "	1/				II 14:29			<i>6</i> А.5	19.0	97.5																		0.3
*	Į,				26/ 11/79											۷.03	.007	- 26	4.005	2.02			4.03	.90	4.02	۷.0۱		

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 µg/l=1 ppb. A-sampled on November 3,1979.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 16/04/6 Compiler: N. RENNIE

County	, GIVE Y						TOWITS	ilih(a)	· OLE	MELG	, 100.1	75,0	010 1	۵. ۱			Date	comp	ieu .	1 80	Con	ipiiei .	10.10	01		
Identific Number	Owner Source		ation	⊣ ≇	Hydrogen Sulphide	Hard	Alka	Iron	РН	Appa	Turbidity Formazin	Cond	Fluc	СНО	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number ¹	Owner or Source	Township	Concession	Sampled S	as H ₂ S	Hardness CaCO3	Alkalinity caco3	as Fe	at lab	Apparent Colour, in Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved Reative	Total	Phenois, in µg/ I
* 0₩6	OBSERVATION WELL NO. 6			31/		326	316	0.26	7.77			640	۷٥. ۱٥	21.5	13.0	76.0	33.0	13.7	2.6	0-515	0.94	0.005	0.28	0.003	0-015	<1.0
				-	_																					
* 0w7	OBSERVATION WELL NO: 7			12/09/19		388	349	0.46	7.7			790		64.0	13.0	77.0	48.0	23.0	3.5	0.2	0.4	0.01	(۵۰۱	<0.02	0.20	
21	ñ.			18/10/		343	350	¥ <0.01	7.39			660		* 24.0	* 9.0	82.0	33.6	¥ 15. 2	* 1.95	x 0.010	0.18	* 0.001	* 40.01	* 0.001	* 0.001	۷۱.0
a	Is.			24/ /19/ /79		319	284	2.40	7.43			740		53.0	35.5	% 0.0	29.0	42.5	2.40	0.045	1.23	0.010	1.00	0.018	0.159	<1.0
*	ц			31/10/		277	(a)	0.18								64.5	28.2	48.0	4.50		3.05				0.053	<1.0
*	21			9/11/70		264	275	0.03	7.75			760		56.0	34.0	59.0	28.2			2.88	4.25	0.129	2.07	0.004	0.055	۷۱.0
*,	11			11/1/		281	282	0.08	7.78			782		54.5	35.0	٥.5 وا	28.8	52.0	5.80	3.0	4.55	0.070	3.8	0.003	0.077	41.0

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp.gal; 1 µg/l = 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 5, CONT W.T.S.R. Date compiled: 16/04/80 Compiler: N. RENNIE

The	Anionic Detergent Aluminum Zinc as Zn Nickel as Ni Nanganese as Mn			121.	. 25 /2.0 5.0 290			0.14
1	Pb as C				1.0 4.25			
September Sept	Chromium as Cr							
No. 7 13.0 13.0 13.0 17.5 16.5 18.0 17.5 16.5 18.0 18.0 18.0 17.5 16.5 18.0 17.5 16.5 18.0	as As							
18 18 18 18 18 18 18 18	Total Dissolved Solids							
OBSERVATION WELL-NO.7 H OBSERVATION WELL-NO.7 H OBSERVATION WELL-NO.7 A OBSERVATION WELL-N	Reactive Silicate as Si							
Cester/ation Cest	Tannins and Lignins							
OBSERVATION WELL NO. 7 H ORSERVATION NO. 7 IS / /10/ /79 II / /10/ /79	Hydrocarbons Total	490			91.0	99.5	98.0	39.0
OBSERVATION WELL NO. 7 II II II II II II II II II		90.5 400			86.0 5.0	72.0 17.5	71.5 16.5	
OBSERVATION WELL NO. 6 OBSERVATION WELL NO. 7 H 13/ 10/ 79	Chemical Oxygen Demand (COD) Biochemical Oxygen Demand (BOD)				13.0 63.0			
OBSERVATION WELL NO. 6 OBSERVATION WELL NO. 7	Sampled	_			19/19/79		31/ 10/ 79	
OBSERVATION WELL NO. 6 OBSERVATION WELL NO. 7	Concession							
OBSERVATION WELL NO. 6 OBSERVATION WELL NO. 7		٥			2			\top
1	Owner or Source	WELL			WELL	н	u	

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1,µg/l=1 ppb. A - sampled on November 3, 1979.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Townshin(s): GLENELS, LOT 95, CON. I W. T.S.R. Date compiled: 16/04/ Compiler: A) PENNIE

	. GREY				,		Towns	ship(s)	. GLE	, , , , , , , , , , , , , , , , , , ,	, 20		ωn	- ω.	T.S.K	*	Date	comp	iled :	104/9	o Cor	npiler	: N.K	ENNIE		
Identific Number	Owner Source	Loca	_	ન #	Hydrogen Sulphide	Hard	Alka	Iron	PH	Apparent in Haze	Turbidity, Formazin	Cond	Fluc	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number 1	er or ce	Lot Township	Concession	Sampled *	as H ₂ S	Hardness caco3	Alkalinity caco3	as Fe	at lab	arent Colour, Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	oride as CI	hate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved	Total	Phenols, in µg/1
0W7	OBSERVATION WELL NO: 7			22/ 11/ 79		278	281	0.23	7.64			765		59.5	36.0	61.0	30.4	51-0	4.75	2.2	3.35	0.10	1.7	20.05	۷0.0 5	۷/۰
*	υ			11/79		257	266	0.84	7.81			755		62.0	33.5	57.5	27.4	56.0	5.45	2-4	3.24	0.076	1.39	0.015	0.058	
N.	11			31/01/80		328	319	0.45	7.85			661	۷٥,10	26.5	11.5	76.0	33.6	15. 2	2.10	0.310	0.70	0.015	0.11	0.007	0.014	<1.0
					_																					
* 0w8	OBSERVATION) WELL			12/09/	-	101																				
	NO . 9			/19 18/ 10/ 11/79		329	320	1.32	7.9			690		135 32.0	7.0	v . €.0	38.6	13.0	1.45	0.10	0.46	0.01		0.01	0.29	5.0
и	b			18/ 19/ 19/ 17:51		347	296	* <0:01	7.48			680		* 31.5	* 10.0		38.0	* 8.3	* 1.55	* 0.130	*	*	* 0.16	* 0.010	×- 0.019	
12	и			19/19/19	_	399	318	* 40.01	7.56			715		* 2 9.5	* 34.0	77.0	50.0	*	* /-25	# 0.005	* 0.220	* 0.003	* 0.07	* 0.011	# 0.017	2,0

¹ Location is shown in Figure $\hat{\mathcal{L}}$; N.D. — Not detected ; < — Refers to less than ; 1 mg/l = 1 ppm = 1|b|100,000 | lmp. gal ; 1 \(\mu g/l = 1 \) ppb.

[&]quot;(E 3495)1/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELS , LOT 95 , CON. I W.T.S.R Date compiled: 16/4/80 Compiler: N. RENNIE

Num	Sou			tion	Date	Bioch	Cher	С	arbo	n	Petro Hydr	Tannins Lignins	Reactive	Total Solids	Sele	Arsenic	Barium	Cadı	Chro	Copper	Cyanide	Leac	Man	Nickel	Zinc	Alui		Anionic
Identification Number ¹	Owner or Source	Township	Lot	Concession	Sampled 3	Biochemical Oxygen Demand (BOD ₅)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	ins and ins	ate as Si	Dissolved	Selenium as Se	nic as As	um as Ba	Cadmium as Cd	Chromium as Cr	per as Cu	nìde as CN	Lead as Pb	Manganese as Mn	el as Ni	as Zn	Aluminum		nic Detergent
* ∞7	OBSERVATION WELL NO. 7	2			2/1/19			71.0	17.0	88.0																		0.3
*	n				26/ 11/79										4.001	.00%	.02	∠.005	<02			.07	. 10	Z.02	٥.0١			
h	II				31/ /01/ /80			જી!.0	19.0	100																		
																												. 143
ows	OBSERVATION WELL NO.8	7			18/ 10/ 779,	11:34		79.5	8.0	86.5										1.5		۷.25	18.0		7.5	320		
υ	19				II:61			83.5	3.5	87.0																		
il	ų				19/19/19			85.0	16.0	/01																		

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 pg/l=1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG , LOT 95 , CONT. W.T.S.R. Date compiled: 164 Compiler: N. RENNIE

	· GICET								iiib(e)	0.0		1			ω. ι				comp.		780	00.	ipiiei				
Iden	Owner Source	Loc	T	-	Date	Hydrogen Sulphide	Harc	Alka	Iron	рн	Apparent in Haze	Turbi	Cond	Fluc	Chic	Sulp	Calcium	Мад	Sodium	Pota	Nitro	ogen	as	N	Phosph as	or us P	Phe
Identification Number ¹	Owner or Source	Township	6 4	Concession	~ I	ogen as nide H ₂ S	Hardness caco ₃	Alkalinity CaCO3	as Fe	at lab	arent Colour, Hazen Units	Turbidity, in Formazin Units	Conductance, in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved	Total	Phenois,in µg/
0W8	OBSERVATION WELL NO. 8				24/ 10/ 19		382	329	0.44	7.58			720		29.5	27.5	91.0	37 ₋ 6	12.2	7.40	0.005	0.48	0.001	ره.0۱	0.001	0.019	2.0
*	ų				31/ 10/79		377		0.05								86.0	39.4	20.6	1.60		0.30				0.009	۷.0
*	и				9/11/79		337	308	0.02	7.80			780		48.5	40.0	76.0	ما 35			0.015	0.51	0.004	0.02	<0.001	0.008	1.0
*	И				26) 11/ 19				.07																		
*	и				31/ 01/ 80		316	3∞	0.26	7.94			675	40.10	35.5	17.5	71.0	33.6	21.4	1.90	0.630	1.14	0.010	0.16	0.005	0.043	L1,0
* owa	OBSERVATION WELL NO. 9	7			12/ 09/ 79		487	292	0.70	7 .9			1040		165.0	15.0	100.0	59.0	٥.مل2	3.3	<0-1	1.4	< 0.01	۷٥.۱	<0.02	0.46	
ži.	u				19/10/79		362	275	* 0.07	7.67			690		* 89.0	* 14.0	69.0	46.0	19.6	* 2.30	# 0.160	# 0.90	¥ 0.005	* 0.01	* 0.002	\$ 00.008	11.5
4	11				24/ 10/ 79		335	282	0.22	7.66			699		44.5	27.0	77.5	34.4	14.0	1-60	0.015	0.33	డు.∞౹	ZO.01	0.001	0.013	2.0

¹Location is shown in Figure λ ; N.D. – Not detected; < – Refers to less than; 1mg/l = 1ppm = 1lb/100,000 lmp. gal; 1µg/l = 1ppb.

MCE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R.

Date compiled: 16/04/80

Compiler: N. RENNIE

Num	Owner	-	_	ion	Date	Bioch	Dema	Chemical	С	arbo	n	Petro Hydro	Tannins Lignins	Silicate	Reactive	Total Solids	Sele	Arsenic	Bari	Cadr	Chro	Copper	Cyanide	Lead	Mang	Nickel	Zinc	Alum		Anionic
Identification Number ¹	Owner or Source	Township	ot	Concession	Sampled 3	Biochemical Oxygen Demand (BOD)	Demand (COD)	nical Oxygen	Inorganic	Organic	Total	Petroleum Hydrocarbons	ins and	ate as Si	tive	Dissolved	Selenium as Se	nic as As	Barium as Ba	Cadmium as Cd	Chromium as Cr)er as Cu	n) de as CN	Lead as Pb	Manganese as Mn	el as Ni	as Zn	Aluminum		ic Detergent
ഠധമ	OBSERVATION WELL NO. 8				24/ 10/ 79				34 .5	17.5	102									-										
X	ц				31/ 10/ 79				85 .0	11.0	96.0																			
*	lı				9/1/9				78.5	8.0	86.5																		ļ	40. I
X 11	11			_	26												< .00I	<.∞1	.03	4.005	<.02			0.07	.86	۷.02	١٥.٧			
μ	n				31/ 01/ 80				75.0	25.0	100																			
0W9	OBSERVATION WELL NO. 9																												Y	
ú	tı				19/19	49	10	٥4	69.0	36.0	105											1.0		۷.25	5.5		1.0	70		
,	a a				10/79				71.5	15.5	87.0																			li .

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1,µg/l=1 ppb. A - sampled on November 3, 1919.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 704/80 Compiler: N. RENNIE

	y. ORL 1	-			_	-		104411	silih(a)	1.022	10000	, 101	,	CO,-3	L CE.			Date	comp	ilea ;	104/80	Cor	npiler	: N.KE	Dioic		
Identifica Number ¹	Owner	-		tion.	Date	Hydrogen Sulphide	Hard	Alka	Iron	PH	Appa in F	Turbidity, Formazin	Cond	Fluoride	Chloride	Sulphate	Calcium	Mag	Sod	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
ation	or or	Township	.ot	Concession	Sampled ×	gen as ide H ₂ S	Hardness caco ₃	Alkalinity CaCO3	as Fe	at lab	Apparent Colour, in Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	oride as F	ride as CI	hate as SO4	ium as Ca	Magnesium as Mg	Sodium as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved	Total	Phenois, in µg/1
× owq	OBSERVATION WELL NO. 9				31/ 10/ 79		347		0.08								80.5	35.4	14.3	1.75		0.46				0.010	3.5
*	ħ				9/11/19		326	296	0.03	7.68			705		38.0	31.0	77.5	32.0			0.010	0.76	0.002	40.01	<0.001	0.083	1.5
*	н				22/ 11/ 79		314	283	0.27	7.64			720		49.0	34.0	73.0	31.8	29.6	1.30	0.100	0.65	ZO.01	0.10	40.05	<0.05	1.0
*	11				26/11/79				.04																		
* 11	Ц				31/ /01/ /80		327	307	0.29	7.92			680	40.10	37.0	16.5	78.0	32.0	20.0	1.85	0.100	0.72	0.002	0.19	0.005	0.017	<1,0
*	OBSERVATION WELL				12/																						
0W 10	NO. 10	\perp	-		19		375		-	7.9	-		700		33.0						0.30	0.6	۷٥٠٥١	40-1	40.02	0.18	
и	15				18/ 10/ 79 11:34		344	338	2.08	7.36			705		24.5	4.0	79.0	35.6	13.1	0.75	0.050	0.36	0.007	0.02	0.014	0.070	<1.0
B	h				II 17:58		385	367	* 0.04	7-34			705		* 23.5	* 12.0	96.0	35.2	* 8.3	1.00	* 0.090	× 0.36	* 0.005	* 0.04	* 0.006	* 0.009	۷۱.0

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1mg/l = 1ppm = 1|b/100,000 lmp.gal; 1µg/l = 1ppb. MOE 3495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONT. W.T.S.R. Date compiled: 17/04/80

d: 17/04/80 Compiler: N. RENNIE

Identifica Number ¹	Sour	Loca		Date	Bioch Demi	Chen	С	arbo	n	Petroleum Hydrocarb	Tannins Lignins	Reactive	Total Solids	Sele	Arsenic	Barium	Cadr	Chro	Copper	Cyanide	Lead	Mans	Nickel	Zinc	Alun			Anionic
Identification Number ¹	Owner or Source	Lot Township	Concession	Sampled M	Biochemical Oxygen Demand (BOD ₃)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	ns and ns	tive ate as Si	Total Dissolved Solids	Selenium as Se	nic as As	um as Ba	Cadmium as cd	Chromium as Cr	er as Cu	i)de as CN	Lead as Pb	Manganese as Mn	el as Ni	as Zn	Aluminum			iic Detergent
* ow9	OBSERVATION WELL NO. 9			31/10/			77.0	11.5	98.5															,				
*	Įt			9/11/79			77.0	12.0	89.0																		((0.1
* "	is .			22/ 11/ 79			73.5	48.5	122																		4	(٥.١
* "	u			26/14																								1.77
¥ 11	11			31/			78.5	606	685					۱۵۵. ک	ا∞،	.06	<.005	<.02			.05	f, j	4.02	2.01				
0010	OBSERVATION WELL NO. 10	7																								-		
и	W			19/10/10/11:34			91.5	9.5	/01									ç	1.0		۷.25	7.5		1.5	55	¥ .x		
п	h			" 17: 55			98.0	5.0	103																			

Location is shown in Figure 2; N.D.— Not Detected; P — Present; < — Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 μg/l=1 ppb. A - sampled on November 3, 1979.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CON I W.T.S.R Date compiled: 18/4/_ Compiler: N. RENNIE

County	. GREY	Marian	-					TOWITS	uih(a)	· OLL	1-220	, 131	10 ,			. 1. 5.1	_	Date	compi	iea .	104/80	Con	npiler:	N.KE	INIE		
lden Num	Owner	-	,	tion.	Date	Hydrogen Sulphide	Hard	Alka	Iron	РН	Apparent in Haze	Turbidity, Formazin	Cond	Fluc	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number 1	er or ce	ownship	Lot	Concession	Sampled ×	gen as ide H ₂ S	Hardness CaCO ₃	Alkalinity CaCO3	as Fe	at lab	arent Colour, Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	ride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved	Total	Phenois, in µg/1
0010	OBSERVATION WELL NO. 10	7			19/10/19		459	340	* 0.02	7.40			720		25.0	* 29.0	98.5	51 6	* /4.6	4 0.70	* 0,015	* 0.250	* 0.021	× 0.44	* 0.003	* 0.011	41.0
*	66				19/19		279	274	0.18	7.37			740		52.0	36.5	68.0	26.4	45.5	2.05	1.56	2.90	0.007	0.87	0.002	0.164	41.0
*	11				31/19/79		325		0.70								82.0	29.2	47.5	5.55		3.70				0.090	41.0
*	ñ				9/11/19		268	279	0.92	7.77			770		56.7	37.5	60.5	28.4			3.50	4.30	0.32	1.9	0.011	0.076	< 1. o
*	Ti				22/11/79		286	283	0.54	7.67			779		59.0	34.0	63.0	31.2	48.0	4.65	2.3	3.35	0.20	1.8	<0.05	0.05	41.0
*,	le .				26				0.70																		
* 11	li.				31/01/80		346	344	0.37	8.08			690	۷٥.۱٥	26.0	8.0	83.5	33.4	14.4	1-20	0.165	0.42	0.006	0.15	0.007	0.015	<1.0
0W11	OBSERVATION WELL NO. 11	3			3/11/79																						
* 0W12	OBSERVATION WELL NO: 12	2			12/ 0/79		33%	320	0.23	7.9			660		27.0	14.0	80.0	34.0	17.0	1.9	0.2	1-2	0.10	ZO:1	<0.02	0.28	

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1mg/l = 1ppm = 1|b/100,000 lmp. gal; 1µg/l = 1ppb. "OE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



OF GROUNDWATER SUMMARY OF CHEMICAL ANALYSES

All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG , LOT 95 , CON I , W.T.S.R. Date compiled: 1/04/80

Compiler: N. RENNIE

			•											_											7					_	_	
Iden	Owner Source	Concession Lot		Date	Bioch	Dema	2	C	arboi	n	Petroleum Hydrocarb	Lignins	1 9	Reactive	Total Solids	Sele	Arsenic	Barium	Cadn	Chro	Сорр	Cyanide	Lead	Mang	Nickel	Zinc				Anioni		
Identification	Owner or Source	Township	ot	Concession	Sampled 3	Biochemical Oxygen Demand (BOD ₃)	Demand (COD)	nical Oxygen	Inorganic	Organic	Total	Petroleum Hydrocarbons	Lignins and	1 8	tive as Si	Total Dissolved Solids	Selenium as Se	nic as As	um as Ba	Cadmium as Cd	Chromium as Cr	Copper as Cu	i)de as CN	Lead as Pb	Manganese as Mn	el as Ni	as Zn			a service of the serv	Anionic Detergent	
	OBSERVATION WELL NO. 10				19/ 10/ 79 15:22			9	12.5	2.5	95.0																					
*	h.				24/ 10/19			7	12.0	22.5	95.0																			1		
*	Ц				31/10/79			7	73.5	17.0	90.5																					
*	ll				9/11/79				70.5	18.0	88.5																			0.	. ¡A	129.
*	ħ				22/ 11/19				75.0	19.0	94.0																			٥	3.2	
* "	И				26,												ا∞. یا	.001	.04	<.005	4.02			.05	. 20	<.02	١٥.٧				_	
"	It				3/0/80			9	87.O	158	245				*** **																	
X ∞11	OBSERVATION WELL NO: 11	2			3/11/ 79																									d	2.1	
					rarar																											
								-																				one	41.000	0 20 00		

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=11b/100,000 lmp.gal; 1,4g/l=1 ppb. A - sampled on November 3, 1979. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Ontario

Township(s): GLENELG , LOT 95 , CON I W.T.S.R. Date compiled: 704/80 Compiler: N. RENNIE

Nur	Owner	_	ation	Date	Hydr	Har	Alk	Iron	РН	Appa	Turb	Conc	Fig	Chlo	Sulp	Calc	Mag	Sodium	Pota	Nitro	gen	as	N	Phospho as	orus P	Phe
Identification Number ¹	ner or	Township	Concession	e Sampled	Hydrogen as Sulphide H ₂ S	Hardness CaCO3	Alkalinity caco3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	Calcium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved	Total	Phenols,in μg/ Ι
01112	OBSERVATION WELL NO. 12	,		31/101/80		327	323	0.08	7.79			662	∠o.10	28.0	7.5	75.0	33.8	16.6	0.95	0.010	0.12	0.011	0.18	0.002	0.002	41.0
* 0W13	OBSERVATION WELL NO. 13			11/79		286	273	0.18	7.62			770		52.0	34.0	68.0	28.2	47.5	2.45	0.745	1.68	0.032	3.12	0.004	0.034	<1.0 C
*	н			25/ /11/ /79		275	263	0.13	7.64			780		71.5	29.0	63.0	28.6	53.5	2.90	0.805	1.36	0.113	3.7	0.002	0.039	<1.0
* ì	h			31/01/80		326	322	0.09	7.81			662	ZO-10	29.0	8.5	75.5	33.4	16.9	1.95	0.010	0.32	0.001	0.38	0.002	0.004	2.0
X 0W14	OBSERVATION WELL NO: 14	0		12/09/70	3	361	344	0.18	7.30			670		22.0	11.0	80.0	39.0	13.0	3.8	0.1	1.8	40.01	40.1	∠o. o z	0.26	

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1lb/100,000 lmp. gal; 1 µg/l = 1 ppb. "DE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 17/04/80

Compiler : N. RENNIE

Source Concession Dimension Observation Ob	oncession ot	Sampled and oncession of	oncession Sampled	Sampled	all A	Demand (COD)	Inorganic 78.5	Organic 8920	n Total	Petroleum Hydrocarbons	Tannins and Lignins	Reactive Silicate as Si	Total Dissolved Solids	Selenium as Se	Arsenic as As	Barium as Ba	Cadmium as Cd	Chromium as Cr	Copper as Cu	Cyanide as CN	Lead as Pb	Manganese as Mn	Nickel as Ni	Zinc as Zn		Anionic Detergent 9
OBSERVAT WELL NO. 13	ion)			16/11/19		72.5	10.0	82-5																	$\frac{1}{1}$ $\frac{1}{1}$ $\frac{31}{1}$
100. 12)				25/ 11/ 79		69.0	15.0	83.0																	
	¥				31/		80.5	1040	1120																	 _
	OBSERVATION WELL NO. 14	N)			09/ 79																					A KD.1

1 Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=11b/100,000 lmp.gal; 1, ug/l=1 ppb. A-sampled on November 3, 1979. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI, W.T.S.R. Date compiled: 17/04/80 Compiler: N.RENNIE

county	GKET							1044112	nip(s)	· Omc	NECC	,	٦٧,	0010 1	, w.	1		Date (Joinpi	icu .	780	0011	ipiici .	10.100	:101010		
Ide	Owner Source	-	cati	ion.	Date	Hydrogen Sulphide	Haro	Alk	iron	РН	Appa	Turb	Conc	Flu	Chlo	dins	Calcium	Mag	Sodium	Pota	Nitro	gen	as	N	Phospho as	orus P	Phe
Identification Number 1	ner or	Township	Lot	Concession	e Sampled s	ogen as hide H ₂ S	Hardness CaCO3	Alkalinity as	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved Reative	Total	Phenols, in µg/1
W 1	OBSERVATION WELL NO. 14	3			25/ /11/ /79		322	319	0.13	7.63			665		23.0	15.0	72.0	34.4	14.1	1.70	0.350	0.62	0.006	0.04	<0.∞1	0.003	3.0
*	ti				31/ 01/ 80		307	302	0.22	7.86			622	40.10	25.5	8.0	70.5	31.8	13.6	1.35	0.150	0.39	0.002	0.23	0.002	0.013	3.0
* 0w15	OBSERVATION WELL NO. 15	9			12/ 09/ 79		384			8.1			650		24.0									<0.1			
* tı	ii				25/ 11/ 779		322	310	0.05	7.74			630		18.0	12.5	72.0	34.4	11.0	0.70	0,020	0.23	0.001	0.18	40.001	0.002	41.0
				Deleteration of the Party																							
				1 - 1400 Term																							
0W16	OBSERVATION WELL NO. 16	2			25/ /11/ ₇₉		391	353	0.16	7.52			720		25.5	11.0	87.0	37.2	11.2	0.75	0.040	0.27	0.002	0.03	0.002	0.004	41.0

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp.gal; 1 µg/l = 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



County: GREY

SUMMARY OF CHEMICAL ANALYSES OF GROUNDWATER

All analyses except pH reported in mg/l unless otherwise indicated

Township(s): GLENELG, LOT 95, CONI, W.T.S.R Date compiled: 17/04/80 Compiler: N. RENNIE

N d	Sot Ow	Loc	ati	ion	Date	0 0	D C P	C	arbo	n	Pet	Tan Lig	SII	Total Solids	Sei	Ars	Ваг	င္မ	Chr	Cop	Суг	Lea	Mar	Nic	Zinc		Anio
Identification Number ¹	Owner or Source	Township	2	Concession	Sampled	Biochemical Oxyger Demand (BOD)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	Tannins and Lignins	Reactive Silicate as Si	Total Dissolved Solids	Selenium as Se	Arsenic as As	Barium as Ba	Cadmium as Cd	Chromium as Cr	Copper as Cu	Cyanide as CN	Lead as Pb	Manganese as Mn	Nickel as Ni	IC as Zn		Anionic Detergent
10.00	OBSERVATION WELL NO. 14	\parallel	1		DIMIY 25/ 11/79			84.5	9.5	94.0														×			
* II	п				31/			77. 0	893	970																	
															i i												133
* 0W15	OBSERVATION WELL NO. 15	7			25/ 11/ 79			G4.0	2.0	960																	A 40.1
į.																											ч
																											41
* 0⊌I6	OBSERVATION WELL NO. 16	2			25/ /11/ /79			72.5	9.0	81.5																	&.I

Location is shown in Figure 2; N.D.— Not Detected; P — Present; < — Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 µg/l=1 ppb. A-sampled on November 3, 1979.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R.

Date compiled: 17/04/80 Compiler: N. RENNIE

Count	, GRET								ilihia			*							p		-1/180		ipiici .				
Nur	Owner	Loc	-		Date	Hydr	Har	Alk	Iron	РН	in Appa	Turb	Conc	Flu	Chic	Sulp	Calc	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number 1	Owner or Source	Township	2	Concession	- ×	Hydrogen as Sulphide H ₂ S	Hardness caco ₃	Alkalinity caco3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	Calcium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenols,in µg/l
×.	OBSERVATION WELL NO. 16	2			31/ 01/ 80		317	310	0.30	7.77			618	40.10	19.5	7.5	77.0	30.2	10.4	0.75	0.050	0.34	0.001	0.30	0.003	0.005	۷۱.0
_			+																								
¥ ∞wis	OBSERVATION WELL NO. 18				31/ 10/ 79		317		0.06								70.0	34.4	40.0	1.70		0.63				0.014	۷۱.0
*	и				9/11/79		281	283	0.04	7.7 4			758		53.0	35.0	0 . حاص	28.2			1.05	/. 8 4	o.062	1.61	0.002	0.029	< 1.0
*	u				12/19		292	272	0.18	7.63			780		61.0	37.0	b7.5	30.0	48.0	3. 4 0	1.10	2.05	0.31	3.4	40.05	۷°.05	<1.0
*	n				26/11/19				0.03																		
*	п				31/01/80		315	315	0.08	7.77			640	40.IO	19.0	12.0	75.0	31.0	16.8	1-05	0.035	0.30	0.003	1.05	0.001	0.006	3.0

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp.gal; 1 µg/l = 1 ppb.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenole and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 1/04/80

Compiler : N. RENNIE

Anionic Detergent		135	0.1	0.2		
Titanium Molybdenum		;004 .00I				
Aluminum		.080				
Zinc as Zn		4.004			۷.0۱	
Nickel as Ni		۷. 004			4.02	
Manganese as Mn		z.004			.08	
Lead as Pb		<. 004			۷.07	
Cyanide as CN						
Copper as Cu		. ۵۵۹				
Chromium as Cr		4.004			۷.02	
Cadmium as Cd		2.004			<.005	
Barium as Ba					۷.02	
Arsenic as As		Z.001			١٥٥٠ ک	
Selenium as Se		Z.001			4.001	
Total Dissolved Solids						
Reactive Silicate as Si						
Petroleum Hydrocarbons						
Total	6350	84.5	81.5	94.5		2000
a r bor Organic	6270	9.0	q.o	15.0	-	1920
C Inorganic	% 0.5	75.5	72.5	<i>6</i> 9.5		77.5
Chemical Oxygen Demand (COD)						
Biochemical Oxygen Demand (BOD ₅)						
•	DIMIY 3/ 0/ 80	31/ 10/ 79	9/11/79	27/ 11/ ₇₉	26/11/79	31/80
Concession						
Lot Township						
2 3	OBSERVATION WELL NO. 16	OBSERVATION WELL NO: 18	u	u	11	ty.
Identification Number ¹	owib		*	H *	*	10

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 µg/l=1 ppb. A - sampled on November 3, 1979.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CON I W.T.S.R. Date compiled: 17/04/80 Compiler: N. RENNIE

	. 0								inp(c)												780		·p				
Ider	Owner Source		cati		Date	Hydrogen Sulphide	Haro	Alka	Iron	рн	Apparent in Haze	Turbidity, Formazin	Cond	Flu	Chic	Sulp	Calcium	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	or us P	Phe
Identification Number ¹	er or	Township	Lot	Concession	Sampled \$	ogen as hide H ₂ S	Hardness CaCO3	Alkalinity caco3	as Fe	at lab	arent Colour, Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved Reative	Total	Phenols,in pg/i
owzz	OBSERVATION WELL NO. 22				% 149		278	267	0.08	7.65			525		5.5	8.5	62.5	29.6			0.045	0.28	0.043	1.65	0.002	0.005	1.0
* TP1	TEST PIT NO. 1				31/ 10/ 14/15/9		267		0.08								58.0	29.6	50.0	5.90		5.35	0.171				<i.0< td=""></i.0<>
*	11				11 21:25		269		0.14								57.0	30.6	4 7.0	5.00		4.90				0.138	<1.0
*	н				3/1/ 79		264	268	0.07	7.70			750		54.5	32.5	60.0	27.60	50.0	6.50	4.35	6.45	0.171	2.7	0.118	0. 275	∠1.0
*	11 PIEZOMETER				9/ _{11/79}		252	262	1.15	7.68			750		55.5	33.0	55.0	27.8			3.45	4.30	0.087	3.92	0.037	0.184	1.0
*	OPEN WATER				n		253	261	0.17	7.80			740		55.0	32.0	56.5	27.2			3.65	4.80	0.014	4.14	0.068	0.180	<1.0
*	13.				22/11/19		251	261	1.46	7.78			785		71.0	35.0	54.5	27.8	67.0	3.0	3.1	4.80	0.09	2.5	0.05	0.20	∠1.0

¹ Location is shown in Figure $l_1 2$; N.D. — Not detected; < — Refers to less than; $l_1 = l_2 = l_3 = l_4 = l_4 = l_5 = l$ MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 1/04/go Compiler: N. RENNIE

Identification Number 1 Owner or Source Zinc Copper as Cu Reactive Solids Selenium as Arsenic as Cadmium as Co Chromium as Cyanide as Lead as Pb Manganese as Nickel as N Molybdenum Vanadium Carbon Silicate as Si Barium as Ba Aluminum Anionic Detergent Location Demand (BODs) Biochemical Oxygen Demand (COD) Petroleum Total Dissolved Titanium Chemical Oxygen Tannins and Township Lot Concession Inorganic Sampled Total Organic 9/1/19 OBSERVATION 5.0 OWIZ WELL 67.5 72.5 NO. 22 × TEST TP1 10/ 4.001 4.001 4.004 4.004 4.004 .07 4001-001 4004 PIT 20.0 89.5 4.004 4.004 .006 NO. 1 * 4004 .009 .163 can 003 4004 4.004 .042 C.001 C.001 26.0 94.5 4.004 4.004 .019 × 0.1 19.5 67.0 86.5 * 31.5 99.0 67.5 PIEZOMETER * 84.0 11.5 11 OPEN WATER 22/1/79 * 66.5 25.5 92.0

Location is shown in Figure 1,2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1µg/l=1 ppb.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY Township(s): GLENELG, LOT 95 , CONI W.T.S.R. Date compiled: 17/04/50 Compiler: N.RENNIE

	, OKL							ilip(3)												730						
Nun	Owner Source		ation	- 3	Hydrogen Sulphide	Haro	Alka	Iron	рН	Appa	Turbi	Cond	Fluo	Chic	Sulp	Calc	Mag	Sod	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number ¹	Owner or Source	Township	Concession	Sampled M	H ₂ S	Hardness CaCO3	Alkalinity Caco3	Iron as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance ; in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	Calcium as ca	Magnesium as Mg	Sodium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved	Total	Phenois, in µg/1
X TP1	TEST PIT NO. 1			25/		247	258	0.04	7.79			720		54.0	31.5	54.5	27.0	50.5	6.40	3.32	4.60	0.039	1.66	1.20	1.28	
* TP2	TEST		-	31/		24.4		0.04								87.5	20.4									
* "	NO. 2		+	9/1/19	1	279	274	0.04				725		52.0	31.0		28.6	40.5	1.10			0.046		0.004	0.050	<1.0 <1.0
X	. N			22/	-	265	259	0.09	7.58			722		53.0	31.0	61.0	27.2	46.5	2.05	0.6	1.45	0.19	3.6	40.05	0.11	Z1.0
*	Н			25/ /11/ /79		253	240	0.05	7.67			595		32.0	20.5	56.5	27. 2	26.6	1.45	0.36	0.56	0.044	2.5	0.005	0.025	
		\parallel																								
<u></u>																										

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 11b/100,000 lmp. gal; 1 μg/l = 1 ppb.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG , LOT 95 , CONI W.T.S.R. Date compiled: 17/04/30

Compiler: N. RENNIE

		139.				
Anionic Detergent		1	40.1	0.1		
Vanadium		4.004				
Titanium		Z.004 . 00				
Aluminum		.090				
Zinc as Zn	< .01	4.004			2.01	
Nickel as Ni	4.02	4.004			2.02	
Manganese as Mn	<.02	•010			2 .02	
Lead as Pb	.05	∠.004			.06	
Cyanide as CN						
Copper as Cu		.019				
Chromium as Cr	<.02	Z:004			€.02	
Cadmium as Cd	4.005	2,004			₹.005	
Barium as Ba	.02				<.02	
Arsenic as As	Z.001	2.001			< .001	
Selenium as Se	۷.۵۵۱	4.001			<.001	
Total Dissolved Solids						
Reactive Silicate as Si						
Tannins and						
Petroleum						
Total		84.5	76.0	80.5		
ar bor Organic		12.0	5.5	12.5		
Inorganic		72.5	70.5	68.0		
Chemical Oxygen						
Biochemical Oxygen						
Date Sampled =	26/ 11/ 79	31/10/19	9/11/79	11/79	26/ 11/ 79	
tion Concession						
Lot Township						
Owner or	· 1		t i	М	н	
	F	T P P				
Number 1	*P1	* TP 2	*	*	*"	

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=11b/100,000 lmp.gal; 1,49/l=1 ppb. A- sampled on November 3, 1979. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG , LOT 95 , CONI WITS.R. Date compiled: 17/04/80 Compiler: N. RENNIE

Nun	Owner	Loc	7	on.	Date	Hydrogen Sulphide	Har	Alk	Iron	На	Appa	Turb	Conc	Flu	Chic	Sulp	Calc	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number 1	ner or	Township		Concession	= 1	H ₂ S	Hardness caco ₃	Alkalinity caco3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	Calcium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved	Total	Phenois, in Jug/I
X TP3	TEST PIT NO.3				31/ 10/ 79		311		0.12								68.5	34.0	7.3	0.70		0.23				0.011	41-0
* 11	11				9/1/79		30%	269	0.02	7.68			640		37.0	21.0	69.0	33.0			0.020	0.36	0.001	0.32	0.002	Ø.018	<1.0
*	и				16/		309	277	0.05	7.69			700		45.5	27.5	71.0	32.0	30.5	1.40	0.010	0.49	0.001	1-11	0.005	0.017	<1.0
* 1	10				22/ /11/ /79		295	265	0.09	7.59			675		45.5	27.0	63.0	31.0	29.4	0.75	0.10	0.50	<0.0I	1.30	40.005	40.005	<1.0
*	li li				25/		297	267	0.01	7.53			705		48.0	28.5	64.0	32.8	39.0	1.50	0.015	0.55	0.002	3.0	0.003	0.013	
*	fi				31/		266	273	0.05	7.75			580	40.10	21.0	11.0	62.0	27.0	22.8	0.93	40.005	0.22	a.004	0.81	0.001	0.003	41.0
X TP4	TEST PIT NO. 4				31/19/19		314		0.04								76.0	30.0	12.2	0.95		0.06				0.002	<1. O

¹ Location is shown in Figure Z; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b|100,000 lmp. gal; 1 µg/l = 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



GROUNDWATER SUMMARY OF CHEMICAL ANALYSES OF

All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG , LOT 95, CONI WT.S.R. Date compiled: 1704/30

Compiler: N. RENNIE

					141.				
Anionia	Anionic Detergent	4	20.		0.1			+-	
Vana	Vanadium	۷.004		L				 	
Titanium	ium	002						+-	
MOLY	Molybdenum	3 (.00)						+	
Aluminum	mum	090.							
Zinc	as Zn	c.004				۲.0۱			
Nickel	as Ni	2.004				۷.02		\rightarrow	
Mang	Manganese as Mn	۷.004 .				८.02			
Lead as Pb	as Pb	۷.004 ۽				.05			
Cyanide	ide as CN								
Copper	er as Cu	P00-							
Chron	Chromium as Cr	∠.∞4				<.02			
Cadmium	nium as Cd	2.004				८.∞5			
Barium	ım as Ba					<.02			
Arse	Arsenic as As	۱۰۰۰ ک				<.001			
Selei	Selenium as Se	∠.∞۱				∠.∞۱			
Total Solids	Total Dissolved Solids								
Reactive	Reactive Silicate as Si								
Tannins Lignins	ns and					_			
Petroleum Hydrocarb	Petroleum Hydrocarbons								
n	Total	75.5	% 0.0	78.0	77.5			145	145
arbo	Organic	5.5	10.0	4.5	8.5			73.0	73.0
С	Inorganic	70.0	76.0	73.5	6 9.0	-		72.0	72.0
Chem	Chemical Oxygen Demand (COD)		L. Carlon						
Bioch	Biochemical Oxygen Demand (BOD _s)								
Date	Sampled 3	31/ 19/79	9/11/79	16/11/79	22/ / 79	26/ 11/ 79	31/	31/ 01/ 80	/01/ 80
-	Concession								
Loca	Lot Township						П		
Owne	Owner or Source	TEST PIT NO. 3	ц	и	li	lı		"	
ident Numi	Identification Number ¹	* TP3	*	* "	*	*,	*	n :	

A- sampled on November 3,1979 1 Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1ib/100,000 lmp.gal; 1μg/l=1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 17/04/80 Compiler: N. RENNIE

County	, GREY							10M112	mpto			,	, ,					oute (,	,		opiioi i				
Nu	Owner	Lo	cat	tion.	Date	Hydr	Har	Alk	Iron	Н	App	Turb	Conc	Fu	Chic	Sulp	Calc	Mag	Sodium	Pota	Nitro	gen	as	N	Phospho as	orus P	Phe
Identification Number 1	Owner or Source	Township	Lot	Concession	San	Hydrogen as Sulphide H ₂ S	Hardness caco ₃	Alkalinity as CaCO3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	Calcium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahi	Nitrite	Nitrate	Dissolved	Total	henols,in µg/1
* TP4	TEST PIT NO. 4				9/1/79		288	268	0.08	7.63			720		52.0	30.5	70.0	27.4			0.015	0.74	0.002	1.72	0.002	0.038	<i-0< td=""></i-0<>
* "	П				11/79		279	264	0.07	7.56			740		51.0	30.0	65.5	28.0	46.0	1.30	∠o. I	-1.10	0.27	5.5	4 0.05	0.14	<1·0
* 1	li .				25/ /11/ /79		270	254	۷٥.٥١	7.64			665		41.0	24.0	62.0	27.8	35.0	1.20	0.190	0.60	0993	4.2	0.001	0.023	
* TP5	TEST PIT NO. 5	and the second s			31/10/79		329		0.04								79.0	31.8	12.2	0.95		0.06				0.002	L1.0
*	tı				9/1/79		323	301	0.02	7.65			625		22.0	6.0	80.0	29.8			0.010	0.14	40.001	0.04	0.007	0.023	41.0
*	11				16/	1	30%	297	0.04	7.52			620		23.0	9.0	75.0	29.2	14.3	0.65	0.010	0.16	0.001	0.02	40.00 l	۷٥.00۱	<1.0
*	lı .				22/1/79		288	281	0.09	7.51			590		22.5	9.0	68.0	28.6	13.0	0.70	40.1	0.20	40.01	0.1	<0.05	40.05	<1.0

¹ Location is shown in Figure 2; N.D. - Not detected; < - Refers to less than; 1mg/l= 1ppm = 1lb/100,000 lmp. gal; 1µg/l= 1ppb.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI WT.S.R. Date compiled: 7/04/90

Compiler: N. RENNIE Identification Number ¹ Barium as Ba Cyanide as Owner or Source Vanadium Arsenic as As Manganese Reactive Selenium as Cadmium as Co Chromium as C Copper as Cu Nickel as N Zinc as Zn Aluminum Molybdenum Titanium Lignins Location Demand (BOOs) Biochemical Oxygen Carbon Hydrocarbons Silicate as Si Total Dissolved Chemical Oxygen Tannins and Township Lot Concession Inorganic Sampled Organic × 9/11/79 TEST TP4 PIT 40.1 70.0 10.0 80.0 NO. 4 * 11/19 0.1 10.5 81.0 70.5 * 4.001 4.001 4.02 4.005 4.02 4.03 4.02 4.02 31/ 19/ 79 * TP5 TEST 1001 4.001 <.004 <.004 <.004 <.004 .071 kool.001 <.004 4.004 4.004 .006 PIT 82.0 5.0 87.0 NO. 5 % * 84.0 5.0 16/1/79. * 3,5 84.0 22/ /11/79 × 6.5 83.5

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=11b/100,000 lmp. gal; 1 µg/l=1 ppb. A - sampled on November 3, 1979. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

Country COT U

Townshinds: GIENEIG LOTGE ONT W.T.S.R. Date compiled: 1704/ Compiler W. PENINIE

County	: GREY						Towns	hip(s)	GLE	NELG	, 6	45,0	ONI	w. 1	, , , , , ,		Date	compi	led:	04/80	Con	npiler:	N. REI	UNIE		
Iden	Owner		ation	Date	Hydrogen Sulphide	Harc	Alka	Iron	рН	Appa in 1	Turbidity, Formazin	Cond	Fluc	Chlo	Sulp	Calcium	Mag	Sodi	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number ¹	Owner or Source	Township	Concession	Sampled S	as H ₂ S	Hardness caco3	Alkalinity Caco3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	ium as Ca	Magnesium as Mg	Sodium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenols, in µg/1
X TP5	TEST PIT NO. 5			25/ /11/ /79		296	298	(0.01	7.64			580		16.5	9.5	67.5	31.0	9.8	0.50	0-100	0.12	<0.00	<0.01	0.602	0.003	
																				lix						- - -
ê																										

¹ Location is shown in Figure 2: N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp. gal; 1 \mug/l = 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 1704/80 Compiler: N. RENNIE

Z G	So	L	oca	tion	D	9 9	P	0	С	arbo	n	I P	г д	s :	, a	n ¬	S	A	8	C	C	C	C	_	Z	z	Z			
m ti	Owner Source	_	_		ř	ocne	mar	3				ydro	Tannins Lignins	ilica	Beactive	Total	eler	Arsenic	ariu	adn	hro	Copper	yan	ead	lang	Nickel	Zinc			
Identification Number 1	Owner or Source	Township	o.	Concession	Sampled M		Demand (COD)	Chemical Oxygen	Inorganic	Organic	Total	Petroleum Hydrocarbons	Tannins and Lignins	Silicate as Si		Total Dissolved	Selenium as Se	nic as As	Barium as Ba	Cadmium as Cd	Chromium as Cr	er as Cu	Cyanide as CN	Lead as Pb	Manganese as Mn	ei as Ni	as Zn	я		
* TP5	TEST PIT NO.5				26/ 11/ 79												2.001	١٥٥١ ک	۷.02	4.005	₹.02			4.03	۷.02	۷.02	١٥٠ >			
																														145.
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																														1
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	8																													
1	n is shown																											L	 	

Location is shown in Figure 2; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1µg/l=1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.

APPENDIX F

SUMMARY OF CHEMICAL ANALYSES OF SURFACE WATER



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG , LOT 95 , CONI , W.T.S.R. Date compiled: 17/04/80 Compiler: N. RENNIE

County	· OKL I				*			niib(a)			,	1 70	, 0010	,		THE PERSON NAMED IN COLUMN TWO IS NOT	CATOLOGICA MICHAEL	NAME OF TAXABLE PARTY.		1780		mannaman-	On the latest of	THE RESERVE OF THE PERSON NAMED IN		principal and the last of the
Nur	Owner	-	ation	Date	Hydr Sulp	Har	Alk	Iron	НФ	Appa	Turb	Conc	Flu	Chlo	Sulp	Calc	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number ¹	ner or	Township	Concession	e Sampled x	Hydrogen as Sulphide H ₂ S	Hardness CaCO3	Alkalinity caco3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	Calcium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenols,in µg/l
1	ROCKY SAUGEEN R. At C.P.R. TRACKS			22/ 10/ 19	T	240	223	0.07	g.18			447		5.0	15.5	53.0	26.4	2.7	1.00	0.010	0.37	0.003	0.84	0.003	0.008	41.0
И	ц			27/ 10/ 70		280	256	0.06	8.16			490		5.0	12.0	68.0	26.8	2.0	1.05	<0.005	0.32	0.001	1.49	۱ ۵۰.۵۰	0.001	<1.0
Iţ	fi			29/ 10/ 79		280	248	0.07	8.06			484		5.5	13.0	68.0	26.8	2.0	0.85	0.005	0.30	0.001	1.40	0.00 2	0.004	۷[.0
II .	Ţ			9/11/79		255	240	0.04	7.36			490		6.5	14.0	٥.٥٥	25.6	2.9	0.90	0.005	0.25	0.001	1.33	<0.001	0.003	<1.0
П	łı			22/		257	236	0.09	8.04			472		5.0	14.0	59.0	26.6	2.2	0.50	0.1	0.30	<0.01	1.3	<0.05	<0.05	<1.0
ly	II			31/		281	2602	0.05	8.15			526		5.5	12.5	66.0	28.2	2.4	0.80	0.010	0.26	0.002	2.30	0.008	0.013	<1.0
												- Conscionation and the constitution of the co														
								ir Nobella (1900)																		
R5R 1	ROCKY SAUGHE RIVER 50 ft above out fall			11/10	1	255	239	1.49	8.19		0.35	464		5.5	14.0	59.0	26.0	2.60	0.80	0.005	0.28	0.001	0.26	0.004	0.005	21.0

¹ Location is shown in Figure 1; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b|100,000 lmp. gal; 1 µg/l = 1 ppb. MOE 0495 11/78



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI, WT. S.R. Date compiled: 17/4/80

Compiler: N. REDNIE

A proper Determine				149.	p.1		,	
Vanadium				4	«		,	(,004
Titanium								.002
Molybdenum					4			100
Aluminum					-04			.07
Zinc as Zn		0١.			۱۵. ک			<.00 4
Nickel as Ni		۷.02			2.02	0		2.004
Manganese as Mn		۷.02			∠.02			.010
Lead as Pb		۷۵٥3			∠.03			∠.00 4
Cyanide as CN		ده.۵۱						
Copper as Cu		۷.0۱			۷.0۱			0.02
Chromium as Cr		۷.02			<.02			2.004
Cadmium as Cd		∠. ∞ 05			∠.⊘05			<.004
Barium as Ba		.02			۷.02			
Arsenic as As	,	۱۵۰.۵			۷.03			۱ ۰۵۰٪
Selenium as Se		∠.001			ا‱، ک			K.001
Total Dissolved Solids								
Reactive Silicate as Si								
Petroleum Hydrocarbons								
Total	5.ام	68.0	68.5	64.5	65.5	64.0		6 7.0
a r bo Organic	11.5	9.0	10.0	7.0	8.5	2.5		q.0
Inorganic	50.0	59.0	58.5	57.5	5 7.0	61.5		58.0
Chemical Oxygen Demand (COD)	12.0	17.0						24
Biochemical Oxygen Demand (BOD)	1.0	0.4		,				0.1
Date Sampled 3	22/19/19	27/ /10/ /19	29/ 10/ 79	9/ 11/79	22/ 11/ ₇₉	31/01/19		6/11/79
Concession								
Lot Township								
Owner or Source	ROCKY SAUGEEN RIVER at C.P.R. TRACKS	ь	п	ís	н	ij		ROCKY SAUGEN RIVER SOFT above outfall
Identification Number ¹		Ju	lr	и	μ	18		R5R 2

Location is shown in Figure 1; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1,µg/l=1 ppb. MOE 0495 11/78



All analyses except pH reported in mg/l unless otherwise indicated

County' GREY

Ontario

Township(s): GLENELG, LOT 95, CONI W. T. SR. Date compiled: 1704/6 Compiler: N. RENNIE

County	1: GKEY							LOMIN	mpte	· OLL		, 101	15,	COIOT	_ w.	1. D.T.		Date i	Joinbi	icu .	1/80	0011	ibile.	15.15			
Nun	Owner Source		-	tion.	Date	Hydrogen Sulphide	Haro	Alka	iron	На	Appa	Turbi	Cond	Fluc	Chic	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number 1	er or	Township	Lot	Concession	3	as H ₂ S	Hardness CaCO3	Alkalinity caco3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved	Total	Phenols,in yg/l
RSR2	ROCKY SAUGREN RIVER 50ft above outfall				9/11/19		260	242	0.12	8.03			480		5.5	13.5	61.0	26.0	2.7	0.90	0.005	0.47	0.001	1.30	<0.∞0 i	0.017	Z1.0
tj -	11				22/ 11/ 79	TO THE REAL PROPERTY OF THE PR	258	237	0.08	8.07			474		6.5	/4.5	59.5	26.6	3.0	0.75	<0.1	0.30	<0.01	1.2	<0.05	<0.05	<u>د</u> ۱.0
																							٨				
RSR 3	ROCKY SAUGEEN R. ~ 300 ft below outfall				6/ 1/49		259	246	0.07	8.09		1.6	519		12.0	14.5	60.0	26.4	8.0	1.65	0.300	0.96	0.035	1.34	0.168	0.400	<i.0< td=""></i.0<>
и	п				9/11/19		260	241	0.06	8.10			481		5.5	13.5	61.0	26.2	2.6	0.90	<0.∞5	0.27	0.001	1.31	40.001	0.003	۷۱.0
и	п		A.		22/11/19		259	236	0.07	8.10			476		6.5	14.5	59.5	26.8	2.8	0.75	40.1	0.30	<0.01	1.2	∠o.o5	20.05	۷1.0
																									Automotive en		

¹ Location is shown in Figure 1; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp. gal; 1 µg/l = 1 ppb. MOE 0495 11/76



All analyses except pH reported in mg/l unless otherwise indicated

County : GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date compiled: 17/04/80

Compiler : N. RENNIE

		1	151.			1	
Anionic Detergent		<0.1		4		40.	
Vanadium				01 < .00			
Molybdenum				0,000			
Aluminum			r E	.060			
Zinc as Zn				∠.004			
Nickel as Ni				<.00 4			
Manganese as Mn				.005			
Lead as Pb				۷.00 4			
Cyanide as CN							
Copper as Cu				١٥.			
Chromium as Cr				۷.00 4			
Cadmium as Cd				<.00 4			
Barium as Ba							
Arsenic as As				١٥٥٠ >			
Selenium as Se				∠.∞1			
Total Dissolved Solids							
Reactive Silicate as Si							
Tannins and Lignins							
Petroleum Hydrocarbons							
Total	65.5	65.5		71.5	0.0	65.0	
or bo	8.0	8.5		11.0	8.5	7.0	
C Inorganic	57.5	57.0		60.5	57.5	58.0	
Chemical Oxygen Demand (COD)				20			
Biochemical Oxygen Demand (BOD ₅)				1.2			
Date Sampled 3	% %	22/1/19		11/ 79	9/ _{11/79}	22/ 11/19	
Concession							
Lot							
wner or burce	ROOKY SAUGHEN RIVER 50ft above outfall	ū		ROCKY SAUGEEN R. 1300ft below outfall	и	Ü	
Identification Number ¹	18R2	и		rsr3	٠	и	

Location is shown in Figure 1; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1 µg/l=1 ppb. MOE 0495 11/78



All analyses except pH reported in mg/l unless otherwise indicated

County: Date compiled: Compiler:

oount)									niip(o)									0000	o o mp								
Nur	Owner	-		tion.	Date	Hydrogen Sulphide	Har	Alk	Iron	РН	Appa	Turb	Conc	Fu	Chlo	Sulp	Calcium	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number 1	ner or rce	Township	Lot	Concession	e Sampled x	ogen as hide H ₂ S	Hardness CaCO3	Alkalinity CaCO3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance, in micromhos/cm-25°C	Fluoride as F	Chloride as Cl	Sulphate as SO4	ium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved	Total	henols,in pg//
25R4	ROCKY SAUGHEA RIVER downstream from Lagoon				22/ 10/ 19		241	224	0.09	8.18			454		6.0	15.0	53.0	26.2	3.1	1.2	0,020	0.61	0.002	0.80	0.001	0-010	
Ří:	Ñ				27/ 10/ 79		266	244	0.06	8.17			493		6.0	12.0	62.5	26.6	2.4	1.9	Ko.o⊙5	0.30	0.001	1.41	0.001	0.003	41.0
Įf.	jų.				29/ 10/ 79		282	248	0.07	8.14			498		65	12.0	68.0	27.2	2.9	1.00	0.005	0.29	0.002	1.55	0.003	0.005	Z+.0
Ж	èş				9/11/79		260	249		8.11			499		6.5	/4.0	60.0	26.6	3.3	1.00	0,010	0.39	0.002	1.60	0.007	0.041	<1.0
ţ,	ЗV				22/1/19		271	249	0.11	8.08			505		8.0	/4.5	61.5	28.4	3.7	0.90	۷٥.۱	0.35	40.01	ا ۱۰	<0.05	40.05	41.0
rs.	ts				26/		218	197	0.16	7.11			414		6.5	16.0	51.0	22.0	2.9	1.00	0.015	0.43	0.004	0.76	0.006	0.015	
h	н				11/19		251	229	0.12	8.07			472		6.5	16.0	58.0	25.%	3.2	/. 05	0.015	0.42	0.003	1.46	0.001	0.011	
л	fi				31/01/80		282	261	0.04	8.16			518		6.0	12.5	66.0	28.4	2.8	0.75	0.005	0.21	0.002	1.94	0.004	0.006	۷۱.0

¹Location is shown in Figure 1; N.D. – Not detected; < – Refers to less than; 1 mg/l = 1 ppm = 1 lb/100,000 lmp.gal; $1 \mu \text{g/l} = 1 \text{ ppb.}$ MOE 0495 11/78



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG , LOT 95 , CONI W.T.S.R. Date compiled: 1/04/80

Compiler:

-		_																			/ 00						 	_
dent	Owner or Source		_	tion	Date	Bioch	Demand ((arbo	n	Petroleum Hydrocarb	Tannins Lignins	Reactive	Total Dit	Sele	Arse	Barium	Cadr	Chro	Copp	Cyanide	Lead	Man	Nick	Zinc		Anionic	
Identification	er or	Township	Lot	Concession	Sampled *	Biochemical Oxygen Demand (BOD ₅)	Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	ins and	ate as Si	Total Dissolved Solids	Selenium as Se	Arsenic as As	um as Ba	Cadmium as cd	Chromium as Cr	Copper as Cu	nide as CN	Lead as Pb	Manganese as Mn	Nickel as Ni	as Zn			Detergent
R4	ROCKY SAUGEN RIVER downstream from lagoon				27 18/19	0.8	9.9	52.0	10.0	62.0																		
n	м				19/19	0.5	9.7	58.5	q.5	68.0					⟨.∞।	ړ.∞۱	.04	<. 005	4.02	اه. ک	۷.0۱	۷.03	2.02	<.02	١٥.٧			
tı	tı				29/ 10/ 79			59.5	8.0	67.5																		
lı .	11				9/1/19			60.5	6.5	67.0												es.					ζO	٦ ^ ئ
11	11			ы	22/ 11/ 19			60.0	6.5	66.5																	4).I
lı	и				26/11/19										L 001	۷.03	4.02	∠. ∞ 5	۷.02	۷.0۱		۷.03	4.02	۷.02	۷.0۱	۷.02		
h	n				26/																							
U	U				31/			62.0	3.0	φ 5 .0																		

1 Location is shown in Figure 1; N.D. — Not Detected; P — Present; < — Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal · 1 μg/l= 1 ppb. MOE 0495 11/78



All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T. S.R.

Date compiled: 18/04/30 Compiler: N. RENNIE

ounty	GKEY							IOWIIS	mp(s)	. 000		,	15,0					Date (Joinpi	icu . ,	4/80	00	·p·i·c·	1011			
Nur	Sou	-		ion.	Date	Hydr Sulp	Har	Alk	Iron	рн	App	Turb	Conc	Flu	СНІ	Sulp	Calc	Mag	Sodium	Pota	Nitro	gen	as	N	Phosph as	orus P	Phe
Identification Number 1	Owner or Source	Township	2	Concession	2	Hydrogen as Sulphide H ₂ S	Hardness CaCO3	Alkalinity CaCO3	as Fe	at lab	Apparent Colour, in Hazen Units	Turbidity, in Formazin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	Chloride as CI	Sulphate as SO4	Calcium as Ca	Magnesium as Mg	ium as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenois, in µg/1
P1	POND NW of SEWAGE LAGOON				27/19		317	294	0.17	7.47			615		17.0	17.0	83.5	26.2	9.6	1.95	0.005	0.77	0.003	40.01	0.002	0.022	1.0
u	.tr				22/ 11/ 79		249	243	0.36	7.36			555		39.5	19.0	65.0	21.0	27.6	3.50	0.2	1.48	20,01	۷٥.۱	0.10	0.23	<1.
ų	н				31/ /01/ /80		416	412	5.3	7.07			790		24.0	5.0	109	34.8	19.0	2.50	0.040	2.45	0.008	0.07	0.050	0.455	12.0
P2	POND SOUTH of SEWAGE LAGOON				II/ 10/ 79		210	199	0.06	7.95			447		19.5	8-5	51.5	19.8	10.0	1.60	0.010	0.20	0.005	0.86	۷٥, ٥٥١	0.001	۷۱.

¹ Location is shown in Figure 1; N.D. - Not detected < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 lmp. gal; 1 µg/l = 1 ppb. MOE 0495 11/78



All analyses except pH reported in mg/t unless otherwise indicated

County: GREY

Township(s): GLENELG, LOT 95, CON I W.T.S.R. Date compiled: 18/04/80

Compiler: N. RENNIE

	6	-	155	4	\downarrow			
Anionic Detergent	(0	0		+		_		
						_		
Zinc as Zn	4.01					۷.σ۱		
Nickel as Ni	4.02					4.02		
Manganese as Mn	0.07	0.02				۷.02		
Lead as Pb	. 01	2.03				∠.03		
Cyanide as CN	/o.o.l	0.0(دم. OI		
Copper as Cu	. 01	2.01				۷.0۱	\	
Chromium as Cr	Z .OZ	C.02				2.02		
Cadmium as Cd	∠.005					∠.∞5		
Barium as Ba	.02	.02				.02		
Arsenic as As	0.00	0.001				٥٠.00١		
Selenium as Se	/ MI					K.001		
Total Dissolved Solids								
Reactive Silicate as Si								
Tannins and Lignins								
Petroleum Hydrocarbons								
Total	975	77.0	131			52.5		
a r boi	22.5		200			4.5		
Inorganic	75.0	57.5	ш			48.0		
Chemical Oxygen Demand (COD)	44					31		
Biochemical Oxygen Demand (BOD ₅)	2.0					1.2		
Date Sampled *	<i>IJ</i> 10/ 79	/79 22/ 11/ 79	31/ 01/ 80		27/	27/10/19		
Concession								
Lot Township				-				
ĕ ₹	POND NW of SEWASE	LAGOON	h		POND SOUTH	of SEWAGE LAGOON		
Identification Number ¹	P1	li	lı			P2		

Location is shown in Figure 1; N.D.— Not Detected; P — Present; < — Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1,µg/l=1 ppb. MOE 0495 11/78

APPENDIX G

SUMMARY OF CHEMICAL ANALYSES
OF SEWAGE LAGOON EFFLUENT



OF CHEMICAL ANALYSES OF SEWAGE LAGOON EFFLUENT SUMMARY

All analyses except pH reported in mg/l unless otherwise indicated

County: GREY

Ontario

Township(s): GLENELG, LOT 95, CONT W.T.S.R. Date compiled: 18/04/0_ Compiler: N.RENNIE

(Charles and American	Aller and the second and the second	Principle of	W THE REAL PROPERTY.		-	Total Control						1	,			. 5.17.			comp.		/ / 8	9 001	ipner	. 10.10	101016		
lden Num	Owner Source	-		tion.	Date	Hydrogen Sulphide	Hard	Alka	Iron	P	Appa in I	Turbidity, Formazin	Cond	Fluc	Chloride	Sulphate	Calcium	Mag	Sodium	Pota	Nitro	ogen	as	N	Phosph as	orus P	Phe
Identification Number 1	er or ce	Township	ot	Concession	Sar	gen as ide H ₂ S	Hardness CaCO3	Alkalinity CaCO3	as Fe	at lab	Apparent Colour, in Hazen Units	dity, in azin Units	Conductance , in micromhos/cm-25°C	Fluoride as F	ride as CI	hate as SO4	ium as Ca	Magnesium as Mg	um as Na	Potassium as K	Free Ammonia	Total Kjeldahl	Nitrite	Nitrate	Dissolved Reative	Total	Phenois, in µg/1
	SEWAGE LAGOON EFFLUENT				28/ 105/ 19														parameter de l'acce su maire au		2.2	8.20			ar and a second point in the	1.00	
n	11				27/ 09/ 79		278	284	0.05	7.7		8.8	782		60.0	·	64.0	29.0		The state of the s		Date to the state of the state		0.1			
ц	lı				18/10/19		257	292	0.02	7.69			790		57.0	34.0	55.0	29.0	61.0	7.40	7.1	12.0	0.41	۷٥.١	0.10	3.08	<1.0
11	П				19/10/79		256	292	0.12	7.61			790		57.5	35.5	55.0	28.8	53.5	8.35	7.5	11.8	0.33	0.04	2.35	3.04	<1.0
d	lı				3/11/79								770			34.0	58.5	27.0	54.0	8.25	6.7	10.0	0.54	40.1	2.45	2.92	۷۱۰۵
11	II				16/19		261						765			33.0	60.0	27.0	53.0	8.00	4.8	11.0	0.43	0.5	2.55	3.22	41.0
* !:	It			-	23/	-																3.65				1.44	۷۱.٥
*	lı				25/11/79		246	268	0.02	8.01			730		55.0	32.5	54.5	26.6	51.5	7.60	4.3	6-30	0.04	1.0	2.45	2.75	
11	fi				31/01/80		275	327	0,20	7.35			800	0.33		28.5	65.0	27. 2	41.0	7.00	13.5	16.1	0.01	۷٥.۱	3.20	3.50	25.0

¹ Location is shown in Figure 1; N.D. - Not detected; < - Refers to less than; 1 mg/l = 1 ppm = 1|b/100,000 | mp. gal; 1 ppb. MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.



SUMMARY OF CHEMICAL ANALYSES OF SEWAGE LAGOON EFFLUENT

All analyses except pH reported in mg/l unless otherwise indicated

County:

Township(s):

Date compiled:

Compiler:

Identifica Number ¹	Owner or Source	T	\neg	ion	Date	Bioche	Chem	С	arboi	n	Petroleum Hydrocarb	Tannins Lignins	Reactive	Total Solids	Selenium	Arsenic	Barium	Cadm	Chromium	Copper	Cyanide	Lead as Pb	Manganese	Nickel	Zinc	Aluminum	Anionic	1
Identification Number ¹	er or	Township	Lot	Concession	Sampled 3	Biochemical Oxygen Demand (BOD ₅)	Chemical Oxygen Demand (COD)	Inorganic	Organic	Total	Petroleum Hydrocarbons	Tannins and Lignins	te as Si	Total Dissolved Solids	ium as Se	ic as As	m as Ba	Cadmium as Cd	nium as Cr	er as Cu	de as CN	as Pb	anese as Mn	as Ni	as Zn	num		c Detergent
	SEWAGE LAGOON EFFWENT				28/ 05/ 19	13.6								24														
lg	W				27/09/19									<15													\perp	
ļs	41				18/10/79	25.6	104	71.0	33.0	104										<.25		1.0	.30		16.0	1.5		
ìr	μ				19/10/79			71.5	32.5	104																		1 65 T
н	И				3/11/79	13	464	63.0	34.5	97.5				8.0														
h	11				16/	28		61.0	34.0	95.0				26														
* "	11				23/	6.1		62.0	32.5	94.5				5.0													-	0.1
* 11	К				25/11/79										2.001	۷.03	.04	¢.005	Z.02			4.03	.05	4.02	<.01			
lt.	lı .				31/01/80	46	111	84.0	14.0	98.0				36														

Location is shown in Figure 1; N.D. - Not Detected; P - Present; < - Refers to less than; 1 mg/l=1 ppm=1lb/100,000 lmp.gal; 1µg/l=1 ppb.

MOE 0495 11/78 * Filtered through glass fiber filter paper (2 microns) except for phenols and carbons.

APPENDIX H

SUMMARY OF BACTERIOLOGICAL ANALYSES
OF GROUNDWATER



Number of bacterial colonies per 100 ml

County: GREY

Ontario

Township(s): GLENELG, LOT 95, CON I W.T.S.R. Date Compiled: 18/4/90 Compiler: N.RENNIE

2.0 101	1.01421												1
Nu	Own	Lo	cati	o n	D a	Feca Coli1	B B	Co	Fec Str	Pse	Sulph Reduc Bacte	Hete Bac	
Identification Number 1	Owner or Source	Township	Lot	Concession	Date ×	Fecal Coliform Bacteria	Background Bacteria	Coliform	Fecal Streptococci Bacteria	<u>Pseudomonas</u> aeruginosa	Sulphate Reducing Bacteria	Heterotrophic Bacteria	
ഡ1	OBSERVATION WELL NO. 1				19/10/79	A300	35,000	A5000	A 200				
п	ŢŢ.				28/ 10/19	21,000	270,000	C120,∞0	A 900				
il	tę				20/11/79	P2-1	200,000	190,000	G /500	340			
h	11				26/11/19		100,000	72,000	2000	085			
te	ñ				31/01/90		Z1000	Z 1000	۷10				
οω 2	OBSERVATION WELL NO. 2				10/79	4 100	6240,000	C14,000	A100				
11	n,				29/ 10/ 79	A20	G 2,400,000	A 900	AGO				

 $^{^{1}}$ Location is shown in Figure 2 ; < - Refers to less than; >- Refers to greater than.

An accompanying "L" or "G" indicates that the actual density was less than or greater than the number reported.

An accompanying "C" or "A" indicates that the result is a best estimate and is less precise than other determinations.

OBS indicates that the analysis has been invalidated because of interference problems.



Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date Compiled: 18/4/80 Compiler: N. RENNIE

	I. ORLI					3111p (37. = = =	. ,						1
Identifica Number ¹	Owner or Source		cati		Date Sampled	Fecal Colifo Bacter	Backgr	Colifo	Fecal Strept Bacte	Pseu	Sulph Reduc Bacte	Heterotrop Bacteria	
Identification Number ¹	C T	Township	Lot	Concession	D M Y	eria	Background Bacteria	form	Fecal Streptococci Bacteria	<u>Pseudomonas</u> aeruginosa	hate cing eria	Heterotrophic Bacteria	
	OBSERVATION WELL NO. 2				26/ 11/79	5100	68,000	71,000	1410	C 32			
u	h				26/ 11/19		90,000	109,000	G1500	250			-
u	ц				31/01/20	110	500	1000	A 10				
													15
											,		
0w3	OBSERVATION WELL NO. 3				10/79	۷100	G 240,000	C9000	<100				
и	11				20/ 11/19	۷ ۱٥	2000	۷100	A 10	<4			
h	li				31/ /01/ /80	AIO	₩	230	A10				

¹Location is shown in Figure λ ; < - Refers to less than; > - Refers to greater than.

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Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T. S.R Date Compiled: 19/04/80

Compiler: N.RENNIE

Joun ()	: GREY				LOWIIS	111 p (371 OPE	10000, 101 13	, WAL W.			120		1
Nu	Own	Loc	catio	n	Da	Fecal Colife Bacte	B B	Вас	Fecal Strept Bacte	Pse	Sulph Reduc Bacte	Hetero	
Identification Number 1	Owner or Source	Township	Lot	SSI	Date Sampled	Fecal Coliform Bacteria	Background Bacteria	Coliform	Fecal Streptococci Bacteria	Pseudomonas aeruginosa	ulphate educing acteria	Heterotrophic Bacteria	
										,			
004	OBSERVATION WELL NO. 4				18/10/79	Z100	6240,000	G17,000	2100				
đ	11			×	20/	< 10	44,000	R 600	Z10	<4			
ы	11				31/01/80		100	410	A40				164.
0w5	OBSERVATION WELL NO. 5				18/10/79	4100	51,000	23,000	A 300				
h	и				20/		7,000	11,000	230	< 4	13		
þ	lı.				31/01/79		120	430	A 10				

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Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CON. I W.T.S.R Date Compiled: 19/04/62

Compiler: N. RENNIE

ou n t	Y: GREY				10wn:	snip(s). GLE	Nece, wi	45, 600. 1	W. F. S. R. Da (e Compiled	80	Compiler	-
Identification Number 1	Owner or Source	L Township	Catio	SS	Date ×	Fecal Coliform Bacteria	Background Bacteria	Coliform Bacteria	Fecal Streptococci Bacteria	<u>Pseudomonas</u> aeruginosa	Sulphate Reducing Bacteria	Heterotrophic Bacteria	
												,	
oωω	OBSERVATION WELL NO: 6				20/ 11/ 79	9700	52,000	103,000	14,600	160			
i,	15				31/90	A 20	<100	∠ 100	2100				
-													
0W7	OBSERVATION WELL NO. 7				20/11/19		31,000	45,000	/220	116			
h	ж				31/	۷۱٥	∠100	۷۱۵٥	۷۱٥				
								·					

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Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CONI. W.T.S.R. Date Compiled: 19/04/80 Compiler: N. RENNIE

	7.01-1	3								(8)	m TI (2)	m I	ı
NG	Own	Lo	cati	o n	Sa	Co Co	8 8	Bac Co-	Fecal Strept	Se	Sul _l Red	lete }ac	
Identification Number 1	Owner or Source	Township	Lot	SSK	Date ×	Fecal Coliform Bacteria	Background Bacteria	Coliform Bacteria	Fecal Streptococci Bacteria	<u>Pseudomonas</u> aeruginosa	Sulphate Reducing Bacteria	Heterotrophic Bacteria	
					J ₁ m ₁ ·								
യു	OBSERVATION WELL NO. 8				20/	A 20	26,000	A300	A30	20			
и	11				31/01/80		210	ДЧО	A 10				
						H							
owq	OBSERVATION WELL NO. 9				20/1/9	<10	3100	A 100	∠ 10	۷4			
h	11				31/01/20	<10	610	AIO	210				

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Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELS, LOT 95, CONI W.T.S.R. Date Compiled: 19/04/80 Compiler: N. RENNIE

	,, 0				(A = 1 = 1 = 1	5111 p (6) . ODC	· .	,			7 80		 7
Identification Number ¹	Owner or Source	Lownship	cati o *	SSIC	Date ×	Fecal Coliform Bacteria	Background Bacteria	Coliform Bacteria	Fecal Streptococci Bacteria	Pseudomonas aeruginosa	Sulphate Reducing Bacteria	Heterotrophic Bacteria	
0W10	OBSERVATION WELL NO. 10				20/ 11/ 79	2400	19,000	26,000	600	C 36			
u	'n				31/80	۷10	< 100	∠100	۷10				
0W11	OBSERVATION WELL NO. 11				20/ /11/ /79	2100	33,000	17,000	450	28			
n	u				31/01/90		2200	< 100	۷10				
οωιλ	OBSERVATION WELL NO. 12				20/	/300	11,000	9000	520	24			
и	n				31/01/80	۷ ا۵	∠100	A100	۷10				

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Number of bacterial colonies per 100 ml

County: GREY

Ontario

Township(s): GLENELG , LOT 95 , CONI W.T.S.R. Date Compiled: 19/04/80 Compiler: N. RENNIE

, ORET												
So o	Lo	catio	o n	Da	Вас	ස ස ව ව	Со	Fec Str	Pse	Su- Red Bac	Hete Bac	
ner r urce	Township	Lot	SS		cal liform cteria	ckground	liform cteria	eptococci cteria	uginosa	phate lucing iteria	erotrophic eteria	
				0,111								
WELL				20/	470	/0,000	3400	400	8			
11					<10	500	A300	<10				
OBSERVATION WELL NO. 14				20/ /11/ /79	Z10	1500	A100	۷.10	8			
ıı						700	1200	∠10				
	OBSERVATION WELL NO. 13	OWNER TOWNShip OBSERVATION WELL NO. 13 OBSERVATION WELL NO. 14	OBSERVATION WELL NO. 13 OBSERVATION WELL NO. 14	Source Location Concession OBSERVATION WELL NO. 13	Source Location Sampled Date Concession Concession Dimit OBSERVATION WELL NO. 13 OBSERVATION WELL NO. 14	Source Location Sampled Coliform Concession Date Township OBSERVATION WELL NO. 13 Description Downer A70 A70 A70 A70 A70 A70 A70 A7	Colifor Bacterial Bacter	Source Coliform Sample Coliform Bacterial Coliform Concession DIMIY A70 10,000 3400	Docation Sample Background Backgroun	Concession Some part Som	Description Some plant So	No. 13 No. 14 No. 15 No. 16 No. 16 No. 16 No. 16 No. 17 No. 18 N

¹Location is shown in Figure 2; < - Refers to less than; > - Refers to greater than.

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Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date Compiled: 19/04/90 Compiler: N. RENNIE

										1,740		
S o Ow	Lo	cati	o n	D a	Ba C e	B B	Co Ba	St. Ba	P S	Su Re	Het Ba	
ner r urce	Township	Lot	SSic		cal liform cteria	ckground	liform cteria	cal reptococci cteria	<u>eudomonas</u> ruginosa	Iphate ducing cteria	erotrophic cteria	
										ı		
OBSERVATION WELL NO: 15				20/ /11/ /79	۷10	8500	1000	<10	<4			
n				3/0/80	< 10	40	A30	۷.10				
OBSERVATION WELL NO: 16				20/ 11/79	۷10	49,000	(300	A 10	<4			
tı				31/01	۷10	20	A20	۷10				v 4
												=
												4
	WELL NO. 15 11 OBSERVATION WELL NO. 16	OBSERVATION WELL NO. 15	OBSERVATION WELL NO. 15 OBSERVATION WELL NO. 16	OBSERVATION WELL NO. 15 OBSERVATION WELL NO. 16	OBSERVATION WELL NO. 15 OBSERVATION WELL NO. 16 OBSERVATION WELL NO. 16 OBSERVATION WELL NO. 16 OBSERVATION WELL NO. 16	OBSERVATION WELL NO: 15 OBSERVATION WELL NO: 15 OBSERVATION WELL NO: 16 OBSERVATION WELL NO: 10 OBSERVATION JUNEAU JUNEA	OBSERVATION WELL NO: 15 OBSERVATION WELL NO: 15 OBSERVATION WELL NO: 16 OBSERVATION WELL NO: 16 31/79 40 40 40 40 40 31/79	S S S S S S S S S S	OBSERVATION WELL NO. 15 30/79 40 A30 40 A30 CE OBSERVATION WELL NO. 15 A30 A10 A10 A10	OBSERVATION WELL NO: 15 OBSERVATION WELL NO: 10 OBSERVATION WELL NO:	S S S S S S S S S S	S

¹Location is shown in Figure \mathcal{L} ; < - Refers to less than; > - Refers to greater than.

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Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG , LOT 95 , CONI . WI.S.R. Date Compiled: 19/04/80

Compiler: N. RENNIE

Count	y: GREY				,	11 P (0) . O	100-0 / 00-1	,					1
Z	Sou To No S	Lo	catio	n Se	Da	F e	B B	Co	Fecal Strept Bacte	a Ps	Sulph Reduc Bacte	Heto Bac	
Identification Number 1	Owner or Source	Township	Lot	SS	Date	Fecal Coliform Bacteria	Background Bacteria	Coliform Bacteria	Fecal Streptococci Bacteria	Pseudomonas aeruginosa	Sulphate Reducing Bacteria	Heterotrophic Bacteria	
owia	OBSERVATION WELL NO. 18			20	79	310	20,000	7700	420	32			
h	h			31/0	/20	410	200	A900	A 20				
													1/0.
].º
TP1	TEST PIT NO. 1			21	1/19	19,000	66,000	91,000	2200	164			$\frac{1}{4}$
													$\frac{1}{2}$
TP2	TEST PIT			2.1	0/ 11/19	256	12,000	4000	196	4			1
11	NO. 2				1/79 1/79		7000	6400	284	8			1

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Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S.R. Date Compiled: 19/04/80

Compiler: N. RENNIE

	· GRET					onip (o). On	EIOCEO, 40,	7, 3, 3, 2		e Compiled	7 80	T T	
Z den	Owner or Source		cati	on O	Date Sam	Fecal Colifo	Backg	Colif Bacte	Fecal Strepto Bacter	Pseu	Sulph Reduc Bacte	Hete	
Identification Number 1	r	Township	Lot	Concession	Date Sampled	Fecal Coliform Bacteria	Background Bacteria	Coliform Bacteria	Fecal Streptococci Bacteria	Pseudomonas aeruginosa	ohate ucing teria	Heterotrophic Bacteria	
3		<u> </u>		Š	DIMIY		α.		<u> </u>	la la			
İ													
TP3	TEST PIT NO. 3				20/ / 79	4	920	6344	44	4			
11	I				26/ 11/79	136	4900	1100	64	<4			
ıı	H				31/01/80	44	G24,000	<10	4		-		
TP4	TEST PIT NO. 4				20/ 11/79	G600	13,000	10,900	552	4			
h	ц				26/ 11/79	A40	2000	1110	A 60	4			

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Number of bacterial colonies per 100 ml

Identification Number 1	y: GREY		ocat i	c Concessio	Town Date Sampled	ship(s): GLI Recal Bacteria	ENELG, LOT9 Background Bacteria	5 ,CONI Coliform Bacteria	Dat Fecal Streptococci Bacteria	Pseudomonas m o aeruginosa	Sulphate Reducing Bacteria	Compiler Heterotrophic Bacteria	RENNIE
TP5	TEST PIT NO 5				20,	<4	800	16	~4	<4			
Li.	Ð				26/	۷.4	440	۷4	4	<4			
					-								

¹Location is shown in Figure λ ; < - Refers to less than; >- Refers to greater than.

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APPENDIX I

SUMMARY OF BACTERIOLOGICAL ANALYSES
OF SURFACE WATER



SUMMARY OF BACTERIOLOGICAL ANALYSES OF SURFACE WATER

Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CONI, W.T.S.R Date Compiled: 19/04/80

Compiler: N. RENNIE

000111	y: GKET				1011	ship (s). OLL	NELO, LOI			C COp11.cc	7 70		-
Nur	Owner or Sourc	Lo	cati	on	Dat San	Fecal Colifor Bacter	Back Bact	Colif Bacte	Fec Stre	Pse	Sulph Reduc Bacte	Hete Bac	
Identification Number ¹	Owner or Source	Township	Lot	Concession	Date Sampled	a B	kground teria	Coliform Bacteria	Fecal Streptococci Bacteria	Pseudomonas aeruginosa	phate ucing teria	Heterotrophic Bacteria	
RSR1	ROCKY SAUGEEN R at C.P.R. TRACKS				22/ 10/ 19	28	2600	280	48				
31	(8.8)				20/	64	4900	C 290	28	44			
II	IV.				31/01/80	4	210	AIO	24				
R5R2	ROCKY SAUGEEN RIVER 50ft above outfall				20/ 11/ 79	A10	2100	0PA	210	4			
	÷					,							
RSR 3	ROCKY SAUSEEN RIVER 300 ft below out fall				20/	A10-	3400	C370	AIO	24			

 $^{^{1}}$ Location is shown in Figure 1; < - Refers to less than; > - Refers to greater than.

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SUMMARY OF BACTERIOLOGICAL ANALYSES OF SURFACE WATER

Number of bacterial colonies per 100 ml

County: GREY

Township(s): GLENELG, LOT 95, CONI W.T.S. R. Date Compiled: 19/04/80

Compiler: N. RENNIE

Count	y: GREY				TOWN	Snip(s). OH	ENLLO, LOT	5,001-2	or i. S. A. Dat	e Compiled	. 754/80	Compilerio	1
Identification Number 1	Owner or Source	Lownship	cat i	SSK	Date ×	Fecal Coliform Bacteria	Background Bacteria	Coliform Bacteria	Fecal Streptococci Bacteria	<u>Pseudomonas</u> aeruginosa	Sulphate Reducing Bacteria	Heterotrophic Bacteria	
RSR4	ROCKY SAUGEEN BYER downstream from lagoon				22/ 10/ 79	64	3900	C550	7ω				
Jt.	1 ₁				20/1/19	160	2500	(390	16	<4			
W	n				31/01/90		264	88	4				
P1	POND NW of Sewage				20/11/79	570	6600	2900	110	<4			-
u	Lagoon'				31/01/90		5100	∠ 100	۷.10				
							×						

¹Location is shown in Figure 1; < - Refers to less than; > - Refers to greater than.

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An accompanying "C" or "A" indicates that the result is a best estimate and is less precise than other determinations.

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APPENDIX J

SUMMARY OF BACTERIOLOGICAL ANALYSES
OF LAGOON EFFLUENT



SUMMARY OF BACTERIOLOGICAL ANALYSES OF SEWAGE LAGOON EFFLUENT

Number of bacterial colonies per 100 ml

Ontario

County: GREY

Township(s): GLENELG, LOT 95, CON I WITS.R. Date Compiled: 19/04/80 Compiler; N. RENNIE

Het	Heterotrophic						
Ва	Bacteria						
Sulph Reduc Bacte	ulphate educing acteria						
Ps.	Pseudomonas aeruginosa	Z4	PRESENT	C528			
Fecal Strept Bacte	Fecal Streptococci Bacteria	A20	6900	6600	A 3000		
Co Ba	Coliform	C65∞	440,000	280,000	48,000		
B B	Background Bacteria	29,∞0	570,000	540,000	100,000		
Со Ва	Fecal Coliform Bacteria	6600	117,000	56,000	A 2000		
D a	Date ×	18/	21/	26/	31/01/80	-	
o n	Concession					-	
ocati	Lot						
Lo	Township					1	
Sou	Owner or Source	SEWAGE LAGOON EFFLUENT	Į1.	н	EI	-	
Z a	Identification Number 1		£0.	ıb	ч		

 $^{^{1}}$ Location is shown in Figure 1 ; < - Refers to less than; > - Refers to greater than.

An accompanying "L" or "G" indicates that the actual density was less than or greater than the number reported.

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MINISTRY OF THE ENVIRONMENT

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